

pH-triggered surface charge-switchable polymer micelles for the co-delivery of paclitaxel/disulfiram and overcoming multidrug resistance in cancer

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Abstract: Multidrug resistance (MDR) remains a major challenge for providing effective chemotherapy for many cancer patients. To address this issue, we report an intelligent polymer-based drug co-delivery system which could enhance and accelerate cellular uptake and reverse MDR. The nanodrug delivery systems were constructed by encapsulating disulfiram (DSF), a P-glycoprotein (P-gp) inhibitor, into the hydrophobic core of poly(ethylene glycol)-*block*-poly(L-lysine) (PEG-*b*-PLL) block copolymer micelles, as well as 2,3-dimethylmaleic anhydride (DMA) and paclitaxel (PTX) were grafted on the side chain of L-lysine simultaneously. The surface charge of the drug-loaded micelles represents as negative in plasma (pH 7.4), which is helpful to prolong the circulation time, and in a weak acid environment of tumor tissue (pH 6.5–6.8) it can be reversed to positive, which is in favor of their entering into the cancer cells. In addition, the carrier could release DSF and PTX successively inside cells. The results of in vitro studies show that, compared to the control group, the DSF and PTX co-loaded micelles with charge reversal exhibits more effective cellular uptake and significantly increased cytotoxicity of PTX to MCF-7/ADR cells which may be due to the inhibitory effect of DSF on the efflux function of P-gp. Accordingly, such a smart pH-sensitive nanosystem, in our opinion, possesses significant potential to achieve combinational drug delivery and overcome drug resistance in cancer therapy.

Keywords: pH response, charge reversal, multidrug resistance, paclitaxel, disulfiram

Introduction

Multidrug resistance (MDR), where cells are instinctive or acquired, immensely limits the effectiveness of anticancer agents and remains one of the primary obstacles of chemotherapy in clinical practice.^{1–4} The mechanisms of MDR have been well studied, and one of the major mechanisms is the overexpression of P-glycoprotein (P-gp) in the tumor cell membrane.^{5,6} P-gp, a kind of adenosine triphosphate-binding cassette transporter, is able to remove many intracellular substrates, such as paclitaxel (PTX), doxorubicin (DOX), docetaxel, vincristine and etoposide into the tumor stroma in an adenosine triphosphate-dependent manner.^{7–9} This consequently leads to reducing the intracellular drug concentration and eventually limits its antitumor activity.^{10,11}

Many strategies aiming at effective reversal of MDR have been investigated through the co-delivery of chemotherapeutic agents and P-gp inhibitor by nanoparticles.^{6,12–15} For example, Wang et al¹⁶ reported that polymeric micelles co-encapsulated with PTX and tacrolimus (a P-gp inhibitor) (P-F/M) could improve intracellular concentration of PTX and showed much stronger cytotoxicity against PTX-resistant human ovarian cancer A2780/PTX cells than PTX-loaded micelles. Unfortunately, the P-gp inhibitor actions in a concentration-dependent manner and the simultaneous release

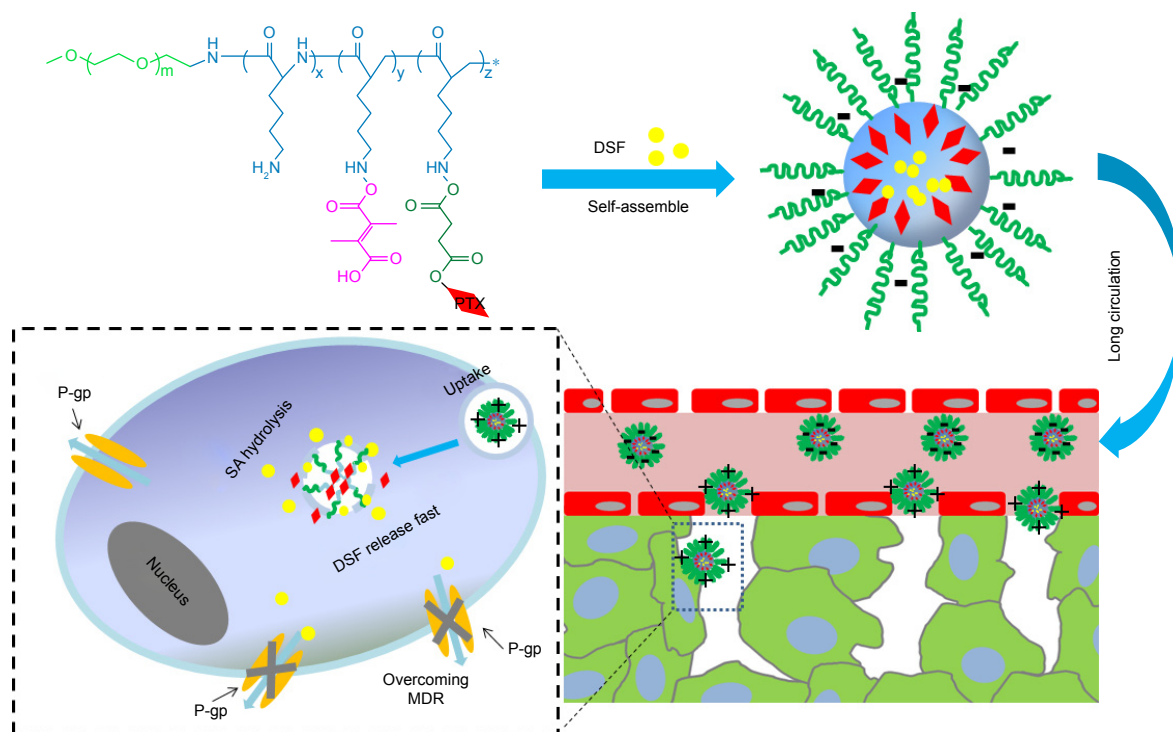
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of P-gp inhibitor and cytotoxic agents may cause a lot of free chemotherapeutic agents being pumped out of the cells before the P-gp functions are restrained. Consequently, it is an attractive strategy to effectively overcome MDR with the P-gp inhibitor possibly releasing relatively quicker than the chemotherapy drug from the same carriers in the cells. Recently, Duan et al¹⁷ reported a pH-sensitive polymeric micelle system loaded with DOX and disulfiram (DSF; a P-gp inhibitor as well as apoptosis inducer) simultaneously, and they found that, because the DSF was released faster than DOX in the cells, the co-delivery system could obviously increase the intracellular accumulation of DOX and to demonstrate strong tumor suppression effects *in vivo*.

On the other hand, a class of stimuli-responsive nanostructures are recently developed and such systems could prolong circulation time in blood and enhance uptake by tumor cells through acidic tumor microenvironment-triggered surface charge switch based on acid-labile bonds.^{18–23} For example, Yuan et al²⁴ prepared a charge-switchable nanoparticle (S-NP/DOX) based on zwitterionic polymers. The S-NP/DOX remained negative surface charge in the blood circulation system and switched to positive surface charge in response to the tumor acidity; the tumor-acidity-activated charge conversion could effectively increase the cellular

uptake of the carriers and the intracellular concentrations of DOX and, in turn, resulted in enhancing tumor inhibitions. In addition, intracellular organelle microenvironments such as low pH in endo-/lysosomal compartments are also available for triggering drug release from drug-carrier conjugates which contain acid-cleavable linkages.^{25,26}

In the aim to overcome the P-gp-mediated MDR, in this study we report novel surface charge-reversible polymer micelles which could cascade release PTX and DSF inside drug-resistant tumor cells (Scheme 1). The carriers were constituted with succinic anhydride-modified PTX (SA-PTX) and 2,3-dimethylmaleic anhydride (DMA)-conjugated poly(ethylene glycol)-*b*-poly(L-lysine). Besides, DSF, a hydrophobic drug, was encapsulated into the micelles. Such a smart drug delivery nanosystem was characterized as follows: 1) the micelles could ideally remain negative surface charge under blood circulation, while quickly changes to positive under the acidic tumor tissues owing to the hydrolysis of the DMA group.²⁷ In addition, this change could increase the endocytosis rate of the micelles; 2) the internalization carriers could successively release two drugs, which means that DSF is quickly released accompanied with the deformation of micelles, and then copolymer-PTX prodrugs are released in a sustained manner in acidic organelles including



Scheme 1 Schematic of the PTX-conjugated copolymers and the charge reversal of DSF and PTX co-loaded micelles for achieving pH-sensitive charge switches and cascade drug release to overcome MDR in cancer.

Abbreviations: DSF, disulfiram; MDR, multidrug resistance; P-gp, P-glycoprotein; PTX, paclitaxel; SA, succinic anhydride.

endo-/lysosomes by pH-dependent forms,²⁸ and 3) PTX, due to P-gp functions inhibited by DSF, the PTX could accumulate in cancer cells at a high concentration and effectively kill tumor cells. The drug co-delivery nanosystem adopted two strategies including enhanced cellular uptake and sequential drug release to combat cancer drug resistance and therefore had great potential in clinical applications.

Materials and methods

Materials

PTX was purchased from Beijing Huafeng United Technology Co., Ltd (Beijing, People's Republic of China). DSF was obtained from Aladdin (Shanghai, People's Republic of China). Succinic anhydride (SA) and DMA were purchased from Adamas Reagent Co., Ltd (Shanghai, People's Republic of China) and used as received. *N,N*-dimethylformamide (DMF) was stored over calcium hydride (CaH₂) and purified by vacuum distillation with CaH₂. Amino-terminated methoxy poly(ethylene glycol) (PEG-NH₂; molecular weight [MW] = 5,000) was purchased from Shanghai Yare Biotech, Co., Ltd (Shanghai, People's Republic of China) and used after dried under vacuum. ϵ -(Benzyloxycarbonyl)-L-lysine *N*-carboxyanhydride (Lys(Z)-NCA) were synthesized by the Fuchs-Farthing method using bis(trichloromethyl)carbonate (triphosgene) according to the literature.²⁹ 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide (MTT) and Hoechst 33342 were purchased from Beyotime Institute of Biotechnology and used without further purification.

Characterization

¹H nuclear magnetic resonance (NMR) spectra were performed on a Bruker Avance 500 spectrometer operating at 500 MHz by using D₂O or dimethyl sulfoxide (DMSO)-*d*₆ as the solvent depending on polymer solubility. The chemical shifts were calibrated against solvent signals. The morphology of nanoparticles (NPs) was observed using transmission electronic microscopy (TEM; Hitachi Ltd., Tokyo, Japan). The particle size and zeta potential were investigated using dynamic light scattering (DLS; Zs90; Malvern Instruments, Malvern, UK).

Synthesis of poly(ethylene glycol)-*block*-poly(L-lysine) (PEG-*b*-PLL) block copolymer

Methoxy PEG-*b*-PLL was synthesized by the ring-opening polymerization of Lys(Z)-NCA using PEG-NH₂ as the macroinitiator and by the subsequent deprotection of benzyloxycarbonyl groups.³⁰ Briefly, PEG-NH₂ (1.50 g, 0.30 mM) was dissolved in 30 mL of dry DMF as the macroinitiator. Then,

Lys(Z)-NCA (2.94 g, 9 mM) was dissolved in 40 mL of dry DMF and added to the PEG-NH₂ solution under a dry argon atmosphere. The reaction mixture was stirred for 48 h at 40°C under a dry argon atmosphere, and then the solvent was evaporated under reduced pressure. The resulting product was precipitated into excess ice-cold diethyl ether to obtain methoxy poly(ethylene glycol)-*block*-poly- ϵ -(benzyloxycarbonyl)-L-lysine (PEG-*b*-PLLZ; yield 87.2%).

PEG-*b*-PLL was obtained through the deprotection of Z groups on PEG-*b*-PLLZ. Typically, PEG-*b*-PLLZ (2.0 g) was dissolved in CF₃COOH (20 mL). Then, 3 mL of HBr/acetic acid (33%) was added at 0°C. The reaction mixture was stirred for 30 min at 25°C, and then the reaction mixture was precipitated in excessive cold diethyl ether. The precipitation was dissolved in DMSO and further purified by dialyzed (molecular weight cutoff [MWCO] = 3,500 Da) against distilled water. The product was obtained by lyophilization (yield 63.5%).

Synthesis of PEG-*b*-(PLL-*g*-PTX)

PTX was conjugated to PEG-*b*-PLL through an SA linker, as reported previously,^{28,31} and had some differences. Briefly, PTX (1.0 g, 1.17 mM) was dissolved in 80 mL of dry DMF. SA (0.176 g, 1.76 mM), 4-dimethylaminopyridine (DMAP; 0.143 mg, 1.17 mM) and triethylamine (TEA; 0.178 g, 1.76 mM) were added, and the reaction mixture was stirred at 40°C for 24 h. The solvent was evaporated under reduced pressure, and the product was washed with 0.01 M HCl three times and distilled water three times. After filtration and drying under vacuum for 24 h, a white solid of SA-PTX was obtained (yield 85%).

PTX-SA was conjugated to the amino groups of PEG-*b*-PLL to obtain the PEG-*b*-(PLL-*g*-PTX). In brief, in a 100 mL flame-dry flask, SA-PTX (0.2 g, 0.21 mM), 3-(ethyliminomethylideneamino)-*N,N*-dimethylpropan-1-amine-HCl (0.113 g, 0.63 mM) and *N*-Hydroxysuccinimide (0.073 g, 0.63 mM) were dissolved in 20 mL of dry DMF, and the reaction mixture was maintained under stirring for 2 h at 35°C to activate carboxyl group. Subsequently, PEG-*b*-PLL (2.0 g, 0.21 mM) and TEA (0.045 g, 0.45 mM) were dissolved in dry DMF and then added to the abovementioned solution. The reaction mixture was maintained at 35°C for 36 h. The whole solution was transferred to a dialysis bag (MWCO = 3,500 Da) and dialyzed against DMF, and then deionized water. Finally, the product was obtained by lyophilization. PTX content in the PEG-*b*-(PLL-*g*-PTX) was determined by ¹H NMR and high-performance liquid chromatography (HPLC) measurements.³² For the determination of

PTX by HPLC method, the mobile phase consisted of water and acetonitrile (35:65); an Ultimate AQ-C₁₈ column was used with a flow rate of 1 mL/min; the detection wavelength was 227 nm and the column temperature was maintained at 30°C. The PTX content was calculated according to the following formula:

$$\text{PTX content (wt \%)} = \frac{\text{Weight of PTX}}{\text{Weight of PEG-}b\text{-(PLL-g-PTX)}} \times 100\%$$

Synthesis of PEG-*b*-(PLL-*g*-PTX/DMA)

The charge-reversal polymers were prepared by grafting the amino residues of PLL blocks into β -carboxylicamide using DMA according to the previous literature.¹⁸ Briefly, PEG-*b*-(PLL-*g*-PTX) (0.20 g) and DMA (0.07 g, three equivalents of amino groups) were dissolved in 30 mL of deionized water, and the pH of the solution was kept at 8–9 by adding 1 M NaOH. After the reaction mixture was stirred for 12 h, the solution was dialyzed (MWCO = 3,500 Da) against slightly basic water (pH 9–10) and a white solid was obtained by lyophilization. As a control, the no charge-reversal polymers, PEG-*b*-(PLL-*g*-PTX/SA), were synthesized through the reaction between PEG-*b*-(PLL-*g*-PTX) and SA by using the same method.

Preparation of the DSF and PTX co-loaded micelles

The DSF and PTX co-loaded micelles with charge reversal (abbreviated as DA-NPs) were fabricated by the nanoprecipitation method. Briefly, PEG-*b*-(PLL-*g*-PTX/DMA) (10 mg) and DSF (1 mg) were co-dissolved in 3 mL DMF stirred for 2 h, and then the mixture solution was added dropwise to 10 mL of pure water under vigorous stirring. The micelle solution was then dialyzed (MWCO = 3,500 Da) against deionized water and filtered through a Millipore filter (pore size 0.45 μm) to remove unencapsulated DSF. The drug loading content (DLC) and drug loading efficiency (DLE) of DSF were measured by HPLC.

Moreover, the no charge-switchable micelles co-loaded with DSF and PTX were prepared as a control using the same protocol as DA-NPs, except that PEG-*b*-(PLL-*g*-PTX/DMA) polymer was replaced with PEG-*b*-(PLL-*g*-PTX/SA), and these micelles were abbreviated as SA-NPs.

The coumarin-6-labeled micelles were also prepared, and the method was similar to those of DA-NPs, except that DSF was replaced with coumarin-6.

Determination of the charge-reversal ability of DA-NPs

To study the surface charge characteristics of DA-NPs, the DA-NPs (0.1 mg·mL⁻¹) were dispersed in PBS (pH 6.5 or 7.4, 0.01 M) and incubated at 37°C.³³ Samples were taken at designated time intervals, and zeta potentials were measured.

Characterization of the stability of DA-NPs and SA-NPs

The DA-NPs and SA-NPs were gently mixed with RPMI-1640 medium supplemented with 10% fetal bovine serum (FBS) at pH 7.4 or 6.5. The mean diameters of DA-NPs and SA-NPs at different periods of incubation time were measured.²⁴

Protein adsorption of the DA-NPs and SA-NPs

Bovine serum albumin (BSA) was used as a model protein to investigate the protein adsorption on the different micelles. Typically, DA-NPs and SA-NPs with the final concentration of 0.57 mg/mL were incubated with the solution of BSA (0.5 mg/mL) dissolved in PBS with different pH values (pH 6.5 or 7.4), respectively. After incubation at 37°C for 2 h, each sample was withdrawn and centrifuged at 10,000 rpm for 15 min to precipitate the protein adsorbed on DA-NPs or SA-NPs. In addition, the concentration of free BSA in supernatant was determined by the commercial BCA Protein Assay kits.¹⁸

In vitro pH-responsive release profiles of PTX and DSF

The release behaviors of PTX and DSF from DA-NPs were measured by ultrafiltration centrifugation method in PBS (0.2 M, pH 7.4) or HAc-NaAc buffer (0.04 M, pH 5.0) containing 0.5% (w/v) Tween-80 at 37°C.³⁴ Briefly, DA-NPs were dispersed into 4 mL media, and every sample was shaken at 100 rpm and 37°C. At predetermined intervals, the solution was transferred to Centriprep centrifugal filter units (MWCO = 3,000 Da) and then centrifuged at 4,000 rpm for 10 min. The ultrafiltrate was collected, and the amount of the released PTX and DSF was measured by HPLC.

Cell culture

MCF-7 was obtained from the Institute of Biochemistry and Cell Biology, Chinese Academy of Sciences (Shanghai, People's Republic of China) and cultured in RPMI 1640 medium containing 10% FBS, 100 U/mL penicillin G sodium and 100 $\mu\text{g/mL}$ streptomycin sulfate at 37°C and 5% CO₂ in

humidified incubator. MCF-7/ADR was obtained from Central South University (Changsha, People's Republic of China) and maintained in media containing DOX at a concentration of 1 µg/mL as reported previously.^{12,35} All cell experiments were performed according to protocols evaluated and approved by the ethics committee of Nanjing Medical University.

PTX accumulation assay

MCF-7 and MCF-7/ADR cells were seeded in 24-well plates (5×10^4 cells/well) and then treated with PTX, PTX + DSF, DA-NPs and SA-NPs for 1, 2 and 4 h at 37°C. The concentrations of PTX and DSF were 5 and 1 µg/mL, respectively.¹⁷ After incubation, the cells were washed twice with cold PBS. Subsequently, 200 µL of cell lysis buffer (1% of TritonX-100) was added and incubated for 30 min. After that, the cell lysate (100 µL) was mixed with acetonitrile (200 µL) by ultrasonication for drug extraction followed by centrifugation at 6,000 rpm for 10 min, the supernatant was collected and the concentration of PTX was measured by HPLC as described earlier. All the determination of PTX content was normalized to protein concentrations of cell lysate.

Fluorescence microscopy (FLM) observation

MCF-7/ADR cells were seeded into six-well plates at a density of 1×10^5 cells/well and incubated for 24 h. Subsequently, cells were incubated with coumarin-6-loaded DA-NPs or SA-NPs at pH 7.4 or 6.5 for 2 h at 37°C. After that, the cells were washed with precooled PBS and fixed with 4% formaldehyde for 15 min, and then the cell nuclei were stained by 10 mM Hoechst 33342 for 10 min. Afterward, the cells were analyzed by using FLM (Leica Microsystems, Wetzlar, Germany).

Rhodamine 123 (Rh123) efflux assay

Rh123, as a P-gp substrate fluorescent dye, is an index of assaying the transport activity of P-gp.^{12,35} The transport activity of P-gp was first studied in MCF-7/ADR and MCF-7 cells via fluorescence spectrophotometer. Briefly, MCF-7/ADR or MCF-7 cells were seeded in 24-well plates at 5×10^4 cells/well. After overnight attachment, the cells were incubated with Rh123 at different concentrations (5, 10 or 20 µg/mL) at 37°C for 30 min. Subsequently, the free Rh123 was removed with PBS for washing three times, and then the cells were maintained in dye-free medium for another 2 h. At the end of the incubation period, the cells were lysed and centrifuged. The concentration of Rh123 in the supernatant was measured

by fluorescence spectrophotometer (F-4600; Hitachi Ltd.). All the determination of Rh123 content was normalized to protein concentrations of cell lysate.

To study DSF inhibition of P-gp efflux function, MCF-7/ADR cells were seeded in 24-well plates and treated with verapamil, taxol, taxol + DSF, DA-NPs and SA-NPs overnight, followed by incubation with 10 µg/mL of Rh123 for another 30 min at 37°C. The other steps were the same as mentioned earlier.

In vitro cytotoxicity

The MTT assay was used to examine the vitro cytotoxicity of different drug-loaded NPs against MCF-7 and MCF-7/ADR cells. Briefly, the cells were seeded in 96-well plates at a seeding density of 1×10^4 cells/well. After 24 h of incubation, the medium was replaced with a 0.2 mL of fresh medium containing PTX, DSF, PTX + DSF, DA-NPs and SA-NPs at various drug concentrations (mass ratio of PTX to DSF was fixed at 5:1) for 48 h at pH 7.4. Afterward, 20 µL of MTT (5 mg/mL in PBS) was added to each well, and the plates further were incubated for 4 h in the dark. Then, the solution was removed, and 200 µL of DMSO was added to each well. Finally, the absorbance was detected at 490 nm by using a microplate reader (Thermo Multiskan MK3; Thermo Fisher Scientific, Waltham, MA, USA).

Statistical analysis

All experimental values were expressed as mean \pm SD, $n=3$ or 5. All data were analyzed using the one-way ANOVA analysis. A p -value of <0.05 was considered to be statistically significant.

Results and discussion

Preparation and characterization of PEG-b-(PLL-g-PTX/DMA)

The detailed synthesis protocol of PEG-b-(PLL-g-PTX/DMA) is shown in Figure S1. The preparation of PEG-b-(PLL-g-PTX/DMA) mainly contained three steps: first, PEG-*b*-PLL copolymers and PTX-SA were synthesized; then, PTX-SA was conjugated to the side groups of PEG-*b*-PLL through esterification reaction; and finally, DMA was grafted on the residue amino groups of PEG-*b*-PLL through amidation reaction.

In this study, the PEG-*b*-PLL was synthesized by the ring-opening polymerization. The ¹H NMR spectra of PEG-*b*-PLLZ and PEG-*b*-PLL demonstrated that PEG-*b*-PLL was synthesized successfully. All the peaks in the spectra were consistent with the literature (Figure 1).^{30,36} The degree of

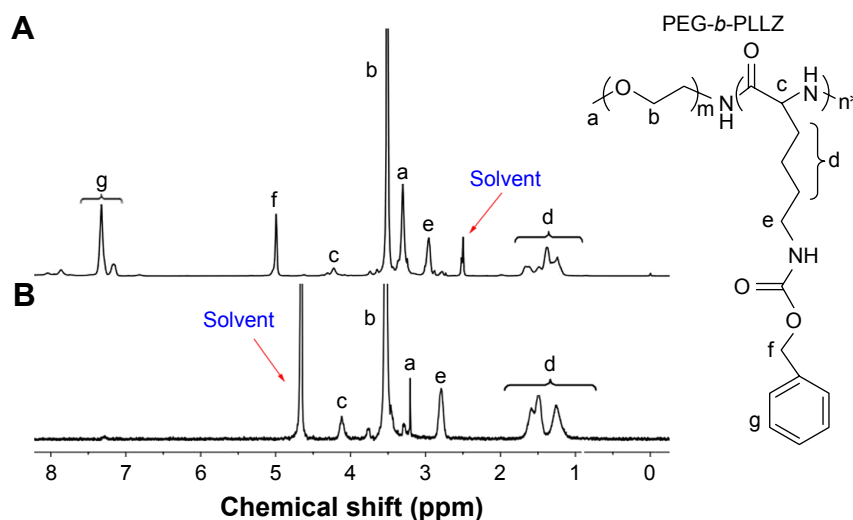


Figure 1 The ^1H NMR spectra of PEG-*b*-PLLZ in $\text{DMSO-}d_6$ (A) and PEG-*b*-PLL in D_2O (B).

Note: *Indicates a hydrogen atom.

Abbreviations: DMSO, dimethyl sulfoxide; NMR, nuclear magnetic resonance; PEG-*b*-PLL, poly(ethylene glycol)-*block*-poly(L-lysine); PEG-*b*-PLLZ, poly(ethylene glycol)-*block*-poly- ϵ -(benzyloxycarbonyl)-L-lysine.

polymerization (DP) of PLLZ was calculated by comparing the signal intensities of PLLZ methylene protons ($-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{NH}$) with methylene protons of PEG ($-\text{CH}_2-\text{CH}_2-\text{O}-$), and the value was 15. After deprotection, the typical peaks of carbobenzyoxy groups at δ 5.00 and 7.17–7.35 ppm disappeared in PEG-*b*-PLL (Figure 1B), which indicated that the deprotection was complete. The DP of PLL in PEG-*b*-PLLZ was calculated to be 15 by using the same method as for PEG-*b*-PLLZ.

PTX, a cytotoxic drug, was then conjugated to PEG-*b*-PLL through SA as a linker. ^1H NMR spectrum confirmed that PTX-SA was successfully prepared and the signals at δ 2.75 and 2.92 ppm were the methylene peaks of SA.³⁷ Moreover, the 2'-CH proton peak on the ^1H NMR spectra was shifted from 4.67 ppm (free PTX) to 5.90 ppm (conjugated PTX; Figure 2).^{38,39}

PTX-SA was conjugated to the side chains of the amino groups in PEG-*b*-PLL through a typical amidation reaction to obtain an ester bond containing polymer PEG-*b*-(PLL-*g*-PTX). As shown in Figure 3, PTX-SA was successfully conjugated to PEG-*b*-PLL, which was certified by the typical signals of the phenyl proton (δ 7.28–8.0 ppm) of PTX that appeared in the ^1H NMR spectra of PEG-*b*-(PLL-*g*-PTX). Subsequently, the mass fraction of PTX to PEG-*b*-(PLL-*g*-PTX) was 12.8% by calculating the peak area of the ^1H NMR spectrum (Figure 3)³² and was 11.3% by HPLC detection through complete acid hydrolysis of SA. Finally, DMA was modified on the PEG-*b*-(PLL-*g*-PTX) to obtain PEG-*b*-(PLL-*g*-PTX/DMA), and the characteristic peaks of DMA were shown in the ^1H NMR spectrum (Figure 4).

The characteristics of DA-NPs

The well-defined amphiphilic PTX-conjugated copolymers could form micelles (DA-NPs or SA-NPs) by a self-assembly process in aqueous solution. The morphology and size of DA-NPs and SA-NPs were investigated by TEM and DLS, respectively. As shown in Figure 5, the size of DA-NPs and SA-NPs was 129.8 ± 8.5 and 118.6 ± 2.1 nm ($n=3$) with a narrow distribution, respectively. The TEM images showed that both micelles had good monodispersity and were nearly in spherical shape with diameter about 100 nm, which were consistent with the results determined by DLS.

Surface charge conversion behavior of DA-NPs

Recently, nanocarriers with the characteristics of pH-dependent charge conversion which could enhance tumor cell adhesion and trigger drug release were developed for drug delivery. DA-NPs or SA-NPs were dispersed in the PBS solution with different pH values (6.5 and 7.4), respectively, and the zeta potential was monitored. As shown in Figure 6A, the zeta potential of DA-NPs changed rapidly with time at pH 6.5, and the corresponding values were increased from -13.3 to 6.5 mV after 15 min and further increased to 10.5 mV after 1 h, which could be attributed to the gradual hydrolysis of DMA groups under acidic conditions. However, the zeta potential of DA-NPs was increased slowly at pH 7.4 and still maintained a negative charge (-1.5 ± 0.5 mV) within 1 h, which suggested that the DMA groups of DA-NPs remained relatively stable at physiological pH. These phenomena were also in good consistent with

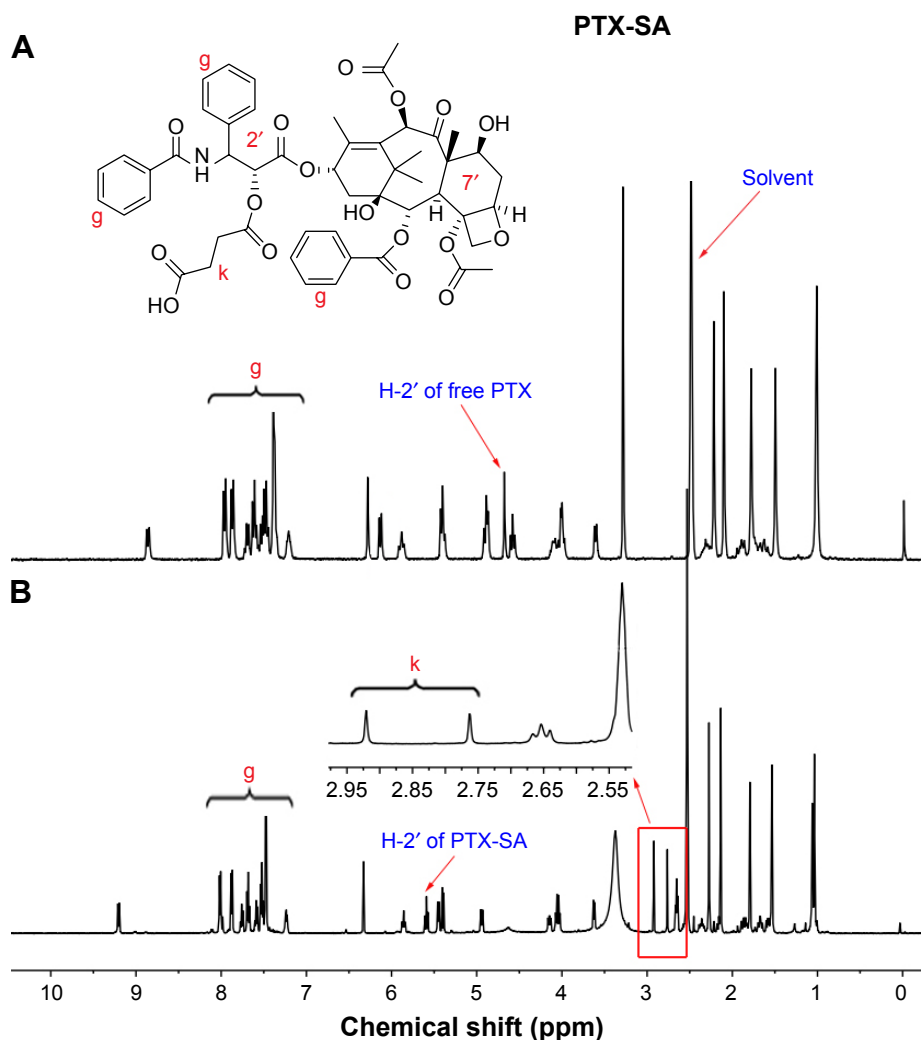


Figure 2 The ^1H NMR spectra of PTX (A) and PTX-SA (B) in $\text{DMSO}-d_6$.
Abbreviations: DMSO, dimethyl sulfoxide; NMR, nuclear magnetic resonance; PTX, paclitaxel; SA, succinic anhydride.

previous report.^{19,40,41} In sharp contrast, the zeta potential of SA-NPs, which were set as a negative control, did not exhibit significant change, whether at pH 6.5 (12.3 ± 1.7 mV) or at pH 7.4 (12.4 ± 1.3 mV), in 2 h, which indicated that SA groups possessed higher stability compared to the DMA group (Figure 6B). Overall, these results indicated that DA-NPs were capable of changing surface charge on receiving a pH signal, which would promote its adsorption on tumor cells and have a higher probability of entering the cells.

In vitro stability study and protein adsorption

To study the stability of DA-NPs and SA-NPs, cell cultured with 10% FBS was used to simulate blood microenvironment (pH 7.4) and tumor tissues at slightly acidic conditions (pH 6.5). As shown in Figure 7A, after 24 h of incubation at pH 7.4, the particle size of DA-NPs was maintained at

138 ± 8.5 nm, indicating that they are relatively stable at blood circulation. However, the size of DA-NPs drastically changed from 138 to 1,000 nm after 24 h of incubation at pH 6.5, which maybe because the negative surface charge changes to positive charge resulting in abundant protein were adsorbed on the surface of NPs. As a control, there was almost no change in the size of SA-NPs, whether at pH 7.4 or at pH 6.5. These phenomena were also in good consistent with previous reports.²⁴

Reduced nonspecific protein adsorption of the drug-loaded NPs is an important indicator for the prolonged blood circulation.^{19,42} We hypothesized that the surface charge reversal of DA-NPs may be advantageous to the reduction in nonspecific protein adsorption to NPs during circulation. To test this hypothesis, BSA was set as a model of plasma protein to simulate the interaction between micelles and protein. As shown in Figure 7B, when the pH values decreased to 6.5,

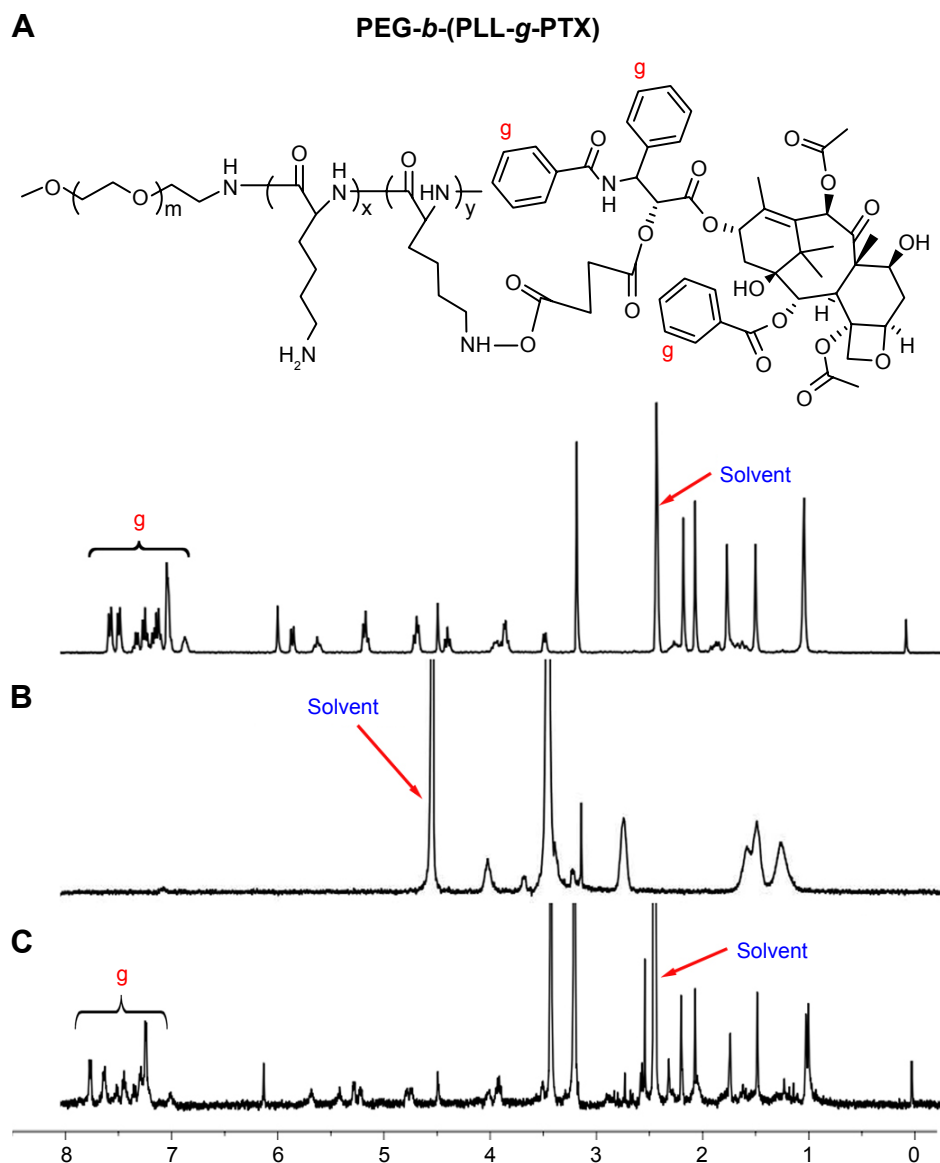


Figure 3 The ^1H NMR spectra of PTX (A) in $\text{DMSO-}d_6$, PEG-*b*-PLL (B) in D_2O and PEG-*b*-(PLL-*g*-PTX) (C) in $\text{DMSO-}d_6$.

Abbreviations: DMSO, dimethyl sulfoxide; NMR, nuclear magnetic resonance; PEG-*b*-PLL, poly(ethylene glycol)-*block*-poly(L-lysine); PTX, paclitaxel.

DA-NPs with positive charge strongly interacted with the protein and adsorbed more than 60% protein within 2 h. In contrast, SA-NPs showed negligible protein adsorption under similar conditions. However, under the physiological pH conditions (pH 7.4), both DA-NPs and SA-NPs exhibited minimal protein adsorption (less than 20%) after 2 h incubation. These results were in good agreement with the pH-dependent charge-switchable and size-changeable behaviors of DA-NPs at pH 6.5.

Drug release

As previously described, DA-NPs were expected to have the ability of temporal release of DSF and PTX, which was aimed that DSF could inhibit the function of P-gp first and

then the release of PTX was activated by a pH change in the intracellular endosomal compartments in a sustained manner. To confirm this property, the release kinetic behavior of each drug from DA-NPs was measured. PBS (pH 7.4) and HAC-NaAc buffer (pH 5.0) were used to simulate the environment of plasma and endosome, respectively. The DLC of PTX and DSF in DA-NPs was $10.73\% \pm 0.8\%$ and $1.97\% \pm 0.5\%$ and in SA-NPs were $11.42\% \pm 0.7\%$ and $2.15\% \pm 0.5\%$ ($n=3$), respectively, as measured by HPLC method (Table 1). As shown in Figures 8A and S2, 72.5% and 44.3% of DSF were released from DA-NPs within 12 h after incubation at pH 5.0 and 7.4, respectively, which was convenient to inhibit P-gp quickly. In contrast, trace amounts of PTX were released from DA-NPs at pH 7.4 within 12 h

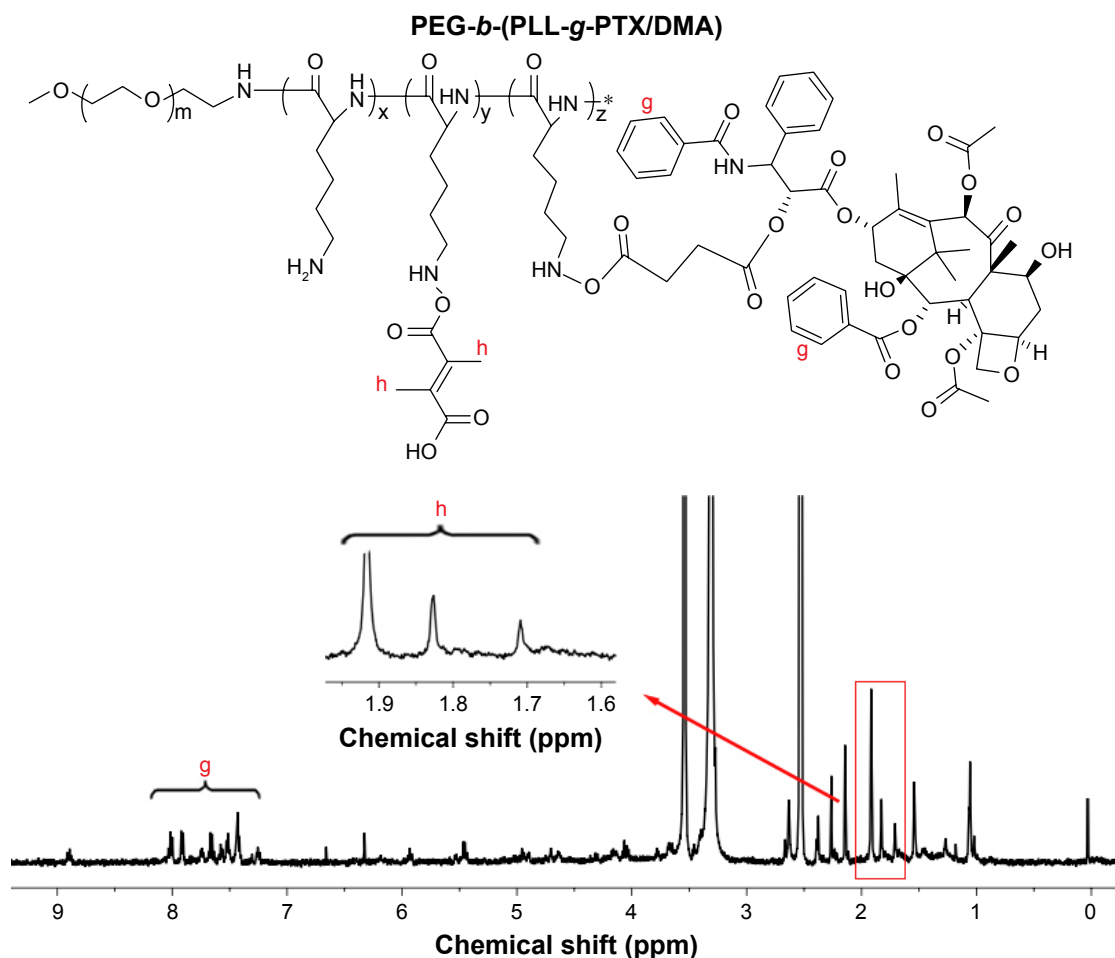


Figure 4 The ^1H NMR spectra of PEG-*b*-(PLL-*g*-PTX/DMA) in $\text{DMSO-}d_6$.

Note: *Indicates a hydrogen atom.

Abbreviations: DMA, 2,3-dimethylmaleic anhydride; DMSO, dimethyl sulfoxide; NMR, nuclear magnetic resonance; PTX, paclitaxel.

(Figure 8B), which may be that the ester bond between PTX and SA was still stable under such conditions. However, the release amount of PTX from DA-NPs was modestly increased at pH 5.0 in a sustained manner up to 72 h and

released 53.6%. These results suggested that DA-NPs could minimize the amount of premature drug release during the blood circulation, and release cargoes in the acidic intracellular compartments.

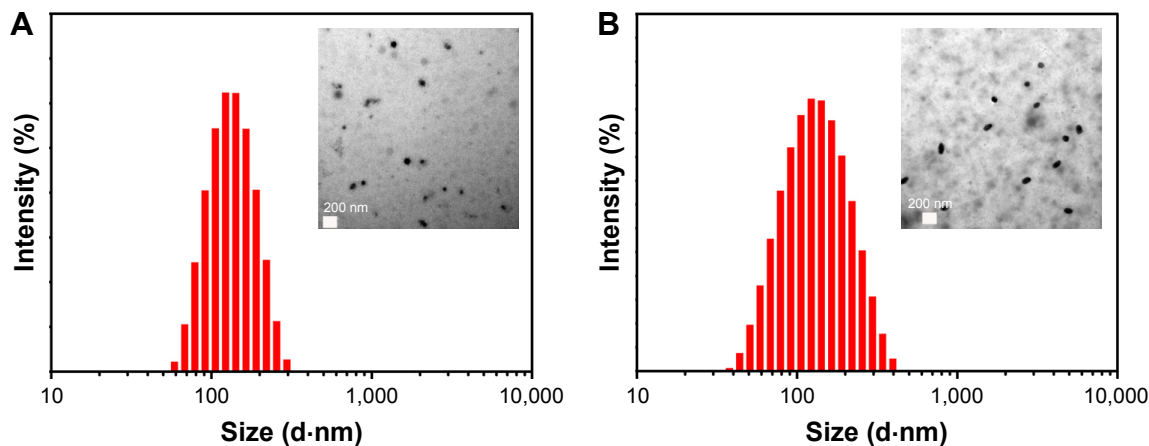


Figure 5 The TEM images and hydrodynamic size distribution of DA-NPs (A) and SA-NPs (B).

Note: Scale bar: 200 nm.

Abbreviations: DA-NPs, DSF and PTX co-loaded micelles with charge reversal; NPs, nanoparticles; SA, succinic anhydride; TEM, transmission electronic microscopy.

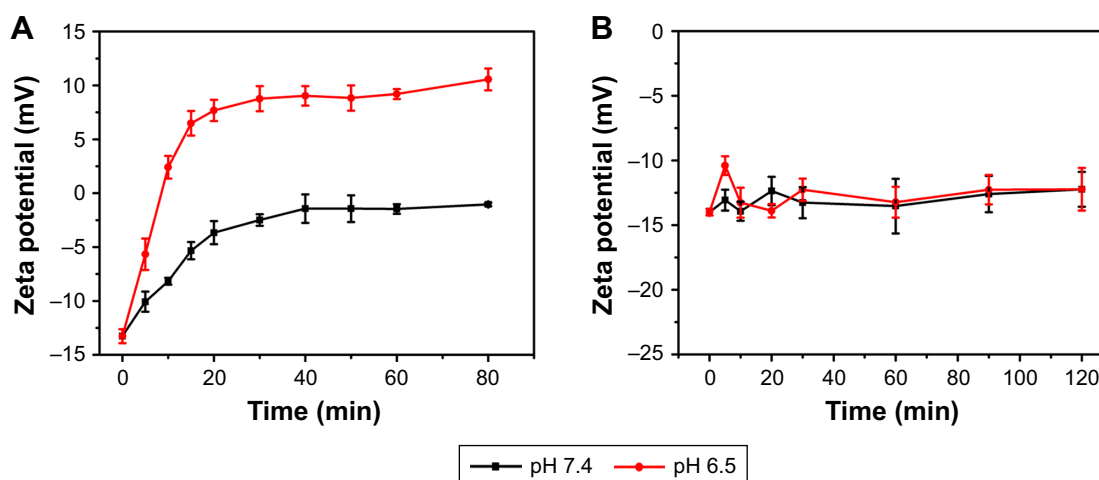


Figure 6 Zeta potentials of DA-NPs (A) and SA-NPs (B) at pH 7.4 and 6.5 with different incubation times.

Abbreviations: DA-NPs, DSF and PTX co-loaded micelles with charge reversal; NPs, nanoparticles; SA, succinic anhydride.

Cellular uptake of DA-NPs

To test whether the charge conversion feature of DA-NPs could enhance their cellular internalization, the cellular uptake of DA-NPs and SA-NPs in MCF-7/ADR cells was observed quantitatively by fluorescence microscope and using coumarin-6 as fluorescent marker. As shown in Figure 9, after 2 h of incubation at pH 7.4, little green fluorescence of DA-NPs was observed in the cytoplasm of MCF-7/ADR cells, but abundant green fluorescence could be measured when incubated at pH 6.5. However, whether at pH 7.4 or 6.5, little green fluorescence of SA-NPs was shown in the cytoplasm of MCF-7/ADR cells. These results confirmed that the surface charge-reversal capability of DA-NPs could effectively enhance their internalization into tumor cells.

The intracellular accumulation of PTX was also evaluated in both PTX-sensitive MCF-7 cells and PTX-resistant MCF-7/ADR cells at pH 7.4 and 6.5 by HPLC method. From the results, the cellular uptake of DA-NPs in MCF-7/ADR cells at pH 6.5 was 1.2-, 1.5- and 1.5-fold higher than that of SA-NPs after incubation for 1, 2 and 4 h, respectively (Figure 10A). In addition, the cellular uptake of DA-NPs in MCF-7 cells at pH 6.5 was 1.6-, 1.8- and 1.5-fold higher than that of SA-NPs after incubation for 1, 2 and 4 h, respectively (Figure 10B). However, under the physiological pH conditions (pH 7.4), the intracellular accumulation of PTX of DA-NPs and SA-NPs in both MCF-7/ADR cells and MCF-7 cells exhibited slight accumulation and no obvious difference. All these results further demonstrated that the

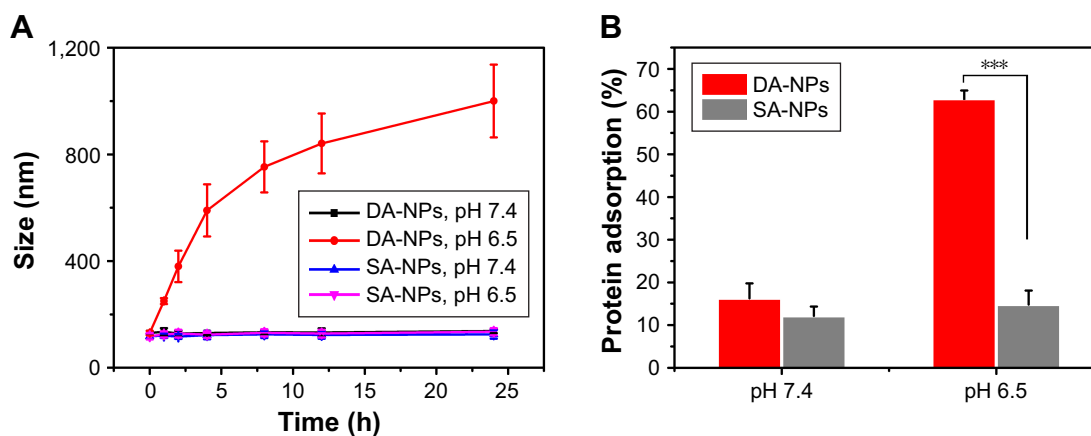


Figure 7 (A) Changes in particle size of DA-NPs and SA-NPs following incubation with cell culture medium with 10% FBS at pH 7.4 or pH 6.8. (B) BSA adsorption on the DA-NPs and SA-NPs after 2 h incubation at 37°C, pH 7.4 or 6.8. Error bars indicate SD (n=4), ***p<0.001.

Abbreviations: BSA, bovine serum albumin; DA-NPs, DSF and PTX co-loaded micelles with charge reversal; FBS, fetal bovine serum; NPs, nanoparticles; SA, succinic anhydride.

Table 1 The drug loaded in DA-NPs and SA-NPs (n=3)

NPs	DLC of PTX (%)	DLC of DSF (%)	DLE of DSF (%)	Ratio (PTX/DSF)
DA-NPs	10.73±0.5	1.97±0.8	85.7±2	5.4
SA-NPs	11.42±0.7	2.15±0.5	89.3±1	5.3

Abbreviations: DA-NPs, DSF and PTX co-loaded micelles with charge reversal; DLC, drug loading content; DLE, drug loading efficiency; DSF, disulfiram; NPs, nanoparticles; PTX, paclitaxel; SA, succinic anhydride.

acid-triggered surface charge reversal of the micelles could effectively increase cellular uptake.

Inhibition of P-gp analysis

DSF is a member of the dithiocarbamate family which was traditionally used in the treatment of alcoholism by irreversibly inhibiting the acetaldehyde dehydrogenase.^{43,44} Recently, it is reported that DSF could permanently inhibit P-gp activity by covalently modifying the cysteine residues in the nucleotide-binding domains of P-gp and/or block its maturation.^{45,46} The influence of DSF on the intracellular accumulation of PTX and the P-gp-mediated drug efflux was investigated in both MCF-7 cells and MCF-7/ADR cells.

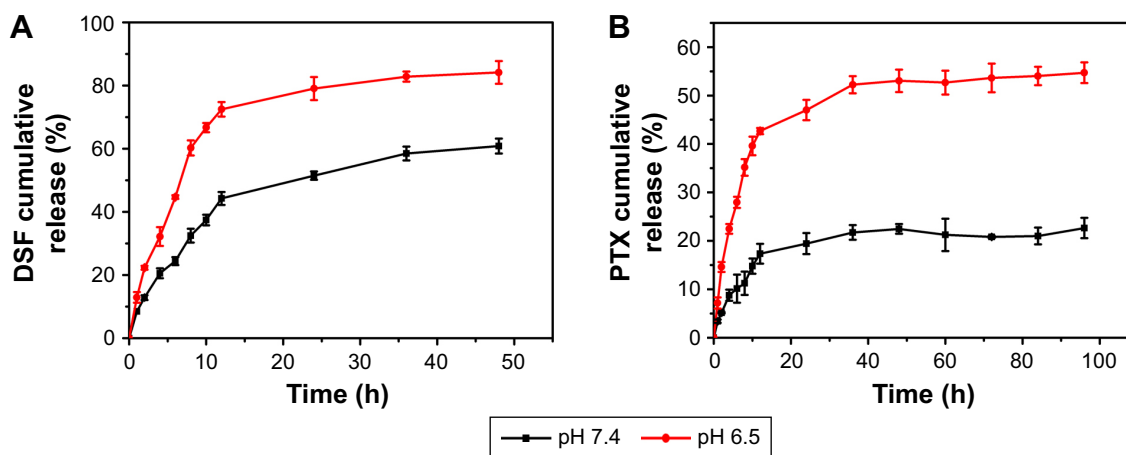
As shown in Figure 11A, free PTX the accumulation of PTX in MCF-7/ADR were very low (11.07 ± 1.71 ng/ μ g protein), but the amount significantly increased to 18.83 ± 1.72 ng/ μ g protein when co-incubation with DSF after 4 h. Furthermore, DA-NPs and SA-NPs were found to increase intracellular PTX concentrations compared with free drug and this may be caused by internalized of these micelles into the cells via endocytic pathways. Moreover, the free PTX accumulation in MCF-7/ADR cells was lower than that in MCF-7 cells, which may be one of the reasons that MCF-7/ADR was resistant

to PTX. However, there was no significant difference of the concentration of PTX among each group in MCF-7 cells (Figure 11B). These results suggested that DSF may increase the intracellular accumulation of PTX.

Overexpressed P-gp on the membrane of MCF-7/ADR cells was one of the major causes of the MDR phenotype.^{12,47} To overcome this drug efflux mechanism, it is necessary to either inhibit its expression, inhibit its function, or bypass transporter recognition.⁴⁸ The potential effect of DSF on the P-gp activity in MCF-7/ADR cells was investigated using Rh123 as a P-gp substrate. As shown in Figure 11C, when the concentration of Rh123 was 5, 10 and 20 μ g/mL, the mean fluorescence intensity of the MCF-7 cells was 108.5-, 147.8- and 155.9-folds higher than that of MCF-7/ADR cells, respectively, which indicated that MCF-7/ADR cells retained strong MDR properties, while the MCF-7 cells were relatively sensitive.³⁵ Furthermore, the drug efflux inhibition was evaluated on different NPs using MCF-7/ADR cells. As shown in Figure 11D, the mean fluorescence intensity of Rh123 in the cells treated with taxol did not alter significantly. However, the intracellular accumulation of Rh123 in MCF-7/ADR cells treated with taxol + DSF, DA-NPs, SA-NPs or verapamil (positive control group) was significantly higher than that in untreated cells. This phenomenon suggested the P-gp transport function was greatly inhibited by DSF.

In vitro cytotoxicity

The vitro cytotoxicity of various PTX-loaded NPs was further investigated by MTT method.⁶ It reported that MCF-7/ADR cells, a typical resistant cancer cell line of DOX, exhibit a

**Figure 8** The time-dependent cumulative release of PTX (A) and DSF (B) from DA-NPs at different pH values.

Abbreviations: DA-NPs, DSF and PTX co-loaded micelles with charge reversal; DSF, disulfiram; NPs, nanoparticles; PTX, paclitaxel.

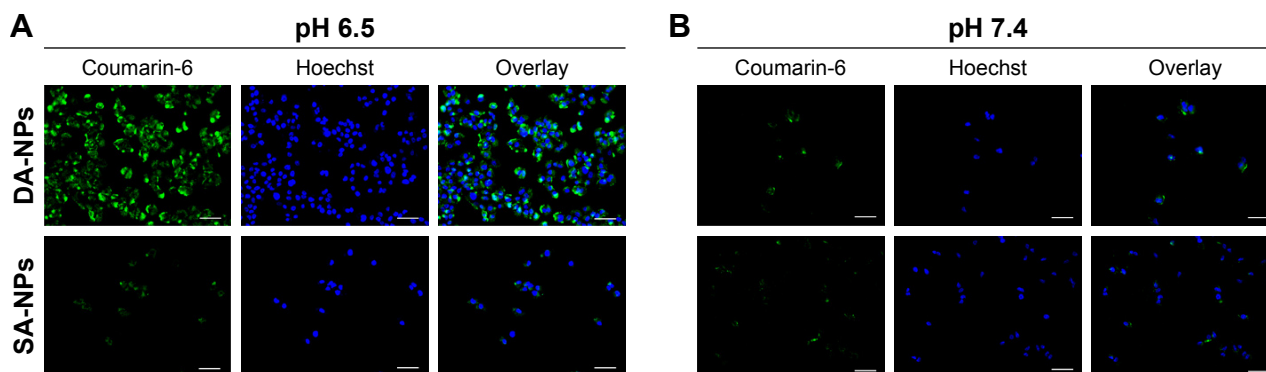


Figure 9 The FLM images of MCF-7/ADR cells after incubation with coumarin-6-loaded DA-NPs (A) and SA-NPs (B) for 2 h at different pH values.

Note: Scale bar =50 μm .

Abbreviations: DA-NPs, DSF and PTX co-loaded micelles with charge reversal; FLM, fluorescence microscopy; NPs, nanoparticles; SA, succinic anhydride.

cross drug resistant to PTX.^{49,50} This cell line has been used to study the reversal of MDR for PTX in this study. Figure 12A revealed that the cell viability of taxol in MCF-7/ADR cells were above that in MCF-7 cells at each dose, suggesting a drug-resistant property against PTX.¹² Compared with taxol, the combination of PTX and DSF could effectively improve the killing effect of drug-resistant cells at each dose, which may be due to the P-gp inhibitor effect of DSF. This phenomenon was also observed in DA-NPs and SA-NPs. However, in MCF-7 cells, the cell viability of combination of PTX and DSF was obviously lower than that of other drug forms, which could be due to the effect of DSF on cell apoptosis (Figure 12B). Moreover, the results showed that the cytotoxic effect of the DA-NPs was pH dependent on both normal and drug-resistant cells (Figure 12C and D). At pH 6.5, the cytotoxicity of DA-NPs was significantly higher than

that of SA-NPs, this may be due to the acid-triggered surface charge reversal.^{24,29} In addition, all the results indicated that DA-NPs with charge reversal could co-deliver PTX and DSF, and showed strong cytotoxicity to drug-resistant MCF-7/ADR cells.

Conclusion

In this study, we designed and constructed a surface charge-switchable polymer-based multidrug delivery system, which provided promising approaches to reverse MDR and achieved more effective treatment. The nanostructure exhibited a variety of attractive functions, such as tumorous acid-triggered surface charge reversal and sequential release of two drugs in tumor cells, which could enhance intracellular drug delivery and produce good synergistic effects of P-gp inhibitors and cytotoxic agents, respectively. As a result, this system

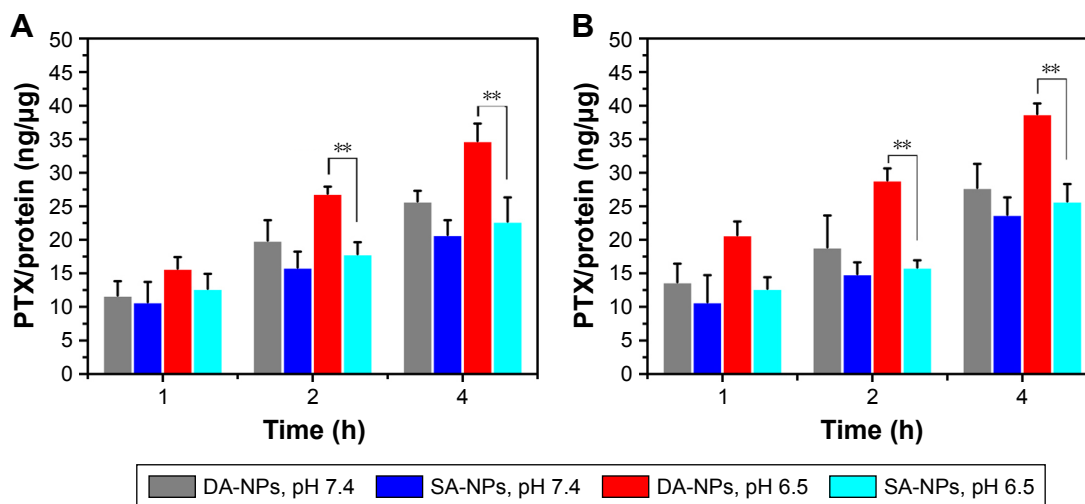


Figure 10 The accumulation of PTX at pH 7.4 and 6.5 in MCF-7/ADR (A) and MCF-7 (B) after incubated with DA-NPs and SA-NPs with different times.

Note: The results are expressed as mean \pm SD, n=4, **p<0.01.

Abbreviations: DA-NPs, DSF and PTX co-loaded micelles with charge reversal; NPs, nanoparticles; PTX, paclitaxel; SA, succinic anhydride.

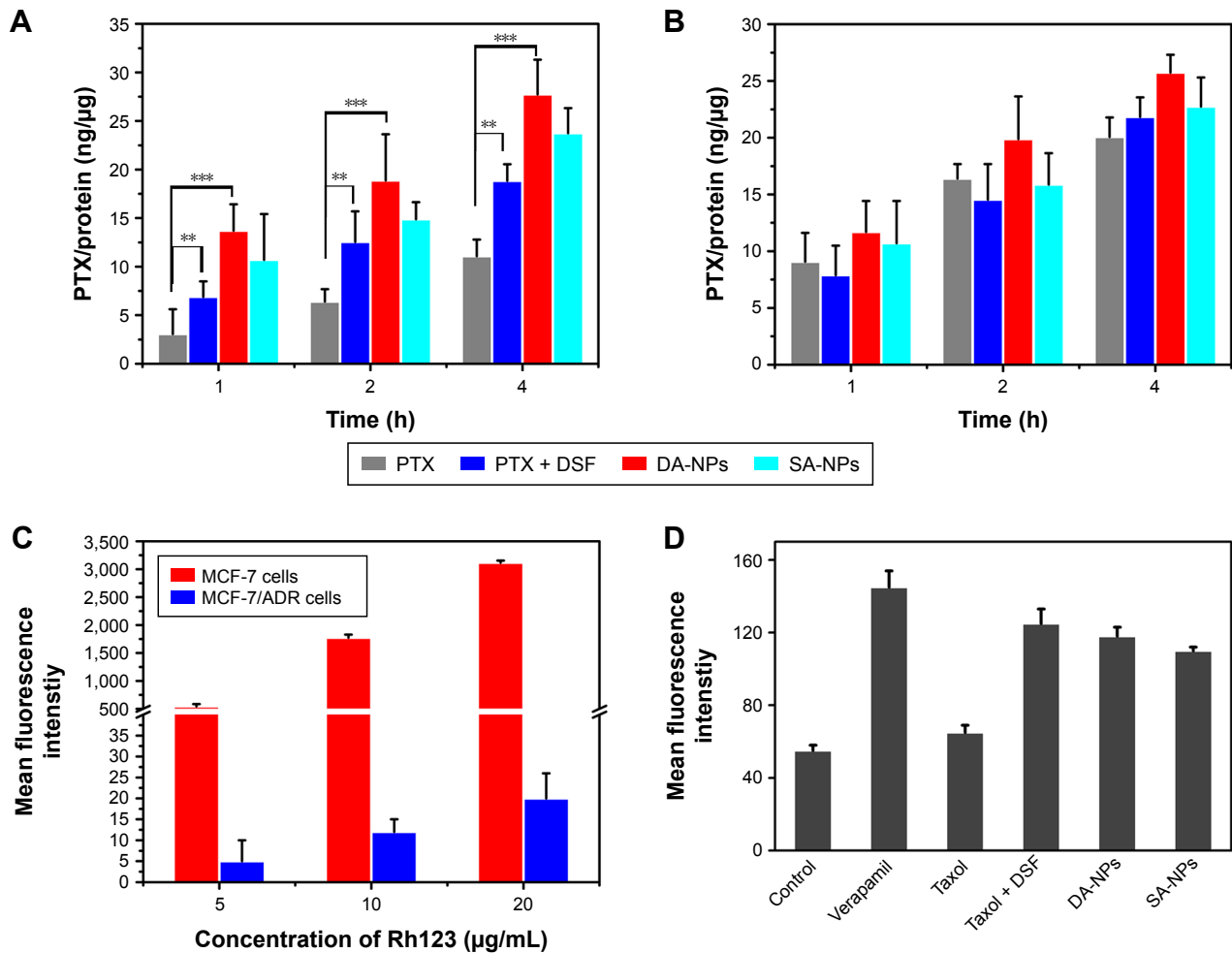


Figure 11 The accumulation of PTX in MCF-7/ADR cells (A) and in MCF-7 cells (B) incubated with PTX, PTX + DSF, DA-NPs and SA-NPs for different times. (C) Cellular uptake of Rh123 in MCF-7 and MCF-7/ADR cells without treatment. (D) Rh123 retention in MCF-7/ADR cells treated with verapamil, taxol, taxol + DSF, DA-NPs and SA-NPs. The results are expressed as mean ± SD, n=4, **p<0.01, ***p<0.001.

Abbreviations: DA-NPs, DSF and PTX co-loaded micelles with charge reversal; DSF, disulfiram; NPs, nanoparticles; PTX, paclitaxel; Rh123, rhodamine 123; SA, succinic anhydride.

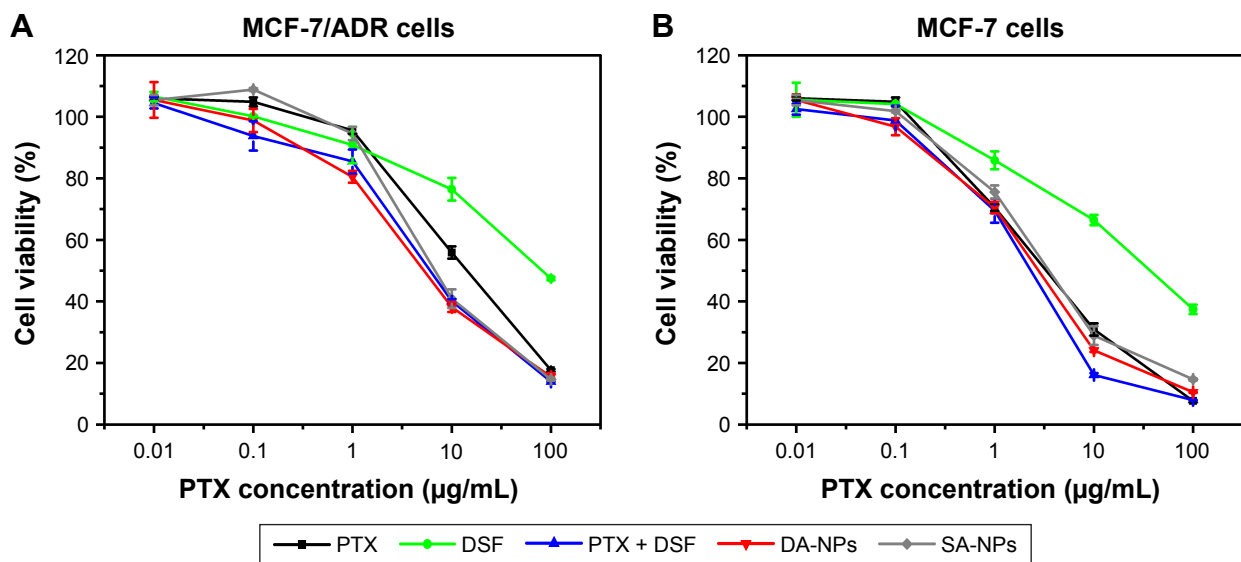


Figure 12 (Continued)

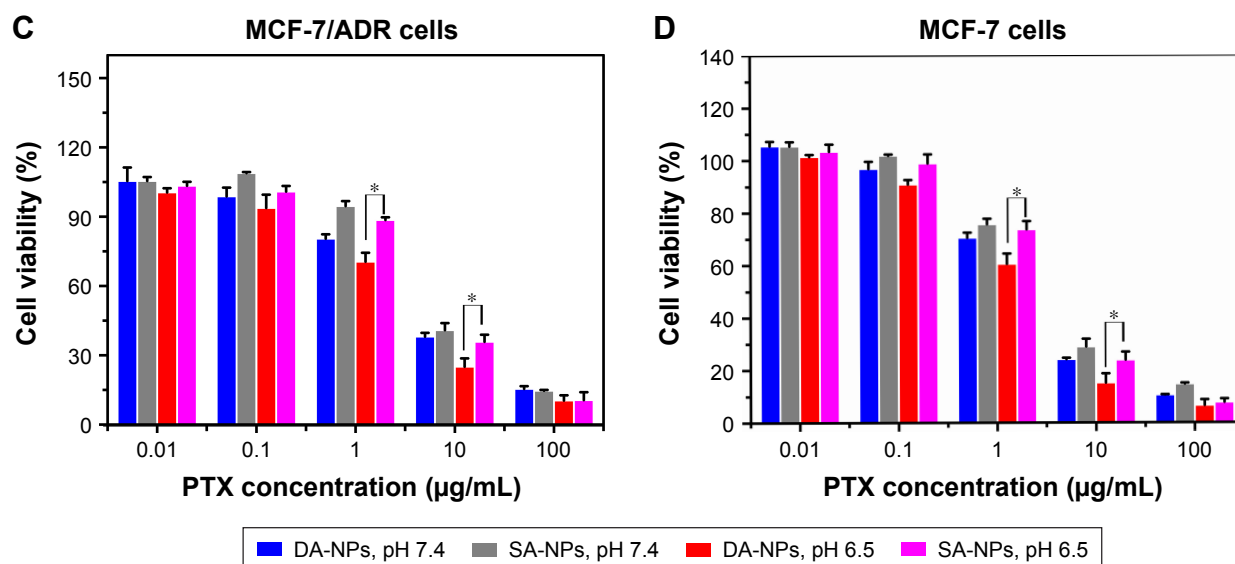


Figure 12 The cell viability of MCF-7/ADR cells (A) and MCF-7 cells (B) incubated with PTX, DSF, PTX + DSF, DA-NPs and SA-NPs for 48 h; and the cell viability of MCF-7/ADR (C) and MCF-7 cells (D) treated with DA-NPs and SA-NPs at pH 6.5 or 7.4 for 4 h and further incubated of 48 h.

Note: Data are shown as mean \pm SD, $n=5$, $*p<0.05$.

Abbreviations: DA-NPs, DSF and PTX co-loaded micelles with charge reversal; DSF, disulfiram; NPs, nanoparticles; PTX, paclitaxel; SA, succinic anhydride.

showed stronger antitumor effect to multidrug-resistant cells and has a great potential to achieve better therapeutic effects in cancer treatment.

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Disclosure

The authors report no conflicts of interest in this work.

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Supplementary materials

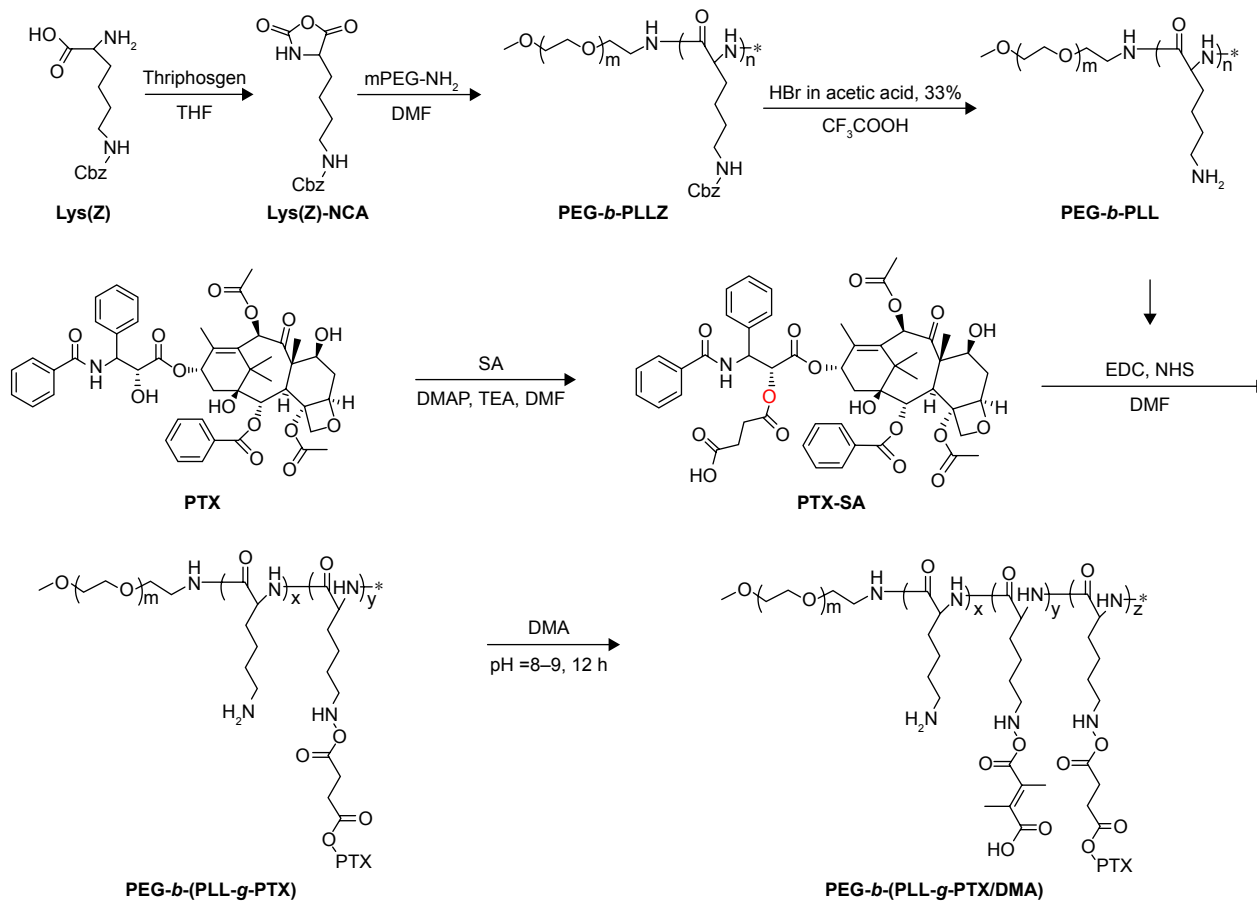


Figure S1 Synthesis of the DMA modified PTX conjugated polymers PEG-b-(PLL-g-PTX/DMA).

Abbreviations: DMA, 2,3-dimethylmaleic anhydride; DMAP, 4-dimethylaminopyridine; DMF, *N,N*-dimethylformamide; PEG-b-PLL, poly(ethylene glycol)-*block*-poly(L-lysine); PEG-b-PLLZ, poly(ethylene glycol)-*block*-poly- ϵ -(benzyloxycarbonyl)-L-lysine; PTX, paclitaxel; SA, succinic anhydride; TEA, triethylamine; Lys(Z)-NCA, ϵ -(Benzyloxycarbonyl)-L-lysine *N*-carboxyanhydride; EDC, 3-(ethyliminomethylideneamino)-*N,N*-dimethylpropan-1-amine; NHS, *N*-Hydroxysuccinimide.

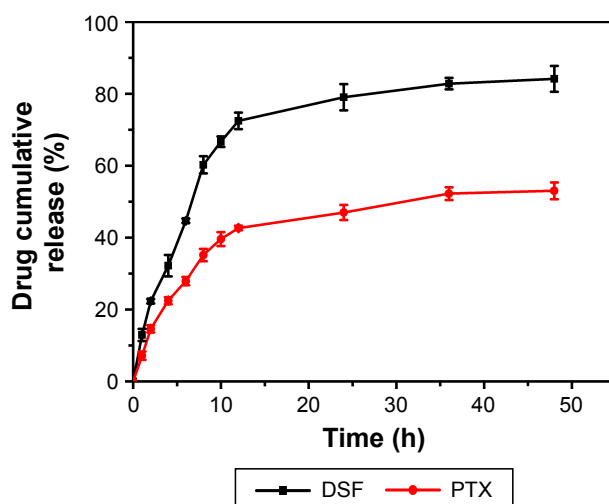


Figure S2 The cumulative release of PTX and DSF at pH 5.0.

Abbreviations: DSF, disulfiram; PTX, paclitaxel.

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