

Capping gold nanoparticles with albumin to improve their biomedical properties

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Abstract: Nanotechnology is an emerging field which has created great opportunities either through the creation of new materials or by improving the properties of existing ones. Nanoscale materials with a wide range of applications in areas ranging from engineering to biomedicine have been produced. Gold nanoparticles (AuNPs) have emerged as a therapeutic agent, and are useful for imaging, drug delivery, and photodynamic and photothermal therapy. AuNPs have the advantage of ease of functionalization with therapeutic agents through covalent and ionic binding. Combining AuNPs and other materials can result in nanoplateforms, which can be useful for biomedical applications. Biomaterials such as biomolecules, polymers and proteins can improve the therapeutic properties of nanoparticles, such as their biocompatibility, biodistribution, stability and half-life. Serum albumin is a versatile, non-toxic, stable, and biodegradable protein, in which structural domains and functional groups allow the binding and capping of inorganic nanoparticles. AuNPs coated with albumin have improved properties such as greater compatibility, bioavailability, longer circulation times, lower toxicity, and selective bioaccumulation. In the current article, we review the features of albumin, as well as its interaction with AuNPs, focusing on its biomedical applications.

Keywords: albumin, drug delivery, photothermal therapy, theranostics

Introduction

Nanotechnology is an emerging field which has provided a wide range of opportunities in applied sciences such as materials science and medicine.¹ Novel therapies for chronic diseases, including cancer, aim to overcome the adverse effects caused by the lack of specificity of conventional treatments, thereby improving drug administration.² In this context, a new and interesting branch of nanotechnology has recently emerged which involves the use of noble metal nanoparticles covered by albumin.³

Gold nanoparticles (AuNPs), of different geometric shapes, can be used for spatially and temporally controlled drug release, thereby taking advantage of their biocompatibility, size and ease of functionalization, in addition to their optic, plasmonic and photothermal properties.⁴ Optic properties in AuNPs can be modulated by controlling their shape, size, and surface modification through synthetic approaches. A wide variety of AuNP shapes have been reported, including gold nanospheres (AuNSs), nanorods (AuNRs), nanoprisms (AuNPrs), nanoclusters (AuNCs), nanoflowers and core-shells.⁵⁻⁷ Irradiation of AuNPs in their plasmonic band, allows them to absorb and dissipate the energy as local heat, which can be exploited in biomedical applications, including photothermal therapy and drug release.⁸

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Localized surface plasmon resonance (LSPR) on AuNPs is highly dependent on their shape, size, and the surrounding medium, and in this context they are useful for biosensing and detection.⁹ However, their unique optical properties allow not only transport and release of molecules with therapeutic properties, but also theranostic applications.¹⁰ The term “theranostic” was coined in 2002 and is defined as a material that combines the modalities of therapy and diagnostic imaging. Therefore, theranostics deliver therapeutic drugs and diagnostic imaging agents at the same time.¹¹ Small nanoparticles can easily scatter light and be detected through techniques as simple as dark field microscopy to more complex methods such as computerized tomography.⁹

In a complementary way, AuNP functionalization and coating with endogenous proteins can increase the circulation time, biocompatibility, selective uptake and biodistribution of AuNP-based nanosystems.^{12,13} Serum albumin (SA) is the most abundant protein in plasma. SA includes a lot of amino acids with charged functional groups, such as carboxyl, amino and sulfhydryl,¹⁴ which allow it to have several binding sites for attachment to therapeutic systems, including polyconjugated dyes, drugs, and nanoparticles.² In addition, SA can undergo cell uptake through specific receptors in tumor cells^{15,16} as a nutrient and amino acid source, thereby increasing the bioavailability of loaded agents.^{17–20}

AuNPs can be capped with SA molecules or encapsulated into SA nanocapsules, thereby giving rise to multifunctional nanoplateforms, which can transport drugs and deliver them to target sites.^{21,22} Furthermore, the AuNP-SA system can be used in other biomedical applications, as described in the current review, including drug delivery, photothermal therapy, diagnostics and theranostics.

Recently, numerous studies concerning AuNP-SA-based systems have been reported, which represent the beginning of a promising nanoplateform for biomedical applications. Albumin has been employed extensively for pharmaceutical applications and also in clinical trials. For a more extensive discussion of the applications of albumin the reader can consult illustrative literature.^{2,15,16,23–28} With regard to AuNPs-SA, we believe that further developments are possible, and in the future these systems will be trialed clinically.

In the current review, we present a description of SA, a conventional but promising material in nanocarrier design, in addition to discussing AuNPs as a new and

versatile material with fascinating properties, as well as forefront approaches for the creation of hybrid AuNP-SA systems.

AuNP properties for biomedical applications

Nanoparticles are defined as particles with sizes that lie between 1 and 100 nm, and which are formed in two or three dimensions. AuNPs exhibit a single structure, consisting of a highly symmetric face center cubic crystalline network.²⁹ AuNPs are characterized by specific geometrically dependent surface properties, as well as susceptibility to several environmental *stimuli*, such as light or heat.³⁰ Existing methodologies allow for AuNPs of diverse geometries, and the most common in biomedical applications are nanospheres, nanorods, nanoshells, nanocages and nanoprisms. **Figure 1** shows a pictorial representation of the above-mentioned systems.

For noble metal particles, where the particle size is smaller than the incident wavelength, the properties of the surface of the material influence interactions with electromagnetic radiation, and such interactions usually occur in visible (Vis) and near infrared (NIR) wavelengths.³¹ Interestingly, AuNPs can absorb electromagnetic radiation and convert it to thermal energy through photoexcitation mechanisms, thereby releasing heat efficiently. The energy emission and absorption properties depend on the particle shape and size.

Optical properties of AuNPs

The unique properties of AuNPs are caused by the electromagnetic effect of the collective oscillation of the surface of the electrons (conduction electrons). When conduction electrons interact with light (**Figure 2**), the restoring force, ie Coulombic attraction, from the electrons to the nuclei in the crystalline network, establishes a resonating condition in a relatively tight spectral band,

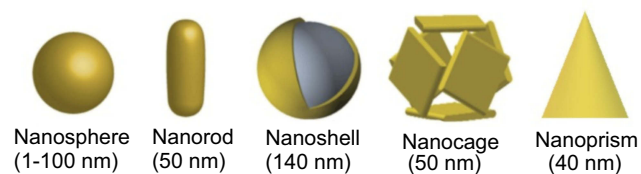


Figure 1 Different shapes of gold nanoparticles used in biomedical applications. Reprinted from *Journal of Experimental & Clinical Medicine*, 6, Ajnai G, Chiu A, Kan T, Cheng C-C, Tsai T-H, Chang J, Trends of gold nanoparticle-based drug delivery system in cancer therapy, pages 172-178, Copyright (2014) with permission from Elsevier.⁴⁷

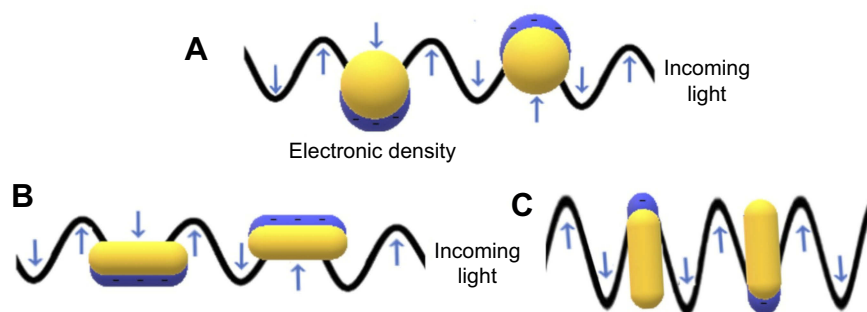


Figure 2 LSPR on gold nanoparticles, generated by the interaction between the conduction electrons on gold surface and the incident light on **(A)** non-anisotropic spherical AuNP, **(B)** longitudinal and **(C)** transversal anisotropic rod-shape AuNP.

the so-called LSPR, or surface plasmon.³² In nanoparticles that possess this property, the excitation of the surface plasmon results in substantial electromagnetic fields at the surface of the nanoparticles, which strongly outweigh the intensity of the incident light field.⁴

The electron density at the surface of the nanoparticle depends on its size, shape, metal structure and the surroundings, which directly affects the location of the surface plasmon in the electromagnetic spectrum (ES). In spherical AuNPs (AuNS), for example, a strong absorption band centered in the visible region (approximately 520 nm) occurs because the frequency of the electromagnetic wave of the incident light is resonant with the collective oscillation of the conduction electrons, creating the reddish tones observed in spherical AuNPs in solution. In contrast, for anisotropic AuNPs, such as rod and prism shaped (AuNRs and AuNPrs) nanoparticles, more than one band appears in the ES. AuNRs have two bands of surface plasmon resonance, which correspond to the resonant electrons in the transverse and longitudinal bands (less and

more intense, respectively).^{33–35} Figure 3 shows a comparison between the spectra of AuNSs and AuNRs.

The LSPR produces unique effects in metallic nanoparticles, such as the photothermal effect through phonons, in which energy is transformed into the vibration of crystalline structures. Absorbed photons are converted into phonons, in a process that involves a rapid relaxation of electrons and phonons, followed by phonon-phonon relaxation, which results in an increase in the system temperature and conduction to its surroundings, thereby producing an increase in local temperature (Figure 4).^{36–39} The temperature increase produced at the AuNP surface can be used in diverse therapeutic applications, and in particular, photothermal therapy.

It is important to mention that for biomedical applications the system should have the least possible toxicity, which requires that the absorption of the plasmon be in the so-called biological window (700–1,100 nm). In this region of the spectrum, none of the radiation is absorbed

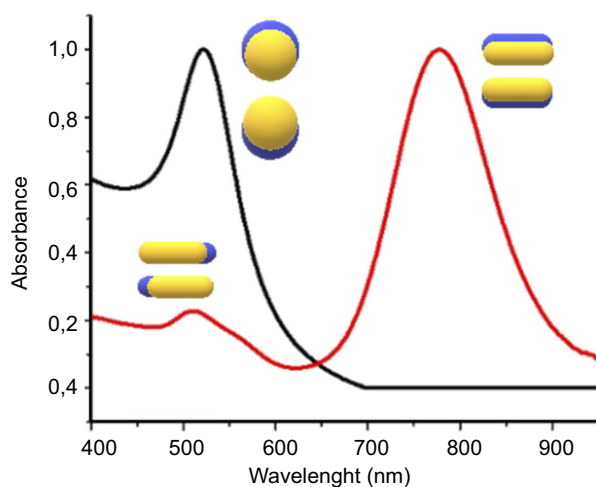


Figure 3 UV-Vis absorption spectrum AuNSs (black line), AuNRs (red line) and the oscillation modes on AuNPs.

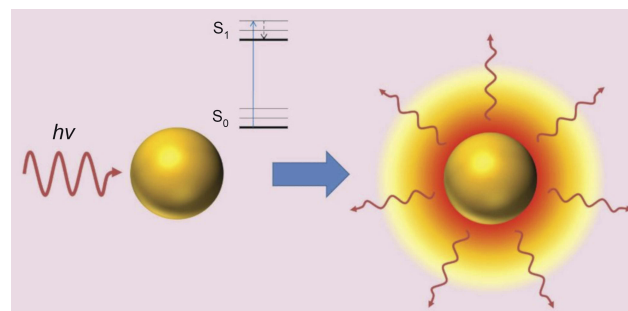


Figure 4 Photothermal effect on gold nanoparticles. After irradiation AuNPs absorb light, which leads to an electronic transition of the surface electrons from a ground state (**S**₀) to an excited state (**S**₁). The energy is released in the environment of the nanostructure as local heat. Republished with permission of Future Medicine Ltd, from Gold nanoparticles for photothermally controlled drug release, Guerrero AR, Hassan N, Escobar CA, Albericio F, Kogan MJ, Araya E, 9, 2014; permission conveyed through Copyright Clearance Center, Inc.³²

by the molecules present in human tissue, allowing the irradiation to penetrate tissues and generate a photothermal effect, without causing secondary damage.⁴⁰

The development of multimodal functional nanoparticles has gained attention as a way to produce promising nanomaterials for biomedical applications.^{41,42} AuNPs have been one of the most used building blocks, not only for their optical properties, but also because of their load capacity, low toxicity and modifiable surface.⁴³ AuNPs in combination with biomaterials have shown significant efficacy in the treatment of various diseases, such as Alzheimer's disease^{13,44,45} and several types of cancer.^{43,46,47}

An important issue that should be considered for the pharmaceutical use of nanoparticles is that when nanoparticles are intravenously administered, several plasma proteins bind to their surface, forming the so-called protein corona.⁴⁸ These capped nanoparticles can then be recognized by the macrophage cell surface and be internalized,⁴⁹ leading to a significant loss of the nanoparticles from circulation. A portion of the serum proteins that bind and cap the nanoparticles are termed opsonins. Nanoparticles capped with these proteins can be recognized and captured by macrophages, contributing to the loss of the majority of the administered dose because of retention by the reticuloendothelial system (RES).⁵⁰

It is important to mention that the protein corona can change over time to create different protein profiles.^{51,52} Modification of colloidal particles with polymers such as polyethyleneglycol (PEG) can increase their half-life in the blood⁴⁷ and also enhance the selective delivery of nanoparticles. For example, capping nanoparticles with endogenous serum proteins, such as apolipoprotein-B and/or ApoE, could support blood brain barrier penetration through transcytosis via a low-density lipoprotein receptor.^{53,54} Considering the importance of the protein corona for targeting purposes, it is important to reduce non-specific protein adsorption and favor the recruiting of specific proteins.

Serum albumin

SA is of great interest because it is the most abundant protein in plasma, and it has many important functions, including the regulation of blood pH and osmotic pressure.⁵⁵ Albumin is a globular protein, and is non-toxic, water soluble, stable from pH 4–9 and in organic solvents,²⁴ and exhibits several binding sites.¹⁴ Albumin acts as a solubilizing agent for fatty acids and detoxifies blood plasma, thereby reducing the levels of harmful substances such as heavy metals, reactive oxygen species and ions.²⁵ Furthermore, SA can transport and release many compounds, such as drugs, through the plasma.⁵⁵

Albumin sources

Albumin can be obtained from diverse sources, such as egg (ovalbumin, OVA), bovine serum (bovine serum albumin, BSA), human serum (human serum albumin, HSA), rat serum, soy, milk and grains.^{24,56} OVA, BSA and HSA are most commonly used for biophysical and biochemical studies, and all are commercially available. They have the advantages of great stability, biocompatibility, and pH and low temperature sensitivity.⁵⁷ Albumin is chemically attractive because of its disulfide bonds and sulfhydryl groups, which allow interaction with organic and inorganic ligands. Table 1 presents some properties and advantages of OVA, BSA and HSA.

Structural considerations

OVA, BSA and HSA demonstrate similarities, whereby all three are formed mainly of an α -helix with a globular formation, contain hydrophilic and hydrophobic sites, and demonstrate an acid nature. There are, however, numerous differences. Structurally, OVA consists of a single polypeptide chain of 385 amino acids, with a serpin-like structure⁵⁸ and a helical reactive loop arrangement.²⁴ OVA is widely used in the food industry because of its ability to foam and form gels.⁵⁹

Table 1 Properties of OVA, BSA and HSA

Type of albumin	Isoelectric Point	Molecular Weight (KDa)	S-S Bridges	Free SH groups	Advantage
Ovalbumin	4,8	47	1	4	Low cost, can develop gel networks, emulsion stabilization. Unusual ligand-binding properties, intrinsic fluorescence, stability to pH and temperature, wide acceptance in pharmaceutical industry. ⁵⁷ Source of nutrition for peripheral tissues, avoid possible immunogenic response in vitro, stability to pH and temperature.
Bovine Serum Albumin	4,7	69,3	17	1	
Human Serum Albumin	4,7	66,5	17	1	

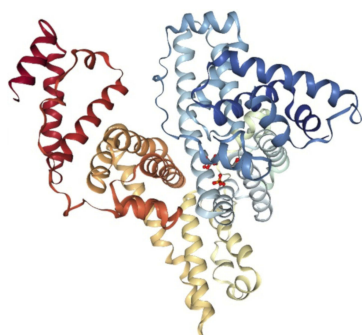


Figure 5 Crystal structure of BSA. Reprinted from Elsevier, 52(3-4), Majorek KA, Porebski PJ, Dayal A, et al. Structural and immunologic characterization of bovine, horse, and rabbit serum albumins, 174-182, Copyright 2012, with permission from Elsevier.⁵⁵

BSA (Figure 5) has 583 amino acids,⁶⁰ and this protein is made up of three homologous domains, which themselves are the product of two subdomains, and their structure is 67% α -helical. BSA is attractive because of its availability, binding affinity in the formation of ligand-protein complexes, intrinsic fluorescence through its two tryptophan residues (Trp-134 in the first domain and Trp-212 in the second domain),⁶¹ and structural and functional similarities to HSA.⁶²

Similar to BSA, HSA consists of 585 amino acids in a globular conformation, which contains three homologous domains in an α -helical ellipsoid shape.^{15,24,55} The amino acids in HSA can be dissimilated and used as a source of nutrition for peripheral tissues and preferentially undergo uptake by tumors and inflamed tissue.^{15,16} Moreover, HSA possesses different binding sites, which gives it the ability to transport drugs, hormones, fatty acids and ions through the body.^{24,63} These characteristics make it a good candidate for a drug delivery vehicle,⁵⁷ and for use in pharmaceutical preparations.

BSA and HSA

BSA and HSA are homologous in structure, conformation and properties. In fact, BSA and HSA share 76% sequential identity,²⁷ and the major difference lies in tryptophan (Trp) residue location. In HSA there is just one Trp, located at position 214, whilst BSA has a Trp-212 at subdomain IIA, and additionally a Trp-134 exposed in solvent at subdomain IB.⁶⁴

The important differences between OVA and the other two albumins are shown in Figure 5 and Table 1. The molecular weight, size, structure, S-S bridges and SH groups differ significantly between OVA and BSA/HSA.

These properties are related to binding abilities, surface modifications² and the formation of nanostructures, which will be the core issue in the current review. For this reason, we will focus only on BSA and HSA for biomedical applications.

Compared with other albumins, BSA and HSA are less immunogenic and they are considered to be well tolerated by humans¹⁵ and are used as a protein carrier in different delivery systems.⁶⁴ BSA and HSA are currently used in drug delivery approaches in which drugs are conjugated to the binding sites of the protein,⁶⁵ thereby improving water solubility, therapeutic efficiency, bioavailability, biocompatibility,⁶⁶ biodistribution⁶⁷ and reducing drug side effects,⁶⁸⁻⁷⁰ including for anti-inflammatory, chemotherapeutic¹⁹ and hypoglycemic drugs.^{18,28}

Capping AuNPs with albumin to increase their colloidal stability and to reduce their interaction with plasma proteins

Surface modification of nanoparticles generally provides better water solubility, increases colloidal stability, increases biocompatibility, modifies cell uptake and intracellular trafficking⁷¹ and improves blood circulation time compared with bare nanoparticles. Nanoparticles can be stabilized using a wide variety of molecules, such as peptides, proteins, DNA and polymers.⁷²⁻⁷⁴ Functionalization of AuNPs using PEG helps to diminish interactions with plasma proteins, reduces nanoparticle phagocytosis by the reticuloendothelial system with consequent accumulation in the liver and spleen, increases the half-life in circulation and allows selective delivery to tumors.^{74,75} Stabilization can also be achieved by capping AuNPs with albumin.⁷⁵ Capping is produced through electrostatic or hydrophobic interactions between nanoparticles and albumin or through chemisorption of the thiols to the gold surface.⁷⁶ For capping, it is possible to incubate AuNPs in a solution of the protein, taking into account various parameters, such as concentration, ionic strength and pH. Controlling the pH is necessary to regulate the degree of ionization of the species, considering the isoelectric point of albumin (approximately 5) and the zeta potential of the nanoparticles to allow electrostatic interactions. Albumin-capped nanoparticles can be stabilized through steric stabilization. BSA-encapsulated AuNPs (Au-BSA) have been used to load large amounts of methotrexate, which was useful for the delivery of this drug to breast cancer cells.⁷⁷

Albumin to favor the selective delivery of AuNPs

One of the exploitable features of cancer cells, in terms of increasing the selectivity of anticancer treatments, is their accelerated rate of proliferation, resulting in a higher demand for energy and nutrients.¹⁷ This demand can be supplied by increasing the protein consumption and metabolism rate. Since albumin is the predominant plasmatic protein, it is the primary nutrition source for tumor cells.^{18–20} This can be seen by the fact that advanced cancer patients have a low level of serum albumin (hypoalbuminemia), as a result of the widespread need for tumors to be supplied with amino acids.²⁵

For drug delivery, there are two predominant ways in which nanosystems can reach a tumor site; these are *active targeting* and *passive targeting*. The first involves the interaction of a nanosystem with overexpressed or specific receptors in a tumor, thereby resulting in capture and further internalization at the tumor site or inside cancer cells. In this manner, albumin interacts with several specific receptors, including glycoproteins (Gp18, Gp30 and Gp60) and Secreted Protein Acidic Rich in Cysteine (SPARC), which is responsible for its recycling and transcytosis.⁷⁸

Because of the active targeting demonstrated by albumin, in addition to its excellent biocompatibility, several albumin-based formulations have been proposed to overcome the drawbacks of unmodified drugs, such as rapid clearance from the body, poor solubility, and a short-half life. As a result, the FDA has approved several albumin-based drugs, including Abraxane, Levimir and Victoza, which are currently used in medical treatments for cancer and diabetes. It has been shown

that these albumin-conjugated drugs can accumulate in inflamed tumors and tissues modulated by GP60 and SPARC receptors, as shown in Figure 6.^{15,16} Moreover, through BSA conjugation, drug biodistribution can be increased,⁶⁷ and release can be promoted through protein digestion.⁶⁶ In the case of albumin nanoparticles conjugated to paclitaxel, as is the case with Abraxane (clinically probed), the selective effect on tumor cells is achieved by using the endogenous mechanism for delivering proteins to cells. Albumin binding to the glycoprotein receptor gp60 on endothelial cells results in activation of caveolin-1 and the transcytosis of intact nanoparticles across the cell membrane. In addition to the active albumin-mediated transport of drugs into tumor cells, a degree of tumor-selective targeting is provided by SPARC, a protein which modulates the interaction of cells with the extracellular matrix and is over-expressed in many cancer types.^{79,80} Although associated with tumors, SPARC is not tumor-specific since it is found in healthy tissues, especially during embryonic development.⁸¹ Increased levels of SPARC are associated with tissue and bone remodeling and hyperproliferation, but a pronounced increase in expression is seen with malignant transformation. SPARC is present on the surface of MX-1 human mammary carcinoma cells.⁸¹ It is thought that the SPARC-mediated concentration of paclitaxel-carrying albumin molecules in the vicinity of tumor cells could lead to locally high levels of drug release.

In contrast, *passive targeting* refers to the accumulation of macromolecules through the enhanced permeability and retention effect (EPR), which in summary, refers to angiogenesis-generated anatomic defects in tumor blood vessels. Tumors demonstrate increased permeability, because

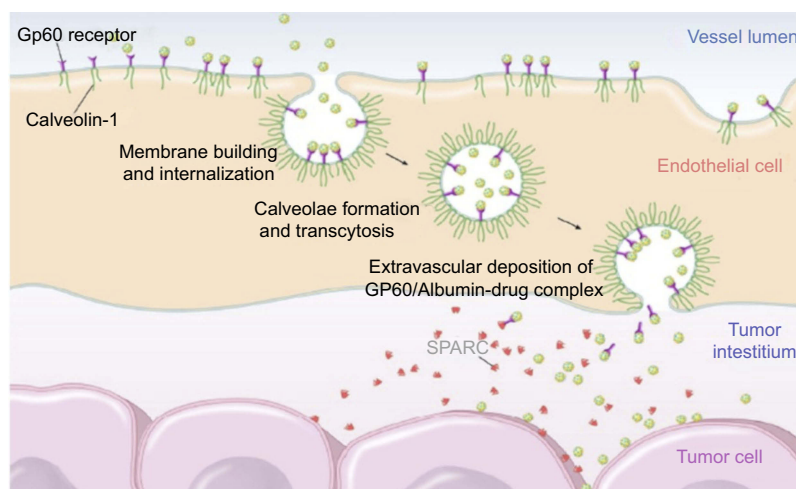


Figure 6 Albumin uptake in tumor interstitial mediated by transcytosis with GP60 and subsequent binding to SPARC in the extracellular tumor matrix. Reprinted from *Journal of Controlled Release*, 157, Elsadek B, Kratz F, Impact of albumin on drug delivery - New applications on the horizon, Pages 4-28, Copyright (2012), with permission from Elsevier.⁹⁹

of intercellular gaps in blood vessels, which can reach up to 4 μm (in healthy tissues such gaps are typically less than 5 nm),^{82–84} and which allow for the extravasation of macromolecules and nanoparticles of more than 10 nm, including viruses and bacteria.

The EPR effect depends on the size of drugs.^{83,85} Small drugs can freely diffuse in/out of the tumor blood vessels without accumulating, while objects between 20 and 200 nm¹ can enter the tumor interstitial space without returning to the bloodstream, because of their large size.

Albumin, in addition to being used to load therapeutic and imaging agents, is also useful for surface modification of numerous types of nanoparticle, as shown in Figure 7. Surface modification of nanoparticles generally provides greater water solubility, and increased biocompatibility and blood circulation time, compared with uncoated nanoparticles.²

Interesting functional drug delivery systems can be obtained from AuNPs, which possess numerous useful properties such as a high surface area, protection of cargo molecules from degradation, and energy absorption and emission dependent on their size, shape and surface modification, which will be subsequently described.^{32,70,86} However, nanoparticles with a suitable size to allow for accumulation through passive targeting, if combined with SA, may demonstrate a dual ability to be captured through passive and active targeting, and therefore represent useful nanoplatforms for drug delivery with significant physiological stability and enhanced half-life compared with albumin or AuNPs alone.⁸⁷

AuNPs and albumin

Nanotransporters have provided a wide range of opportunities in the area of drug administration.⁸⁸ AuNPs offer the possibility of trapping and transporting poorly soluble or minimally biocompatible drugs, thereby improving circulation time, biocompatibility and, most

importantly, preferential accumulation at a disease site.⁸⁹ Metallic nanoparticles, such as magnetic nanoparticles, plasmonics and quantum dots, have various properties that make them useful for biomedical applications.

SA interaction with AuNPs induces structural changes in SA, which are dependent on the size, shape and surface modification of the AuNPs. AuNP size is an important aspect to consider in AuNP-SA interactions, and small nanoparticles have a more efficient interaction with SA compared with larger nanoparticles.^{90–93} However, nanoparticle shape also has a strong influence on protein adsorption in AuNPs, as demonstrated by Moustouai et al,⁹⁴ who assessed protein interaction in AuNSs and branched-shaped nanoparticles (AuNU). AuNSs showed an increase in the hydrodynamic radius, indicating that each type of protein binds on the gold nanoparticle in a specific orientation, and AuNUs showed different protein orientations because of their multi-oriented surfaces (tips) with a higher surface to volume area.

In another study, Chakraborty⁶ et al showed that AuNRs and AuNSs induce variations in the stability, structure and function of SA. In this work, a spectroscopic analysis comparing the interaction between SA and two nanoparticle geometries, namely that of AuNRs and AuNS with either Cetyltrimethylammonium bromide (CTAB) or PEG, was performed. Variation in the HSA structure was found to be higher in the order HSA-CTAB-AuNRs > HSA-CTAB-AuNSs > HSA-PEG-AuNRs > HSA-PEG-AuNSs. Accordingly, a rod shaped structure showed greater levels of denaturation than a spherical counterpart. However, the presence of CTAB on the surface resulted in a higher aggregation level, and was correlated with a positive charge, whereas PEGylated AuNPs have a near zero charge because of the neutral charge of PEG.^{39,95}

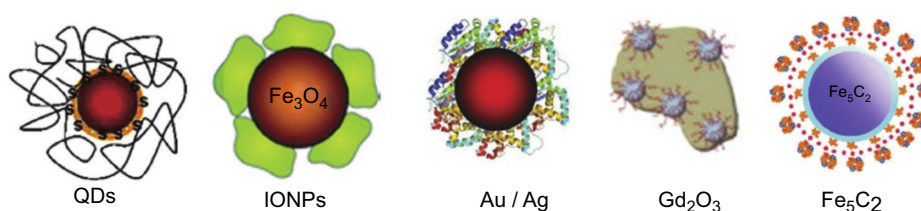


Figure 7 Nanoparticles surface modification with albumin. Copyright © 2016. John Wiley and Sons. Reproduced from Chen Q, Liu Z. Albumin carriers for cancer theranostics: a conventional platform with new promise. *Adv Mater.* 2016;28:10557–10566.²

Interestingly, shapes with planar surfaces have reduced accessibility and therefore less interaction with SA, as demonstrated by Carnovale et al⁹⁶. In this novel study, the interaction between SA and AuNPs with different geometries was compared, namely AuNSs, AuNRs, AuNPrs and AuNCs. It was found that AuNRs have higher adsorption constant values, while geometric shapes with large flat planar surfaces, such as prisms and cubes, presented lower adsorption values.

Strategies for AuNP conjugation with albumin

The characteristics of its structural domains allow SA to bind to different photothermal agents, including AuNPs. The resulting structures can be observed as AuNPs with a protein “halo” or take the form of nanocapsules in which AuNPs are confined in a cavity surrounded by albumin.⁶⁷

The first case, in which AuNPs present a halo, is attributable to SA attaching to AuNPs through physical or chemical adsorption, as summarized in Figure 8. This can be achieved in several ways:

- **Passive adsorption:** in which SA charged functional groups, or specific atoms are attached to the gold surface through covalent or non-covalent interactions. An example of this is the interaction between SH groups in the SA cysteine residues and Au atoms on the AuNP surface, giving rise to Au-S covalent bonds.⁶⁸ As a result of the binding abilities of SA, direct adsorption is usually achieved through simple incubation of an AuNP solution in the presence of SA. This adsorption strategy is remarkable because of its methodological simplicity and economy, avoiding the use of additional reagents and drastic conditions.
- **Active adsorption:** this process is a bit more complicated because it involves the use of modified SA, with the aim to increase the amount or quality of interactions between SA-Au or SA-surface molecules on the AuNPs. Modification of SA can be achieved through functionalization, giving rise to carboxylated or thiolated SA or changes in the total charge, resulting in cationic or anionic SA.^{27,97} This is a convenient strategy when the AuNP surface is functionalized to other molecules or polymers, and an SA coating is required; for example, for an AuNP-PEG coating, cationic BSA can improve the covering.⁶⁷
- **In situ:** this strategy refers to the use of albumin, either as a synthesis reagent (eg as a reducing agent) or as a building block for AuNP synthesis, where SA is added during the synthesis procedure resulting in AuNPs with an SA coating.^{2,98} Several studies report the use of SA as a foaming and stabilizing agent or even as a template for AuNC synthesis. As a result, AuNC-SA hybrids can be obtained through one-pot synthesis.^{2,98} However, SA nanocapsules are also useful as a drug delivery platform. The use of SA encapsulation methods offers some benefits, including the loading of poorly soluble agents and the protection of cargo from natural degradation.^{2,23,69,99} Some of the most common procedures for SA nanocapsule creation are shown in Figure 9 and are detailed below:
- **Desolvation cross-linking:** This method, also known as a coacervation process, is mainly used to produce core-shell SA-AuNPs, as shown in Figure 9A. In practice, AuNPs are added to an SA solution as nuclei for capsule formation, after organic solvent (usually methanol, ethanol or acetone) is added dropwise until a turbid solution is observed, in which SA is

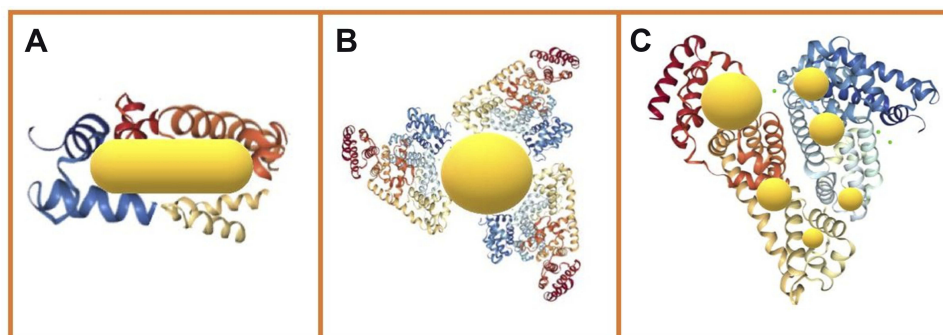


Figure 8 Adsorption of SA on AuNP surface can be achieved by: (A) passive adsorption of SA constituent functional groups attached to gold surface; (B) adsorption of activated SA modified with functional groups improving the surface binding properties; (C) growing of AuNPs embedded on SA structure after in situ synthesis.

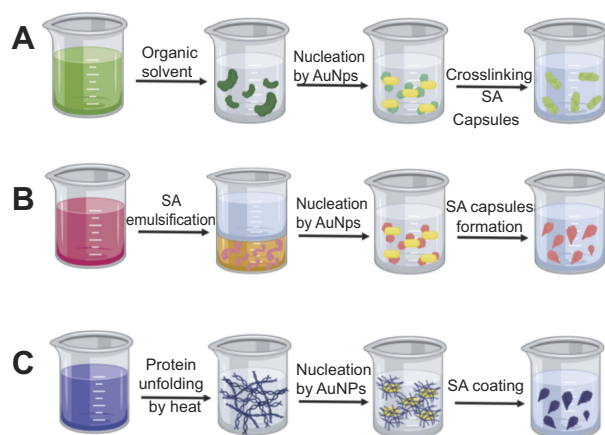


Figure 9 Methods for SA encapsulation of AuNPs. (A) desolvation method using an organic solvent and a crosslinker agent, (B) emulsification in oil phase, (C) thermal gelation by heat promoted unfolding of SA.

desolvated. Therefore, water solubility decreases considerably, giving rise to phase separation, and resulting in albumin aggregates. The desolvated albumin has exposed amine groups, which can be cross-linked by crosslinking agents, such as glutaraldehyde. Advantageously, this strategy allows chemical agents (including nanoparticles, drugs and diagnostic agents) to be physically entrapped inside SA capsules. SA nanocapsules are highly stable and protect entrapped agents against degradation.^{24,63,100}

- **Emulsification:** Albumin has a wide variety of amino acids, some of which are hydrophobic. In a conventional emulsification procedure, a SA solution is mixed with a non-aqueous phase (oil), for shearing and high-pressure homogenization, giving rise to an emulsion. AuNPs functionalized with hydrophobic molecules can be used as seeds for interactions with hydrophobic amino acids in albumin in this procedure, generating a core-shell. For nanocapsule stabilization, either cross-linking agents or stirring at high temperatures can be used. Finally, the solvent is removed through low pressure evaporation.^{23,27} This is a methodology mainly used for the encapsulation of lipophilic drugs, and increases biocompatibility and water solubility.¹⁰¹
- **Thermal gelation:** Protein structure is susceptible to temperature changes. In thermal gelation, a SA aqueous solution is heated to produce protein unfolding, which induces protein-protein interactions through hydrophobic, electrostatic, disulfide and hydrogen bonding. Unfolding also gives rise to SA interactions with AuNPs, resulting in self-assembly on the AuNP surface and protein coating.¹⁰¹

Biomedical applications of albumin-AuNPs

Nanotransporters have provided a wide range of opportunities in the area of drug administration.⁸⁸ Metallic nanoparticles, such as magnetic nanoparticles, plasmonics and quantum dots, have interesting optic properties that make them useful for biomedical applications.¹⁰² AuNPs offer the possibility of trapping and transporting poorly soluble or minimally biocompatible drugs, thereby improving circulation time, biocompatibility and, most importantly, preferential accumulation at a disease site.⁸⁹ An SA coating on the surface of AuNPs increases stability, targeting and biocompatibility.

The biomedical applications discussed in the current review are summarized in Table 2 and are detailed in the following sections.

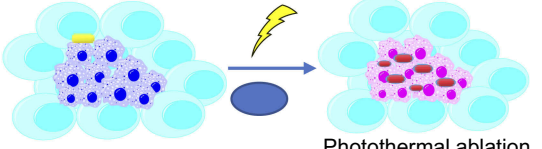
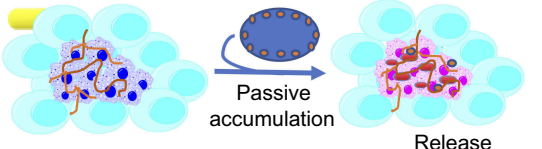
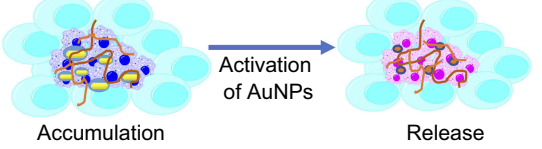
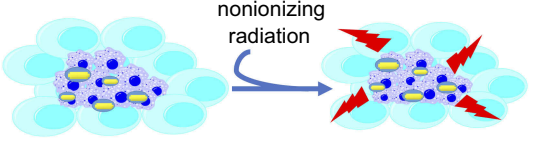
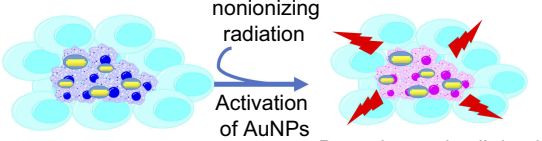
Albumin-AuNPs for photothermal therapy








Photothermal therapy involves the use of photoactive metals, which can convert light into heat through a previously described photothermal effect. The released heat kills malignant cells, resulting in weakening or removal of a treated tumor. Several studies have been conducted on photothermal therapy, and include the use of spherical AuNPs,^{76,103} nanorods, nanoshells¹⁰⁴ and nanostars¹⁰⁵ bound to SA. These nanobioconjugates achieved high stability, selective accumulation in malignant tissue, and selective photothermal ablation through laser irradiation.

An example of an AuNP photothermal system was reported by AL-Jawad et al,⁶ where AuNPs (size <5 nm) covered with BSA were synthesized and administered to rhabdomyosarcoma murine fibroblasts (L20B) and RAW 264.7 monocyte-macrophage cells. A drastic concentration-dependent photothermal ablation was achieved after a 532 nm and 800 nm irradiation, reaching more than 74% cytotoxicity for all tested cell lines. In another study, Mocan et al⁷⁶ developed an original model of liver hepatocellular carcinoma, and tested the effect of AuNPs-BSA. The study showed selective accumulation of the nanosystem in tumor cells, as well as apoptosis induced by laser irradiation at 808 nm. Zhang et al¹⁰⁶ synthesized highly stable AuNRs functionalized with BSA and administered them to fibroblast L929 and gastric adenocarcinoma YCC-2 cells in vitro, resulting in significant anticancer efficacy through NIR hyperthermia.

The inclusion of targeting agents in nanosystems results in improved properties, and increased selectivity

Table 2 Biomedical applications of gold nanoparticles

Photothermal therapy	Photothermal effect has place on AuNPs after irradiation, increasing the local temperature. SA acts carrying, directing and increasing the AuNPs biocompatibility. The released heat kills or weaken malignant cells by hyperthermia.	 <p style="text-align: right;">Photothermal ablation</p>
Drug delivery	Spatially controlled: AuNPs-SA binds and carry insoluble, non-compatible or small drugs, increasing the transport and accumulation properties specifically on malignant tissues by passive targeting.	 <p style="text-align: right;">Release</p>
	Temporally controlled: release of loaded drugs on AuNPs surface can be controlled after an external stimulus, such as light, either through surface reactions or partial decomposition of SA capsules which contains AuNPs and drugs.	 <p style="text-align: right;">Release</p>
Diagnostics	After AuNP-SA accumulation on malignant tissues, imaging techniques such as PA can detect localize the damaged site.	 <p style="text-align: right;">Detection and cell death</p>
Theranostics	AuNP-SA can act as a dual system, allowing not only the detection of a malignant site through imaging techniques, but also the treatment in situ.	 <p style="text-align: right;">Detection and cell death</p>

Notes:  Normal cells,  Malignant cells,  Death cancer cells,  Irradiation,  AuNP-SA,  Photothermal effect,  Carried drug.

and uptake by malignant cells. Folic acid (FA) is a widely used cancer targeting agent, since cancer cells overexpress FA receptors. This strategy was demonstrated in a study by Li et al¹⁰⁰ in which AuNSs were synthesized and conjugated with BSA-FA, resulting in a highly stable BSA-FA-AuNS system with low toxicity. The system underwent cell uptake in HeLa cells because of overexpressed FA receptors, and cell ablation was possible after laser irradiation, reaching 70% and 75% toxicity after NIR and green laser irradiation, respectively.

Aside from FA, other molecular systems can be used to promote the cell uptake of nanocapsules, including AuNR-BSA encapsulated in ssPalmM:DOPE:cholesterol lipid, as described by Paraiso et al.¹⁰⁷ *In vitro* assays with this nanosystem were performed in 4T1 breast cancer cells, in which cell viability decreased from 100% in the absence of irradiation to 11% after NIR irradiation, and an increase in dosage killed all the cells. Cell uptake played a key role in these results, since without lipid encapsulation, the AuNR-BSA system showed a plateau in killing

efficiency (30% cell viability after irradiation), showing the suitability of the system for hyperthermia.

However, nanoparticle-based photothermal therapy must be carefully assessed before conducting pre-clinical trials, since it is necessary to consider the impact and harm caused by laser irradiation; this is particularly relevant for systems in which irradiation is performed at wavelengths outside the biological window.^{6,108,109} In such cases, the risk of radiation-induced damage makes it difficult to discern if cell death is caused by hyperthermia through the photothermal effect or by the laser itself. However, even though photothermal therapy works by killing a high percentage of malignant cells by itself, combined approaches including drug delivery and photodynamic therapy controlled by external stimuli provide added value to photothermal systems.^{36,110}

Albumin-AuNPs for drug delivery

One of the issues that nanotechnology addresses is the lack of selectivity of existing drugs, since this problem is

one of the major causes of poor efficacy, besides the undesirable secondary effects associated with current treatments.

Biomedical applications involving AuNPs are mainly based on nanoparticles with spherical,^{22,39,44,111–113} nanorod,^{12,68,107,114–118} nanoprism^{5,119,120} and nanocluster geometry,^{6,121–125} however, current synthetic methods allow for the development of nanoparticles with different morphologies. Morphological changes can also modulate the physicochemical properties of AuNPs, which in combination with SA may be improved and exploited for therapeutic applications.

Because of its bioaccumulation, targeting, biocompatibility and surface properties, albumin can transport diverse therapeutic systems in a localized way, which is denoted “spatially controlled release”. The interesting optical properties of AuNPs allow them to be monitored using imaging tools, and the photothermal effect can release adsorbed molecules on the AuNP surface, thereby temporally controlling drug release. Therefore, albumin-AuNPs are a promising system for drug delivery.

The photothermal effect is frequently used for controlled drug release, as in Peralta et al,¹⁰ where AuNRs were synthesized, conjugated to the commercial anticancer drug paclitaxel (PAC) and encapsulated in HSA, and 3 µg of PAC/mg could be loaded into the system. The nanoplatform accumulated in 4T1 mouse breast cells through the EPR effect. After irradiation, the system generated approximately 94% cell death, through a combined photothermal effect and PAC release, as shown in Figure 10, which demonstrates that the system reached its highest

toxicity only after irradiation, as an external stimulus, and is therefore temporally controlled.

Callaghan et al¹²⁶ reported a controlled drug delivery system based on HSAP-AuNRs, for the tyrosine kinase inhibitor sorafenib (SRF). In cytotoxicity and uptake trials in renal carcinoma cells (RCCs), this system demonstrated a dual therapeutic action, with drug release and induction of photothermal ablation by hyperthermia. HSAP-AuNR-SRFs have potential for the treatment of advanced RCC. In an additional study by Liu et al,¹²⁷ AuNRs-HSA were used to encapsulate SRF and deliver it in a human metastatic cell carcinoma mouse model. The HSA-AuNR-SRF system produced 11.1% necrosis, while SRF alone produced only 4.2%; in contrast, after irradiation HSA-AuNR produced 62% necrosis and HSA-AuNR-SRF killed 100% of the cells, showing that AuNRs-HSA improved drug delivery compared with SRF alone. In addition to the ablation created by the photothermal effect, the drug load produced complementary effects, causing enhanced cell toxicity.

In another study, Wang et al¹²⁸ developed a delivery system for gemcitabine (Gem) based on AuNP@BSA. The system was synthesized in a facile one-pot reaction via the reduction of HAuCl₄ with BSA as a stabilizer, which was subsequently conjugated to Gem through electrostatic interaction to give rise to the AuNP@BSA-Gem system. In vitro assays in pulmonary carcinoma cells showed that the AuNP@BSA system demonstrated high biocompatibility and, as expected, a high cell death rate was produced by Au@BSA-Gem, even in comparison with free Gem.

Precise control of drug release is required to achieve optimal delivery of therapeutic agents; with this aim, Chiu

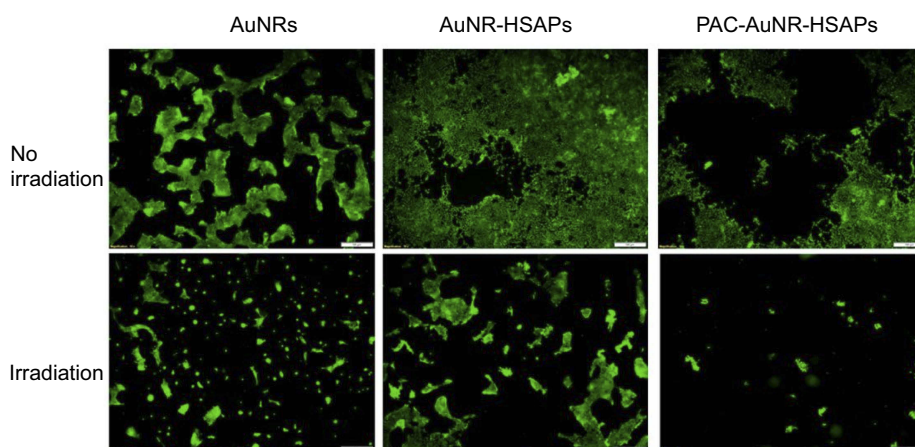


Figure 10 Fluorescence microscopy images of the cytotoxic effects on 4T1 breast cancer cells when treated with free AuNRs, AuNR-HSAPs or PAC-AuNR-HSAPs. (White scale bar denotes 100 µm). Reprinted with permission from Peralta DV, Heidari Z, Dash S, Tarr MA. Hybrid paclitaxel and 1030 gold nanorod-loaded human serum albumin nanoparticles for simultaneous chemotherapeutic and photothermal therapy on 4T1 breast cancer cells. *ACS Appl Mater Interfaces*. 2015;7:7101–7111.¹⁰ Copyright 2015 American Chemical Society.

et al¹²⁹ developed a one-step synthesis for a doxorubicin (DOX) loaded gold nanorod/BSA core-shell (NR@DOX:BSA), which acted as an improved macrophage-mediated delivery system, with photothermal/chemotherapeutic drug distribution and retention abilities. The nanoconstruct showed low toxicity, but after irradiation, DOX was released, causing a drastic increase in cell death. In vivo studies showed that DOX exhibited limited antitumor effects because of its rapid clearance; in contrast, the synthesized nanosystem was delivered through tumor-tropic migration to tumor tissues.

Gold nanoflowers (AuNFs) represent an interesting and novel nanoparticle shape; in a recent study by Uppal and Bose,⁷ AuNFs were successfully capped with HSA through either electrostatic or covalent attachment, resulting in a stable system that was non-toxic in a cancer cell line (human squamous carcinoma, Nt8e). This study demonstrates the promise of AuNF-BSA for biomedical applications, such as drug delivery or theranostics.

In another study, non-conventional gold nanocube-protein hybrids (PGHNs) were constructed by Ding et al,¹²³ using AuNCs, BSA and Trp as building blocks. The PGHNs demonstrated blue emission, high biocompatibility and cell internalization in *S. cerevisiae* cells. The PGHNs were loaded with rhodamine 6G and DOX separately and administered to the cells, resulting in internalization and release in the nucleus, demonstrating that they are a promising vehicle for drug delivery.

Scientific advances in nanotechnology have resulted in new nanoplatforms to treat diseases other than cancer; for example, the systems described in the current review may have applications ranging from the treatment of dermatological diseases to bacterial infections. Examples of this are included in the works of Lademann et al¹³⁰ and Rastogi et al.¹¹²

Lademann et al¹³⁰ focused on dermatotherapy, resulting in photoactivated fluorescein isothiocyanate release from an AuNP-doped BSA system. In an *ex vivo* porcine skin model, the system selectively targeted follicular ducts, because of the suitable size and biocompatibility properties of the particles. The plasmonic release of the encapsulated drug after irradiation demonstrates that this system is feasible as a plasmonic nanocarrier for the treatment of dermatological diseases.

AuNPs have also been used as antibiotic delivery systems, as demonstrated in a study by Rastogi et al,¹¹² in which highly stable albumin-capped AuNSs were

synthesized using sodium borohydride as a reducing agent. The system was functionalized for amino-glycosidic antibiotic delivery (streptomycin, neomycin, gentamicin and kanamycin), and the resulting nanoconstruct showed increased antibacterial activity against Gram-negative and Gram-positive bacterial strains, compared with free antibiotic; therefore, this system has considerable potential for infection prevention and therapeutics.

Albumin AuNPs for diagnostic devices

One of the objectives of nanomedicine is to improve the capabilities of current medicine and make it more selective and sensitive. In this context, it is important to develop diagnostic devices to identify diseased tissue in a non-invasive way.

AuNPs have emerged as a promising diagnostic system, because of their optic properties. Moreover, they can be functionalized; for example, they can be selectively delivered to a malignant site and therefore this site can be detected by the presence of the AuNPs. One of the techniques used to detect AuNPs is photoacoustic (PA) imaging, as it is a highly sensitive nonionizing imaging modality, in which a pulsed laser can penetrate deep tissue (up to 5 cm), thereby causing rapid heating and thermal expansion, which generates an acoustic pressure wave.¹⁰⁵ PA combined with computed tomography (CT) provides 3D visual images with high spatial resolution. For example, in Zu et al,¹⁰⁵ BSA-capped gold nanostars (AuNS-BSA) were synthesized, and demonstrated an average size of 85 nm, excellent biocompatibility in HepG2 cells and PA contrast enhancement, which improves CT accuracy, thereby giving rise to a dual-modal CT/PA imaging contrast agent.

In addition to PA/CT, there are other techniques to detect AuNPs and their derivatives. Chen et al¹³¹ synthesized AuNSs conjugated to indocyanine green (ICG) and stabilized using albumin (AuNS-ICG-BSA), which demonstrated good biocompatibility and low cytotoxicity, together with a superior photothermal ablation effect in U87 glioma cells. This represents a spatially controlled antitumor agent, and it caused cell damage only after light irradiation. However, this nanoconstruct is useful for surface enhanced Raman scattering (SERS) imaging-based real-time sensitive monitoring of intracellular temperature in photothermal therapy (PTT). Interestingly, this system allows for the detection of small temperature variations within a cell during PTT,

and this can be monitored through SERS, as a nanothermometer. A simpler system was demonstrated in a study by Florina et al¹³² in which highly stable and biocompatible PEG-AuNP and BSA-PEG-AuNP systems, with good internalization in A549 cells and SERS activity, were demonstrated to be promising candidates for intracellular sensing applications.

Albumin AuNP theranostic devices

In recent years, a new strategy called theranostics has become increasingly important in nanomedicine. The acronym “theranostics” derives from the words therapy and diagnosis, and a theranostic agent can detect the signs of a pathology (including tumors, infections, damaged tissue, and overexpressed receptors) and generate treatment.¹³³ This field involves the use of systems with therapeutic action for the treatment of diseases, and which in turn, allow for non-invasive diagnosis and monitoring in real time,²⁷ providing an efficient treatment, and early (non-invasive) detection of the disease, in addition to personalized therapy.¹³³

To date, a great variety of theranostic platforms have been reported, including polymers, liposomes, micelles, dendrimers,¹³⁴ organic materials,^{135,136} inorganic nanoparticles,^{137–139} and proteins,^{24,140} of which, the latter have been used as nanotransporters. Surface modifications, changes in particle size and functionalization of nanomaterials can improve biocompatibility, retention and therapeutic efficacy.

An AuNP-BSA system has shown therapeutic effectiveness in chronic diseases such as cancer, as demonstrated in a recent study by Chiu et al.¹³⁸ The study demonstrated a simple one-step fabrication of a biomimetic nanomaterial (AuNR@SA) from a synthetic source. AuNRs were synthesized and encapsulated with SA, giving rise to core-shell gold

nanorods-albumin, AuNR@SA. The SA capping was achieved through either glutaraldehyde crosslinking (GTA), or a denaturing method using methanol/ethanol (EM). GTA showed less interaction with endogenous proteins (SA, transferrin and fibrinogen), thereby leading to reduced protein corona formation, compared to the EM system. Moreover, GTA crosslinking resulted in a higher photothermal efficiency and temperature increase. The system was tested for DOX delivery, and produced in vivo tumor growth inhibition through NIR laser exposure, through a combination of photothermal therapy and chemotherapy. Moreover, the GTA system has proven to be useful for diagnosis with PA, showing a 3.5-fold higher signal than AuNR-CTAB in tramp C1 cells and in vivo mouse models (Figure 11A and B), as well as accumulation in tumor cells.

Another interesting system is that described by Sashidaran et al,¹¹⁰ where gold nanostars (AuNSs) were synthesized through a simple wet chemistry route and subsequently BSA-functionalized, yielding a stable monodispersed solution, with an average particle size of 120 nm. AuNS-BSA did not exhibit any inherent cytotoxic activity towards healthy (L929) and cancerous (KB) cells, but it showed significant photothermal ablation properties after laser exposure, thereby working as a therapeutic agent. Interestingly, the system also showed good CT imaging properties. AuNS-BSAs were loaded with rhodamine 6G, allowing the system to exhibit SERS and it is considered to be promising for Raman imaging.

AuNCs are prominent nanosystems because of their ease of functionalization and small size, which allows them to be easily cleared by the kidneys.¹⁴¹ Cui et al¹²¹ synthesized BSA stabilized AuNCs, which were subsequently cross-linked to create a well-defined gold cluster nanoassembly (AuCNA). The AuCNA showed high

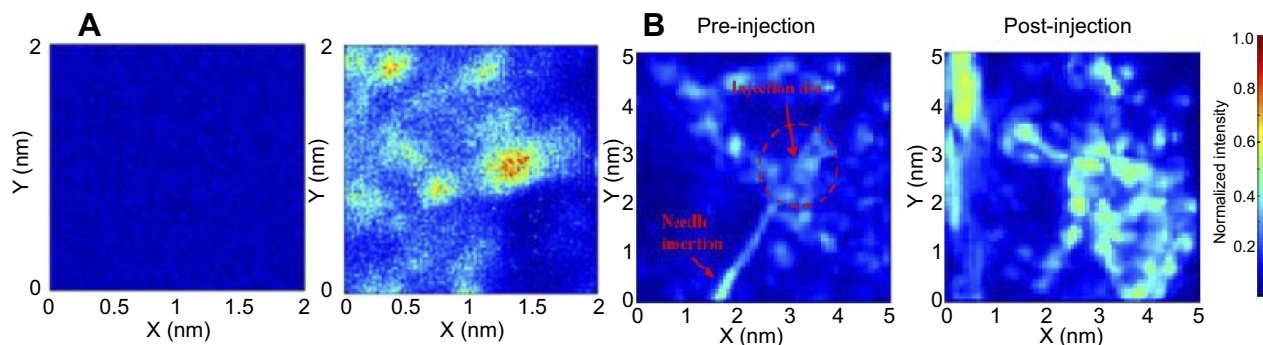


Figure 11 Capability of NR@SA nanoplatform for PA imaging. (A) 2 × 2 mm area of projected C-scan PA images of nontreated and NR@SA treated tramp C1 tumor cells. (B) in vivo photoacoustic imaging from tumor-bearing mice during and after local delivery of NR@SA. Reprinted with permission from Chiu HT, Chen CH, Li ML, et al. Bioprostheses of core-shell gold nanorod/serum albumin nanoimitation: a half-native and half-artificial nanohybrid for cancer theranostics. *Chem Mater.* 2018;30:729–747.¹³⁸ Copyright (2018) American Chemical Society.

biocompatibility and precise tumor targeting to 4T1 breast cancer cells, as seen through NIR fluorescence images. However, laser irradiation (660 nm) generated reactive oxygen species (ROS), which caused significant cell death *in vitro*, and this effect resulted in tumor death in an *in vivo* assay.

In a study by Fu et al,¹³⁷ curcumin, a hydrophobic, but promising therapeutic agent, was encapsulated in BSA functionalized gold nanoclusters (BSA-AuNCs), and showed enhanced internalization in human neuroblastoma (SH-SY5Y) cells, followed by strong fluorescence within the cells and the inhibition of proliferation via the induction of apoptosis. This research is ongoing, but highlights that the system has significant potential as a candidate for theranostic applications, by taking advantage of such fluorescence signals.

The AuNCs-BSA system has also been used for the delivery of approved anticancer drugs, such as in Ding et al,¹²⁵ where constructed BSA@AuNCs were modified using cyclic arginine-glycine-aspartate (cRGD) and loaded with DOX (DOX/RGD-BSA@AuNCs). This nanosystem produced growth inhibition in cancer cells, including Hela, MCF-7, U251 and CAL-27 cells. Remarkably, the DOX/RGD-BSA@AuNCs system proved to be useful not only for drug delivery, but also for localized *in vivo* imaging, as shown in Figure 12, where a decrease in tumor size and specific detection of the tumor site is notable after treatment using the nanosystem and laser irradiation.

Nanosystems offer a wide variety of possibilities because of their versatility and functionalization potential, and they can present enhanced and special characteristics, such as sensitivity to external stimuli (such as light, laser radiation, heat and pH). By means of pH sensitivity, Huang et al⁹⁸ used BSA as a template for the creation of an Au-BSA core/shell, which was subsequently functionalized with DOX and FA (Au-BSA-DOX-FA). The nanocomposite showed *in vitro* non-toxicity in MGC-803 and GES-1 cells, pH-sensitive drug release properties, and superior antitumor activity compared with free DOX. *In vivo* CT imaging showed that the nanocomposite had selective antitumor activity by targeting overexpressed folate receptors in gastric cancer.

AuNPs can be combined with other metallic nanoparticles, such as Gd₂O₃ nanoparticles, which can improve the properties of the system for theranostic applications. An example is reported in a study by Han et al,¹⁴² in which gadolinium oxide-gold nanocluster hybrids (Gd₂O₃-AuNCs), stabilized in BSA were synthesized, showing high biocompatibility, *in vivo* and *in vitro* uptake, as well as signal enhancement for magnetic resonance and X-ray CT imaging. This system, besides enhancing imaging, can be used to kill tumor cells through the production of singlet oxygen species (¹O₂) after NIR irradiation (808 nm). However, this nanosystem also proved to be useful for loading and delivering the fluorescent dye ICG, improving photodynamic properties. Similarly, You et al¹⁴³ designed a multifunctional delivery platform for Gd

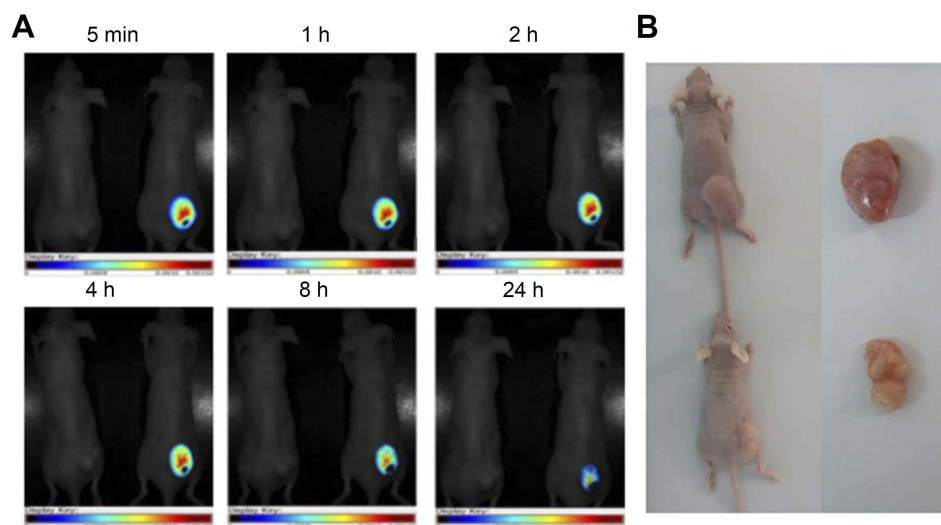


Figure 12 (A) *In vivo* targeted cancer fluorescence images of Hela tumor-bearing mice exposed to the laser (488 nm, 425 mW cm², 25 min) after injection of the DOX/RGD-BSA@AuNCs system. **(B)** The images of tumor excised from mice after injection of PBS solution or DOX/RGD-BSA@AuNCs solutions at 19 day. Reprinted with permission from Ding C, Xu Y, Zhao Y, Zhong H, Luo X. Fabrication of BSA@AuNC-based nanostructures for cell fluorescence imaging 1435 and target drug delivery. *ACS Appl Mater Interfaces*. 2018;10:8947-8954.¹²⁵ Copyright (2018) American Chemical Society.

based on hollow gold nanoshells coated with BSA (Au@BSA-Gd), in which gold conferred photothermal properties, and the construct exhibited properties that are useful for CT and PA. The incorporation of ICG into the system provided photodynamic/photothermal properties and near-infrared fluorescence (NIRF)/PA imaging capabilities, thereby achieving an ideal theranostic agent for NIRF/PA/CT/MR quadmodal imaging.

Other non-conventional AuNP geometries also offer interesting properties, which in addition to SA enhanced biocompatibility, can be useful for theranostic applications. An example is branched gold nanoshells (BAuNSHs), as synthesized by Topete et al¹⁴⁴ through a seeded-growth, and which were loaded with DOX. The system was also coated with a BSA-ICG-FA complex. This is a multifunctional system useful for fluorescence imaging because of the inherent fluorescence of the internalized dye and the presence of FA. The system also has the potential to deliver chemotherapy through its loaded drug, and photothermal and photodynamic therapy upon NIR irradiation. The synergistic effect of the nanoconstruct was observed in terms of an increase in internalization and toxicity in HeLa cells, and bioaccumulation at tumor sites in a tumor-bearing mouse model, observed through fluorescence imaging.

Conclusion

Albumin is a promising biomaterial for new approaches, and its biocompatibility and transport properties allow for the creation of new materials and the improvement of existing materials. The special characteristics of malignant tissues, such as an accelerated rate of proliferation and increased cellular stress and cell growth, resulting in a higher demand for energy and nutrients, can be exploited for controlled drug delivery.

The synergy between SA and AuNPs results in interesting properties, such as an increase in stability, a reduction in the interaction with plasma proteins, passive and active targeting of malignant cells, and an increase in selectivity. The photothermal properties and the NIR absorption of light of AuNPs are suitable for local hyperthermia, photothermal ablation and photothermal release. The system demonstrates significant potential in terms of spatial and temporally controlled release, in addition to its applications in theranostics, using simple detection techniques such as PA, CT or a combination of both, as previously described. Coating AuNPs with SA decreases the interaction of the nanosystem with endogenous proteins in the body, protecting the cargo against degradation.

Future perspectives

Although, a limited number of clinical trials are now in progress, with promising results^{145–147} the use of AuNPs for clinical applications has been delayed due to the inherent potential toxicity of metal nanoparticles. As mentioned previously, the capping of AuNPs with SA allows for an improvement in their biological properties, which may encourage the scientific community to investigate AuNP-SA-based nanosystems for clinical trials, as well as other types of SA-coated NIR-plasmonic nanoparticles for biomedical applications.

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Disclosure

The authors report no conflicts of interest in this work.

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