

Promotion of in vivo degradability, vascularization and osteogenesis of calcium sulfate-based bone cements containing nanoporous lithium doping magnesium silicate

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Abstract: Nanoporous lithium doping magnesium silicate (nl-MS) was introduced into calcium sulfate hemihydrate to prepare calcium sulfate composite (nl-MSC) bone cements. The introduction of nl-MS improved the in vitro degradability of nl-MSC cements, which could neutralize acidic degradable products of calcium sulfate and prevented the pH from dropping. The cements were implanted into the bone defects of femur bone of rabbits, and the results of histological and immunohistochemical analysis revealed that massive new bone tissue formed in the defects while the cements were degradable, indicating that the osteogenesis and degradability of the nl-MSC cements were much better than the control calcium sulfate dihydrate (CSD) cements. Furthermore, the positive expression of vascular endothelial growth factor and collagen type I for nl-MSC cements was higher than CSD, indicating that addition of nl-MS into the cements enhanced vascularization and osteogenic differentiation. The results suggested that the nl-MSC cements with good biocompatibility and degradability could promote vascularization and osteogenesis, and had great potential to treat bone defects.

Keywords: bone cements, nanoporous lithium doping magnesium silicate, degradability, vascularization, osteogenesis

Introduction

Calcium sulfate-based bone cement has a long history as the bone-repair biomaterial for clinical application due to its good degradability and biocompatibility.¹ Calcium sulfate hemihydrate (CSH) can react with water immediately and transform into calcium sulfate dihydrate (CSD).² However, CSD cement has some drawbacks that significantly limit its clinical application, such as short setting time, low bioactivity and poor ability to form new bone tissue.^{3,4} Additionally, the degradation of CSD in vivo causes local acidity, which can lead to inflammatory reaction in host tissues.⁵

Nanoporous silicates (n-SCs) as new kinds of biomaterials have high specific surface area and pore volume, which may greatly accelerate the kinetics of apatite deposition, thus enhancing the bone-forming bioactivity.⁶ Furthermore, the degradable products of n-SCs present weak alkalinity, which is favorable for cell growth and tissue regeneration.^{6,7} Silicate-based biomaterials have been reported to enhance vascularization of osteoblastic cells through stimulating the expression of vascular endothelial growth factor (VEGF) from osteoblastic cells or neighbor cells, and it has been widely accepted that angiogenesis could enhance bone formation.⁸⁻¹⁰ A previous

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study has shown that silicate biomaterials are able to stimulate osteogenic differentiation of osteoblastic cells, and subsequently, strongly affect new bone formation.^{9,11}

Lithium (Li) is an important trace element in an organism, and previous studies have shown that Li ions stimulated the proliferation and osteogenic differentiation of bone marrow mesenchymal stem cells (BMSCs) through activation of the Wnt signaling pathway *in vitro*, and oral administration of Li ions enhanced subchondral bone formation via activation of the Wnt signaling pathway.^{12,13} Moreover, Li-containing mesoporous bioglass scaffolds promoted renaissance of subchondral bone and enhanced hyaline cartilage tissue restoration in rabbit models of osteochondral defects.¹⁴ In this study, nanoporous lithium doping magnesium silicate (nl-MS) was introduced into CSH to prepare calcium sulfate-based composite (nl-MSC) cements, and the *in vivo* degradability, vascularization and osteogenesis of nl-MSC cements were investigated.

Materials and methods

Preparation of nl-MS and nl-MSC cements

The nl-MS was synthesized using a sol-gel method. Briefly, tetraethyl orthosilicate (3 mL) was dissolved into ethanol (25 mL) to form solution A. The cetyltrimethylammonium-chloride (CTAC, 2 g) and ammonium hydroxide (28 wt%, 4.5 mL) were dissolved into ethanol (50 mL) to form solution B. Lithium chloride (0.25 g) was dissolved into ethanol (10 mL) to form solution C. Then, solutions A and B were mixed together under rapid agitation, and solution C (1 mL) was doped into mixed solution drop by drop. Then, the mixed solution was stirred vigorously for another 3 h, and the products collected were centrifuged, and then washed 3 times with water and ethanol, respectively. The final products of nl-MS were dried for 24 h at room temperature, and then calcinated at 600°C for 4 h in air to remove CTAC. The nl-MS was characterized by high-resolution transmission electron microscopy (JEM-2010; JEOL, Tokyo, Japan).

CSH powders were prepared by heating CSD ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$) in a muffle furnace (SX2-2.5-12N; Yiheng, Shanghai, People's Republic of China) at 150°C for 12 h and then at 120°C for 6 h. The CSD, 10 nl-MSC and 20 nl-MSC cement powders were prepared by mixing CSH powders with 0, 10 and 20 wt% nl-MS powders. The CSD, 10 nl-MSC and 20 nl-MSC cement pastes were prepared by mixing cement powders with deionized water (a ratio of cement powder to liquid of 1:0.8 g/mL). The as-prepared cement pastes were cast into Teflon molds under a pressure of 2 MPa for 1 min. The samples ($\Phi 10 \times 2$ mm and $\Phi 6 \times 6$ mm) were taken out

and then cured under the condition of 100% air humidity and 37°C for 2 days. After the hardening of cements, the composition and microstructure of the CSD, 10 nl-MSC and 20 nl-MSC cements were characterized by X-ray diffractometry (XRD; Geigerflex; Rigaku Co. Ltd., Tokyo, Japan), Fourier transform infrared spectroscopy (Thermo Nicolet 6700; Thermo Fisher Scientific, Waltham, MA, USA) and scanning electron microscopy (SEM; S-3400N; Hitachi, Tokyo, Japan), respectively.

Degradation of cements in phosphate-buffered saline

Samples ($\Phi 10 \times 2$ mm) of CSD, 10 nl-MSC and 20 nl-MSC cements were immersed in phosphate-buffered saline (PBS, pH 7.4) at 37°C with a solid/liquid ratio of 0.1 g/20 mL, and the solution was refreshed every 7 days. At different time points, the specimens were removed from the liquid, rinsed with deionized water and dried in an oven at 90°C for 8 h. The percentage of weight loss was calculated as the following formula: $\text{weight loss (\%)} = (W_0 - W_t)/W_0 \times 100$, where W_0 is the starting dry weight and W_t is the dry weight at time t . The change of pH values of the solutions after the cements were soaked for different times was also determined with a pH meter (PHS-2C; JingkeLeici, Shanghai, People's Republic of China).

Implantation of cements into bone defects of rabbits

The *in vivo* experiments were performed according to the NIH guidelines for the care and use of laboratory animals (NIH Publication No 85-23 Rev 1985) and were approved by the ethical committee of the Shanghai University of Traditional Chinese Medicine. The 24 New Zealand White rabbits (female, about 2.1 kg) were divided into 3 groups. The rabbits were anesthetized with pentobarbital sodium. The incision sites were made on the right femur of rabbits, and then $\Phi 6 \times 6$ mm defects were made in the thighbone of the rabbits. The CSD, 10 nl-MSC and 20 nl-MSC cements were implanted into the defects. At 4, 8 and 12 weeks after surgery, the rabbits were sacrificed, and the thighbones were taken out and placed into 4% neutral-buffered formalin. All the rabbits were raised, operated and handled in the Research Center for Laboratory Animals of the Shanghai University of Traditional Chinese Medicine.

SRmCT images analysis

The synchrotron radiation micro-computed tomography (SRmCT) measurement was performed at beamline BL13W of the Shanghai Synchrotron Radiation Facility (Shanghai,

People's Republic of China) using a monochromatic beam with an energy of 30 keV and a distance (sample-to-detector) of 1.6 m. The pixel size was set to $14 \times 14 \mu\text{m}$, and a spacing of $14 \mu\text{m}$ was used to obtain a continuous flat image with a resolution of $1,024 \times 1,024$. The images were reconstructed by supporting software (VGStudio MAX 2.1) to generate 3-dimensional (3D) grayscale images.

Histological evaluation

The specimens were decalcified in 5% formic acid (Sigma-Aldrich, St Louis, MO, USA) at 4°C for 15 days, and then dehydrated in a gradient ethanol series and embedded in paraffin. After that, the specimens were cut into $20\text{-}\mu\text{m}$ -thick sections with a sawing microtome (Leica 1600; Leica Microsystems, Wetzlar, Germany). All sections were stained with hematoxylin/eosin (H&E). The newly formed bone was observed under a light microscope (TE2000U; Nikon, Tokyo, Japan). The ratio of newly formed bone and residual material was calculated by a semiautomatic image analysis system (Image Pro Plus 6.0).

Immunohistochemical evaluation

The immunohistochemical labeling of the decalcified sections was carried out using the immunoperoxidase detection method with the primary antibodies anti-VEGF and anti-COL-I (1:50 dilution; Abcam, Inc., Cambridge, UK) for 12 h at 4°C . Then, the sections were washed 3 times with PBS. After that, the anti-IgG biotinylated antibody (Abcam, Inc.) at a 1:200 concentration was incubated as secondary antibody for 30 min at 37°C . Thereafter, the sections were incubated with horseradish peroxidase-conjugated avidin-biotin complex (Abcam, Inc.) for another 15 min, followed by buffered 3,3'-diaminobenzidinetetrahydrochloride (Abcam, Inc.)

as chromogen. The slides were then observed under a light microscope (TE2000U; Nikon). The positive expression ratio was calculated by a semiautomatic image analysis system (Image Pro Plus 6.0).

Statistical analysis

All quantitative data expressed as the mean \pm standard deviation were analyzed with ORIGIN v 8.0 (Originlab Corporation, Northampton, MA, USA). Statistical analysis was performed with one-way analysis of variance. A value of $P < 0.05$ was considered statistically significant.

Results

TEM and XRD analysis

Figure 1 reveals the TEM image of nl-MS, and XRD of CSD, 10 nl-MSC and 20 nl-MSC cements. It can be seen that the nl-MS exhibited sphere-like particles with a size of about 100 nm. In addition, it can be seen from XRD that the diffraction peak of CSD appeared at $2\theta = 11.7^\circ, 20.8^\circ, 23.4^\circ, 29.2^\circ, 31.2^\circ$ and 33.4° , indicating that the CSH had transformed into CSD. Correspondingly, the peak of CSD was found in 10 nl-MSC and 20 nl-MSC cements, but the peak intensity of CSD significantly decreased with the increase of nl-MS content.

Infrared analysis

Figure 2 shows the infrared spectrum (IR) of nl-MS, CSD, 10 nl-MSC and 20 nl-MSC. The peaks at $3,579, 3,493$ and $1,658 \text{ cm}^{-1}$ were typical of H_2O or OH^- , while the absorption band at 802 cm^{-1} was attributed to the $\text{Si}=\text{O}$ stretching vibration, and the peak at $1,176 \text{ cm}^{-1}$ was attributed to SO_4^{2-} stretching vibration. All the characteristic peaks of CSD can be found in the 10 nl-MSC and 20 nl-MSC cements, and no obvious change of the characteristic peaks of CSD was found.

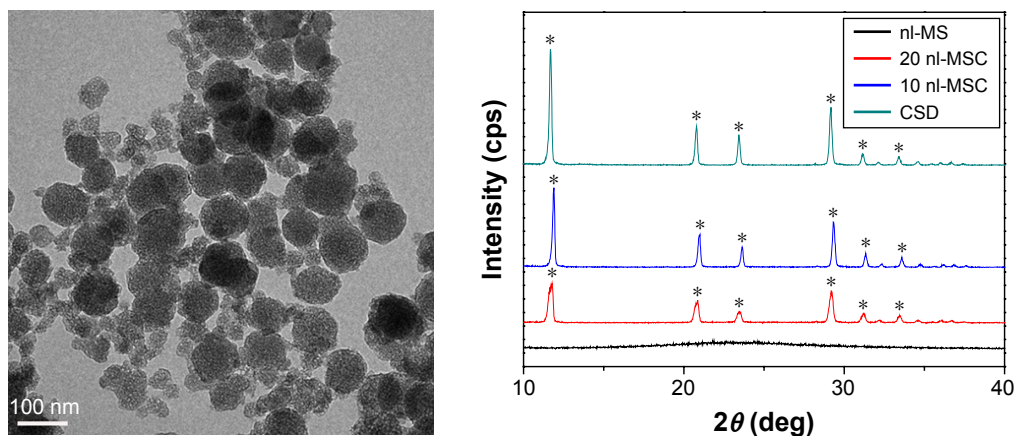


Figure 1 TEM image of nl-MS, and XRD of CSD, 10 nl-MSC and 20 nl-MSC cements. Asterisks represent the peaks of CSD.

Abbreviations: TEM, transmission electron microscopy; nl-MS, nanoporous lithium doping magnesium silicate; XRD, X-ray diffractometry; CSD, calcium sulfate dihydrate; nl-MSC, nl-MS composite.

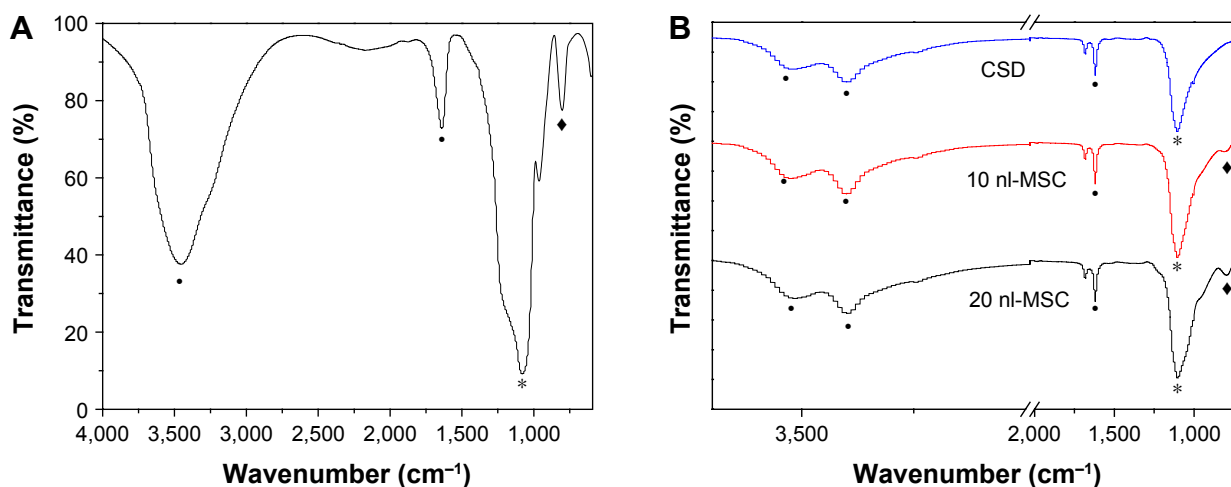


Figure 2 IR of nl-MS (A), CSD, 10 nl-MSC and 20 nl-MSC (B) cements. Filled circles represent H₂O or OH⁻, filled diamonds represent Si=O and asterisks represent SO₄²⁻.

Note: (B) FTIR of CSD, 10 nl-MSC and 20 nl-MSC.

Abbreviations: IR, infrared; nl-MS, nanoporous lithium doping magnesium silicate; CSD, calcium sulfate dihydrate; nl-MSC, nl-MS composite; IR, infrared spectrum.

SEM analysis of cements

The SEM images of surface morphology of the cements are shown in Figure 3. The low-magnification images show that the surface of the CSD cement was compact (Figure 3A), indicating that the CSH (CaSO₄·1/2H₂O) had transformed into CSD (CaSO₄·2H₂O). However, the surfaces of the cements were loose (Figure 3C and E), and some nl-MSs were observed on the surfaces of 10 nl-MSC and 20 nl-MSC cements. At high magnification, the rod-like microcrystalline CSD was seen (Figure 3B), while microporous CSD appeared on the surfaces of the 10 nl-MSC and 20 nl-MSC cements, showing that the amorphous nl-MS dispersed in the microcrystalline CSD (Figure 3D and F).

Degradation of cements in PBS

Figure 4A shows the weight loss of the CSD, 10 nl-MSC and 20 nl-MSC cements after being soaked into PBS for different times. The weight loss of the cements increased with time. Moreover, the weight loss of the cements increased with the increase of nl-MS content. At 12 weeks, the weight loss ratio of the CSD, 10 nl-MSC and 20 nl-MSC cements was 55.4, 64.4 and 72.8 wt%, respectively. Figure 4B shows the pH changes of the solutions after immersion of specimens in PBS for different periods of time. The pH of the solution for CSD significantly decreased at the first 2 weeks (from 7.40 to 7.15); after that, it began to slowly decline and dropped to 6.86 at 12 weeks. However, the pH of the solution for 10 nl-MSC and 20 nl-MSC cements dropped from 7.40 to 7.18 and 7.27 at the first 2 weeks; after that, the pH value began to increase, and reached 7.38 and 7.45 at 12 weeks, respectively.

SRmCT image analysis

Figure 5 shows the cross-section of the 3D reconstruction images of SRmCT after CSD, 10 nl-MSC and 20 nl-MSC cements were implanted into bone defects of rabbit femur for different periods of time. From these images, it can be seen that the bone defects were gradually repaired with time.

Histological evaluation

Figure 6 shows the histological sections of H&E staining of CSD, 10 nl-MSC and 20 nl-MSC cement after being implanted in vivo for 4, 8 and 12 weeks. The amounts of new bone tissue for all the cements increased with time. In addition, the amounts of new bone tissue increased with the increase of nl-MS content (CSD < 10 nl-MSC < 20 nl-MSC). Meanwhile, it was found that the cements in the bone defects gradually reduced with time, indicating the gradual degradation of the cements.

Figure 7A shows the quantitative analysis of new bone formation after the cements were implanted in vivo for different times. It was found that the percentage of new bone formation increased with time for all the cements, and the percentage of new bone formation increased with the increase of nl-MS content (CSD < 10 nl-MSC < 20 nl-MSC). At 12 weeks, the percentages of new bone formation for CSD, 10 nl-MSC and 20 nl-MSC were 47.91%, 56.52% and 67.50%, respectively.

Figure 7B shows the quantitative analysis of residual material ratio after the cements were implanted in vivo for different times. The residual material ratio decreased with time for all the cements, and the residual material ratio decreased with the increase of nl-MS content (CSD > 10 nl-MSC > 20 nl-MSC).

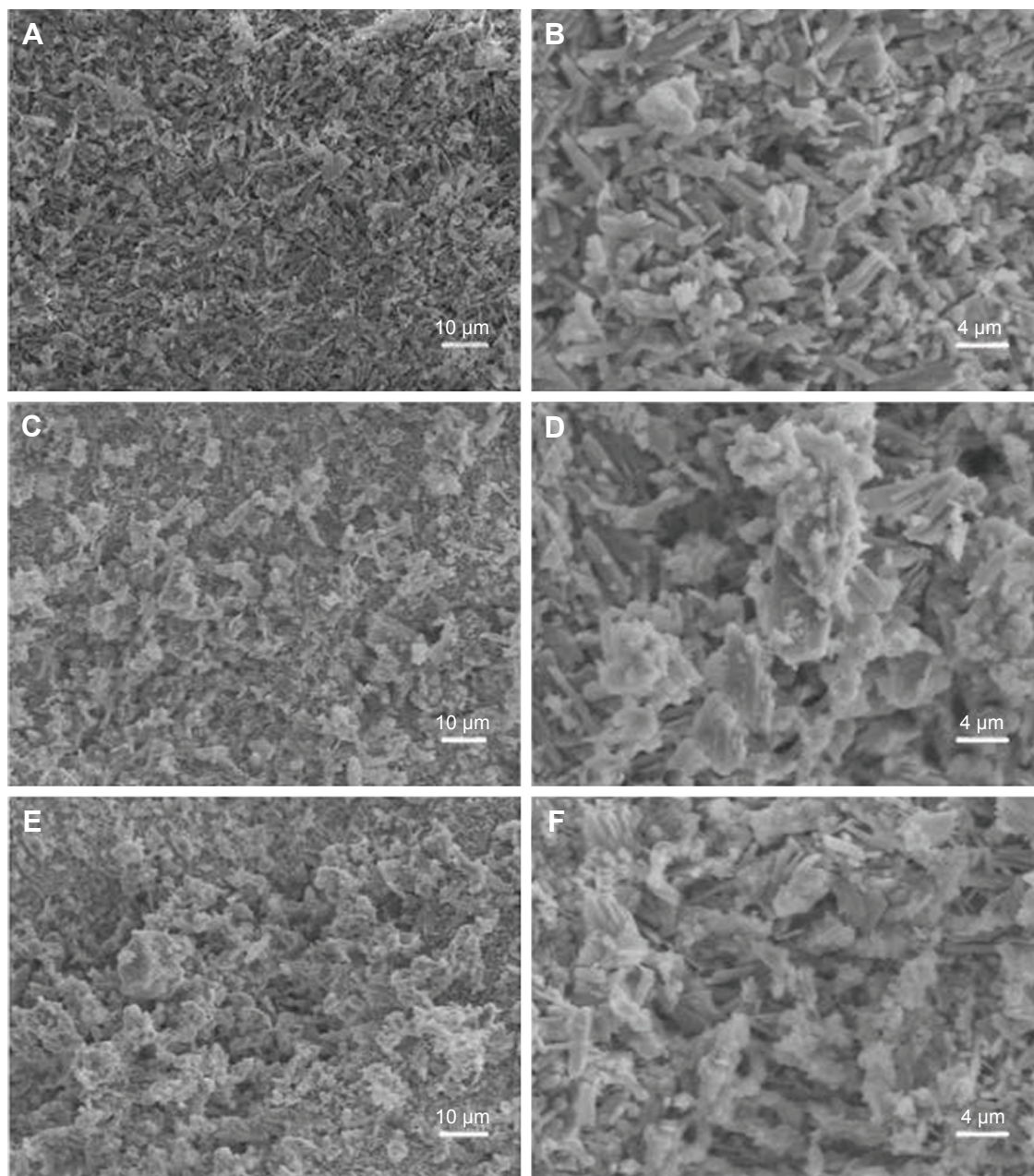


Figure 3 SEM images of surface morphology of CSD (**A** and **B**), 10 nl-MS (**C** and **D**) and 20 nl-MS (**E** and **F**) cements.

Abbreviations: SEM, scanning electron microscopy; CSD, calcium sulfate dihydrate; nl-MS, nanoporous lithium doping magnesium silicate composite.

At 12 weeks, the percentages of residual materials for CSD, 10 nl-MS and 20 nl-MS were 12.20%, 9.43% and 6.34%, respectively.

Immunohistochemical analysis

Figure 8 shows the immunohistological staining of VEGF after the cements were implanted *in vivo* for different times. It was found that the VEGF staining intensity for all the cements increased with time, and the VEGF staining intensity increased with the increase of nl-MS content. Figure 9 shows the quantitative analysis of the immunohistological positive

expression ratio of VEGF after the cements were implanted *in vivo* for different times. It was found that the positive expression ratio for all the cements increased with time and nl-MS content. At 12 weeks, the positive expression ratios of VEGF for CSD, 10 nl-MS and 20 nl-MS were 26.97%, 33.56% and 38.79%, respectively.

Figure 10 shows the immunohistological staining of collagen type I (COL-I) by histological sections after the cements were implanted *in vivo* for different times. It was found that the COL-I staining intensity for all the cements increased with time, and the COL-I staining intensity increased with the

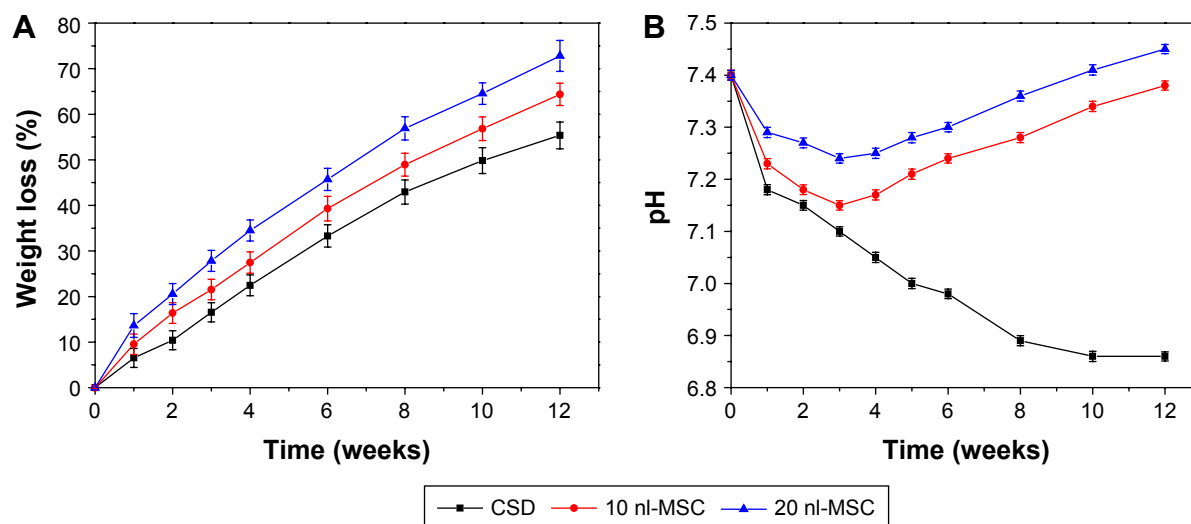


Figure 4 Weight loss (A) of the cements in the PBS and change of pH (B) of solution after the cements were soaked for different times.
Abbreviations: PBS, phosphate-buffered saline; CSD, calcium sulfate dihydrate; nl-MSC, nanoporous lithium doping magnesium silicate composite.

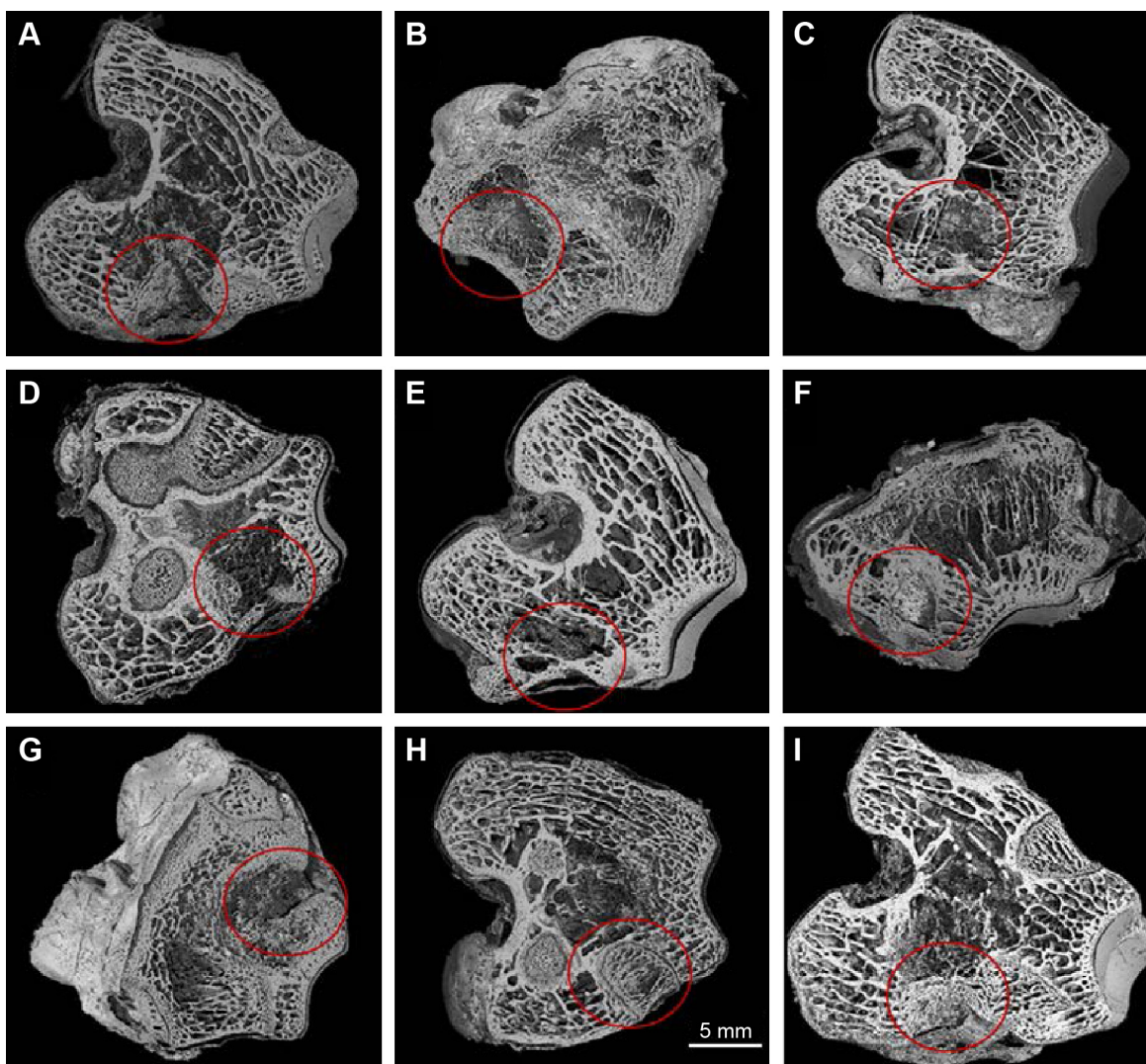


Figure 5 Three-dimensional reconstruction images (cross-section) of SRmCT after CSD (A–C), 10 nl-MSC (D–F) and 20 nl-MSC (G–I) were implanted into bone defects of rabbit femur for 4 (A, D and G), 8 (B, E and H) and 12 (C, F and I) weeks. Bone defects are marked with red circles.

Abbreviations: SRmCT, synchrotron radiation micro-computed tomography; CSD, calcium sulfate dihydrate; nl-MSC, nanoporous lithium doping magnesium silicate composite.

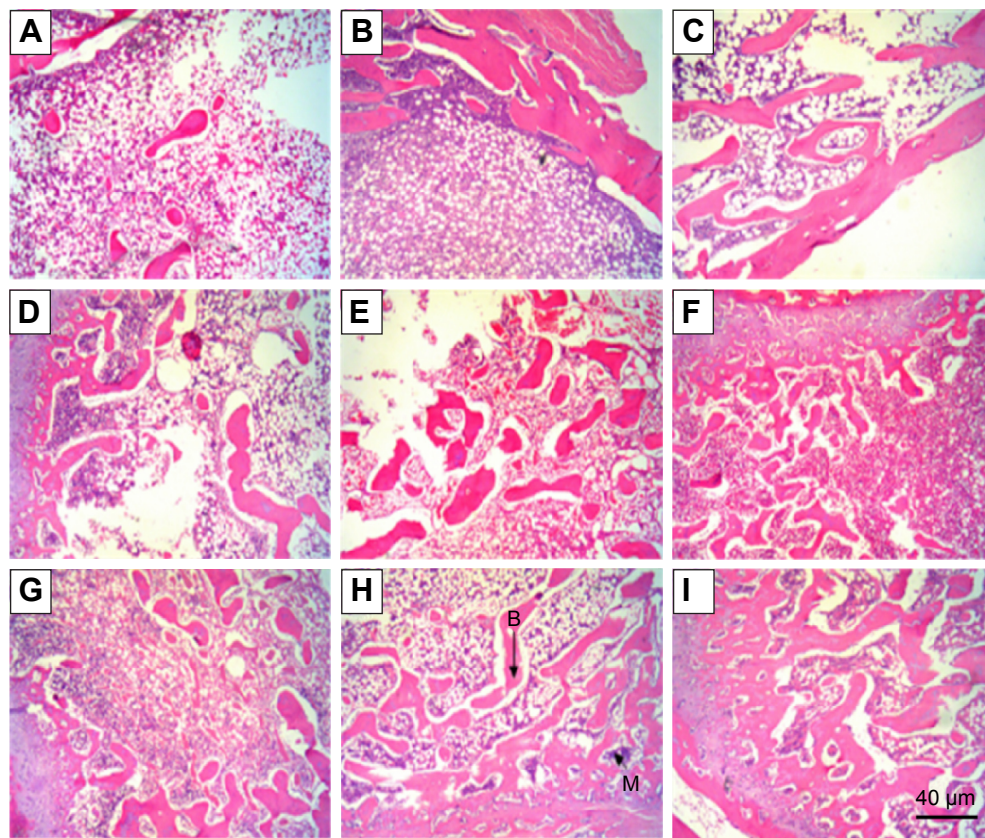


Figure 6 Histological sections of H&E staining after CSH (A–C; 40×), 10 nl-MSC (D–F; 40×) and 20 nl-MSC (G–I; 40×) were implanted in vivo for 4 (A, D and G), 8 (B, E and H) and 12 (C, F and I) weeks (B: new bone and M: materials).

Abbreviations: H&E, hematoxylin/eosin; CSH, calcium sulfate hemihydrate; nl-MSC, nanoporous lithium doping magnesium silicate composite.

increase of nl-MS content. Figure 11 shows the quantitative analysis of the immunohistological positive expression ratio of COL-I after the cements were implanted in vivo for different times. It was found that the positive expression ratio

for all the cements increased with time and nl-MS content. At 12 weeks, the positive expression ratios of COL-I for CSD, 10 nl-MSC and 20 nl-MSC cements were 22.70%, 30.34% and 48.30%, respectively.

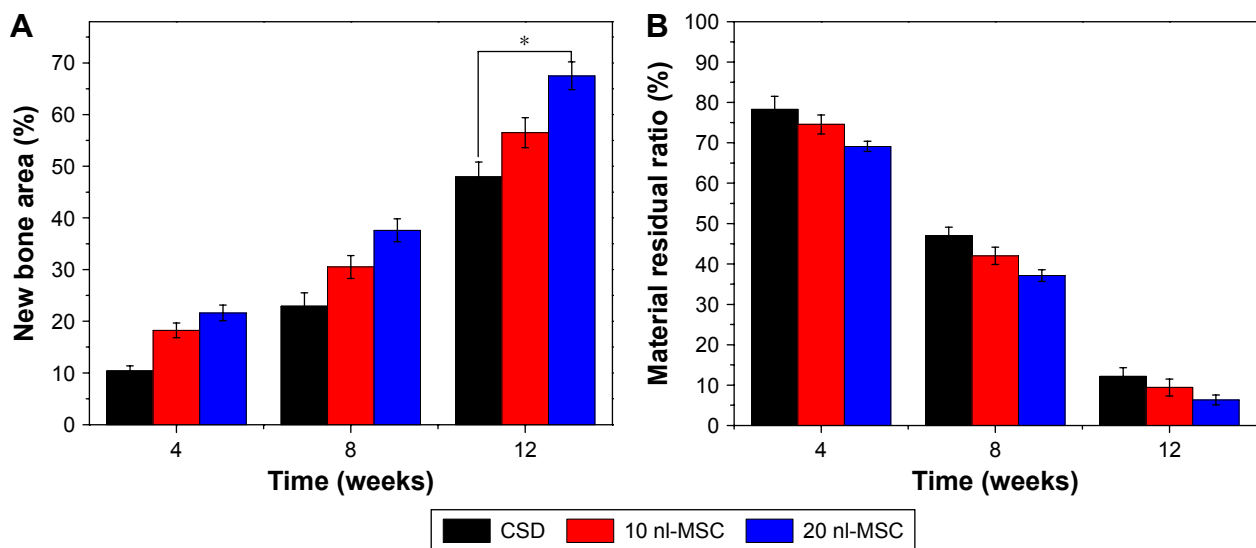


Figure 7 Quantitative comparison of the percentage of new bone formation (A) and material residual ratio (B) by histological section after the implantation of cements in vivo for 4, 8 and 12 weeks. Asterisk represents statistical significance ($P < 0.05$).

Abbreviations: CSD, calcium sulfate dihydrate; nl-MSC, nanoporous lithium doping magnesium silicate composite.

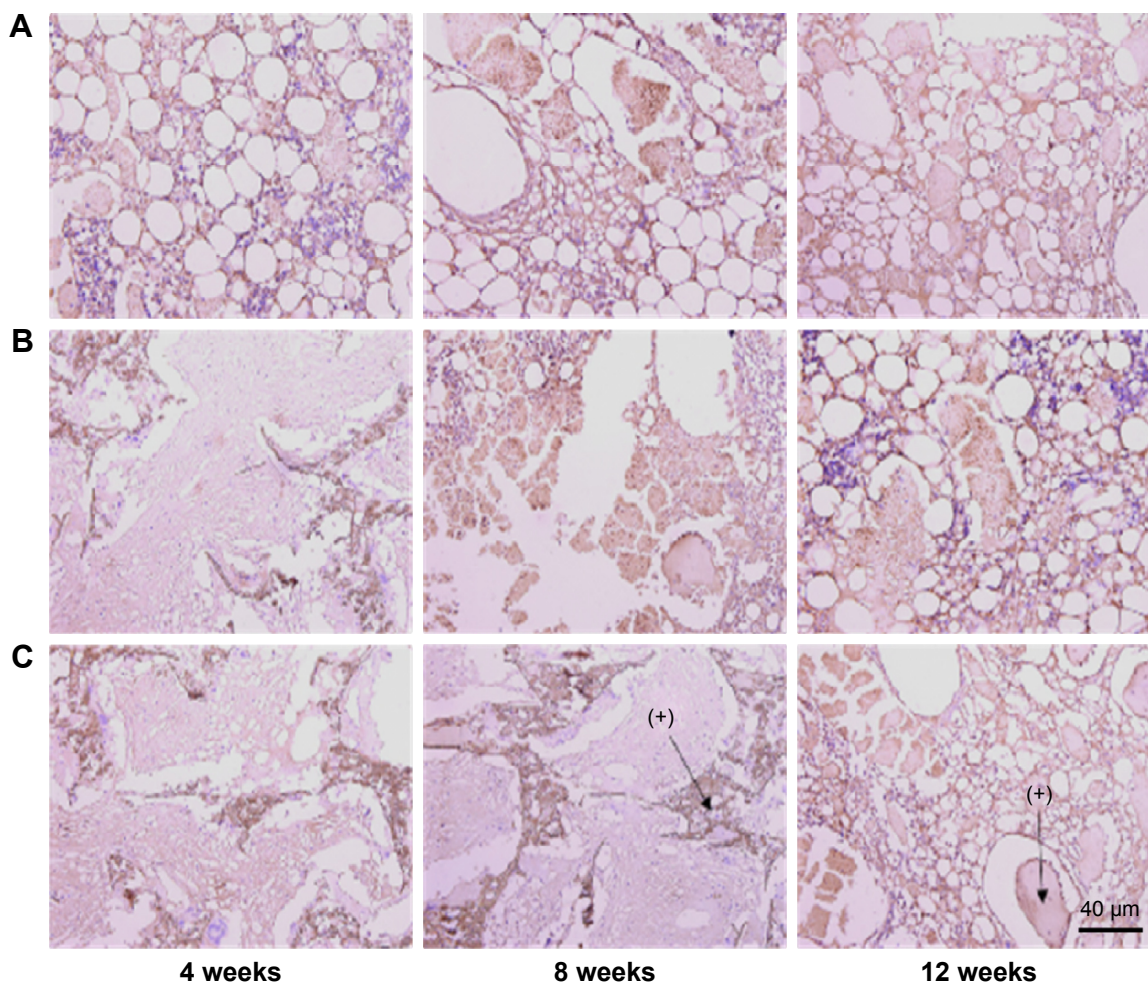


Figure 8 Immunohistological staining of VEGF by histological sections after CSD (A), 10 nl-MSC (B) and 20 nl-MSC (C) cements were implanted in vivo for 4, 8 and 12 weeks. Purple color and plus sign indicate positive expression of VEGF.

Abbreviations: VEGF, vascular endothelial growth factor; CSD, calcium sulfate dihydrate; nl-MSC, nanoporous lithium doping magnesium silicate composite.

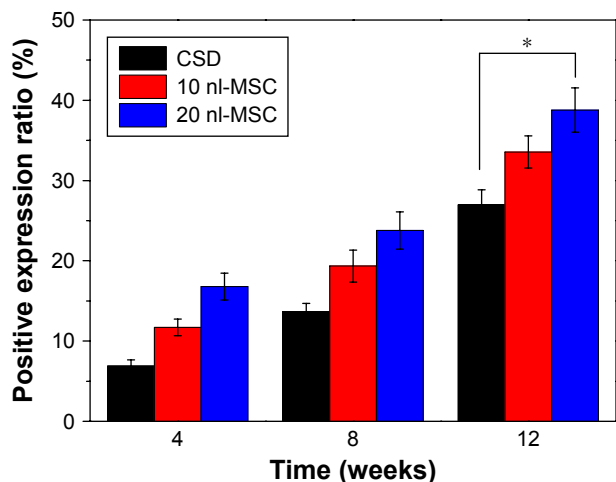


Figure 9 Quantitative analysis of the immunohistological positive expression of VEGF by histological sections after the implantation of cements in vivo for 4, 8 and 12 weeks. Asterisk represents statistical significance ($P < 0.05$).

Abbreviations: VEGF, vascular endothelial growth factor; CSD, calcium sulfate dihydrate; nl-MSC, nanoporous lithium doping magnesium silicate composite.

Discussion

Calcium sulfate as a bone implant biomaterial has been clinically used for more than a century. However, many studies have clearly revealed that the shortage of calcium sulfate was not osteoinductive or osteogenic.¹⁵⁻¹⁷ In this study, nl-MS was synthesized, and the nl-MS-containing calcium sulfate bone cements were prepared. The result indicated that the addition of nl-MS increased the degradability of the nl-MSC cements, which depended on the nl-MS content. The CSD cement was a compact structure after the CSH transformed into CSD, while the 10 nl-MSC and 20 nl-MSC cements were loose microporous structures after addition of nl-MS into CSD. The nl-MS with nanoporosity distributed evenly into the cements, which undermined the compact structure of CSD cement. Moreover, the unique nanostructure properties of nl-MS, including high specific surface area and large pore volume,

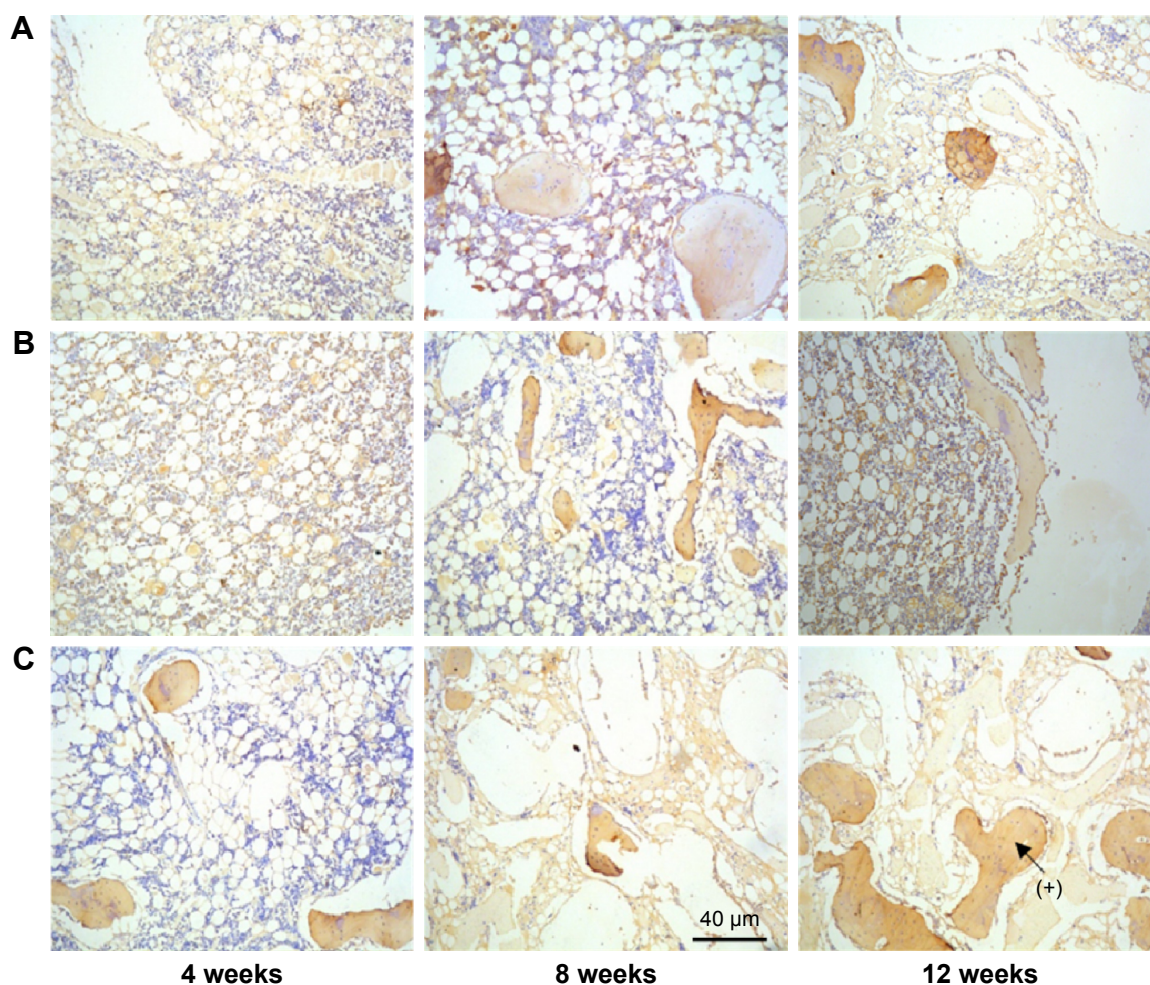


Figure 10 Immunohistological staining of COL-I by histological sections after CSD (A), 10 nl-MSC (B) and 20 nl-MSC (C) were implanted in vivo for 4, 8 and 12 weeks. Brown color and plus sign indicate positive expression of COL-I.

Abbreviations: COL-I, collagen type I; CSD, calcium sulfate dihydrate; nl-MSC, nanoporous lithium doping magnesium silicate composite.

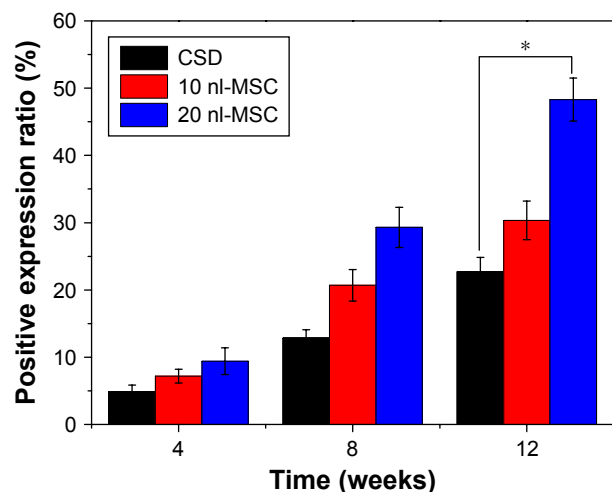


Figure 11 Quantitative analysis of the immunohistological positive expression of COL-I by histological sections after the implantation of cements in vivo for 4, 8 and 12 weeks. Asterisk represents statistical significance ($P < 0.05$).

Abbreviations: COL-I, collagen type I; CSD, calcium sulfate dihydrate; nl-MSC, nanoporous lithium doping magnesium silicate composite.

increased the porosity of the cements. Therefore, incorporation of nl-MS into CSD formed a microporous structure in the 10 nl-MSC and 20 nl-MSC cements, which increased the contact area with water when soaked into liquid, and thus improved the degradability of the nl-MSC cements.

Some studies have shown that the degradation of CSD produced acidic degradable products and formed a local acidic microenvironment, which caused inflammatory reaction in vivo.^{18,19} In this study, the pH of the solution decreased from 7.40 to 6.86 after the CSD cement immersion into PBS for 12 weeks, indicating that some acidic degradable products emerged during soaking. However, the pH of the solution for 20 nl-MSC slightly increased from 7.40 to 7.45, indicating that some basic products emerged after the degradation of nl-MS, which might neutralize the acidic degradable products of CSD. According to previous studies, a slightly alkalinized microenvironment might be favorable for cell

growth and tissue formation.^{20,21} Therefore, the pH of 7.45 for 20 nL-MS might promote cells/bone tissue ingrowth and new bone regeneration.

In order to investigate the *in vivo* degradability, vascularization and osteogenesis of the bone cements, the specimens were implanted into femur defects of rabbits. The results revealed that all the cements exhibited excellent biocompatibility, and neither significant foreign body response nor chronic inflammation to the cements was observed throughout the *in vivo* study. The results of the cross-section of the 3D reconstruction images of SRmCT revealed that the bone defects were gradually repaired with time after the cements were implanted into bone defects of rabbit femur for different times. The effects of bone defects repair increased with the time and nL-MS content, in which 20 nL-MS was the best for defects repair at 12 weeks, indicating that the osteogenesis of 20 nL-MS was the best. It can be suggested that the fast degradation of 20 nL-MS and release of Si and Li ions promoted osteogenesis and repaired the bone defects. These results indicated that the addition of nL-MS into the cements promoted the bone defects repair. The results of the histological sections of H&E staining showed that the amounts of new bone tissue increased not only with time but also with the increase of nL-MS content in the cements. The results indicated that the addition of nL-MS improved the new bone formation of nL-MS cements. Moreover, the residual material ratio of the cements decreased with time, and with the increase of nL-MS content, indicating that addition of nL-MS improved the degradability of nL-MS cements and new bone formation. The nL-MS cements were degradable at skeletal sites over 12 weeks. Therefore, the degradation rate of nL-MS cements can meet the requirement of bone defects repair. The first degradation of nL-MS in the nL-MS cements at early stage led to microporous structure in the specimens, which was favorable for the new bone tissue formation and ingrowth. The degradation rate of nL-MS increased slightly with the increase of nL-MS content *in vivo*, which was consistent with the *in vitro* degradation. It can be suggested that the nL-MS cements not only had good biocompatibility and osteogenesis, but were also gradually degradable over a period of time and progressively replaced by the natural host bone tissue.

Insufficient vascularization would restrain new bone tissue formation and delay bone defects healing, possibly because of deviations from the principles of vasculature functioning in osteogenesis.^{22,23} Moreover, the vascular development always precedes the osteogenesis, and bone tissue ingrowth greatly depends on the vascularization. Under the circumstance of

a well-developed vascular network, the osteoblasts produce osteoid, calcify and differentiate to osteocytes and exhibit healthy bone formation.²⁴ VEGF has also been reported to stimulate the osteogenic differentiation of osteoblasts.²⁵ In this study, the results from immunohistochemical analysis showed that the nL-MS cements containing nL-MS significantly promoted the positive expression of VEGF, which was nL-MS content-dependent (CSD <10 nL-MS <20 nL-MS). Clearly, the nL-MS obviously enhanced the expression of VEGF, which stimulated the vascularization.

COL-I is a major component of extracellular matrix deposition, which accounts for 90% of the bone matrix proteins.²⁶ COL-I is secreted at the early phase of osteoid matrix deposition, so it is an early marker of osteogenic differentiation associated with the formation of extracellular matrix.^{27,28} In this study, the results from immunohistochemical analysis showed that the cements containing nL-MS significantly promoted the positive expression ratio of COL-I, which was nL-MS content-dependent (CSD <10 nL-MS <20 nL-MS), indicating that nL-MS cements containing nL-MS promoted excellent *in vivo* osteogenesis for bone defects repair.

Previous studies revealed that silicate-based biomaterials with simple composition of Si stimulated osteogenic differentiation of BMSCs and angiogenesis of human umbilical vein endothelial cells.²⁹ In the present study, VEGF and COL-I expressions for both 20 nL-MS and 10 nL-MS cements were significantly upregulated by addition of nL-MS, and it can be reasoned that the nL-MS played important roles in the upregulation of vascularization and osteogenic differentiation. Therefore, this study demonstrated that the nL-MS cements containing nL-MS clearly enhanced the vascularization and osteogenic differentiation of cells as compared to the CSD. The results also proved the feasibility that the incorporation of proper amount of nL-MS into CSD could enhance the stimulatory effects on vascularization and osteogenic differentiation as compared to CSD.

Previous studies have shown that the Si and Li ions could promote the proliferation and differentiation of osteoblasts.^{14,30} In this study, the degradation of nL-MS could lead to gradual release of Si and Li ions both *in vitro* and *in vivo*, which might be critically important for regulating various cellular processes of osteoblasts, including cell proliferation, differentiation and new bone tissue formation. Therefore, the better bone-regenerating ability of nL-MS cements was due to fast degradation after addition of nL-MS and the release of Si and Li ions, and formed a Li- and Si-rich microenvironment *in vivo*, which might show the synergistic effects on stimulating the proliferation and differentiation of

osteoblasts, and then promoted osteogenesis and bone tissue regeneration.

Conclusion

In this study, degradable nl-MSC cements were prepared by adding nl-MS into CSH. The results revealed that the incorporation of nl-MS into the cements obviously improved the degradability of nl-MSC cements both in vitro and in vivo. Moreover, the nl-MSC cements containing nl-MS showed the ability to neutralize the acidic degradation by-products released from the cements and prevented the pH from decreasing during degradation. The SRmCT and histological elevation results revealed that the nl-MSC cements containing nl-MS obviously enhanced new bone formation and promoted the degradability of the cements in vivo, indicating good biocompatibility. In addition, the immunohistochemical analysis results revealed that the nl-MSC cements containing nl-MS could promote the positive expression of VEGF and COL-I, indicating that nl-MSC could enhance the vascularization and osteogenic differentiation in vivo. In short, the nl-MSC cements with good biocompatibility and degradability could promote vascularization and osteogenesis, which have great potential to treat bone defects.

Acknowledgments

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Disclosure

The authors report no conflicts of interest in this work.

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