



Mechanical Improvement of Gas Monitoring System in Monoplace Hyperbaric Chamber to Advance the Safety and Efficacy

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Introduction: A Monoplace hyperbaric chamber delivers oxygen to the patient's tissues through breathing. Gas monitoring inside the chamber is important because oxygen (O₂) is consumed, and carbon dioxide (CO₂) is increased because treatment is performed in a closed volume. This study aimed to advance the safety and efficacy of the monoplace hyperbaric chamber (MHC) through mechanical improvement in a gas monitoring system (GMS).

Methods: First, as the oxygen supply method was changed to the direction of the patient's face, it was compared the values of O₂, CO₂, humidity, and temperature were measured in the MHC and the GMS when operating at 2.0 atmosphere absolute (ATA) and 3.0 ATA. Second, to evaluate the effects of variables across measuring time, it was analyzed in a 3-way repeated measure ANOVA (10 min.×20 min.×30 min.). Lastly, the values before and after the optimization of the MHC were compared by applying a cooler to prevent temperature rise inside the MHC.

Results: In 2.0 ATA, the average humidity was higher in the MHC than in the GMS (p<0.001). Also, the average temperature was lower in the MHC than in the GMS (p<0.001). In 3.0 ATA, the average CO₂ and humidity were higher in the MHC than in the GMS, respectively (p<0.001, p=0.004). The 3-way repeated measures ANOVA revealed a significant difference in most main and interacted factors (p<0.05). O₂ and temperature, comparing before and after MHC optimization, revealed a significant difference (p<0.05).

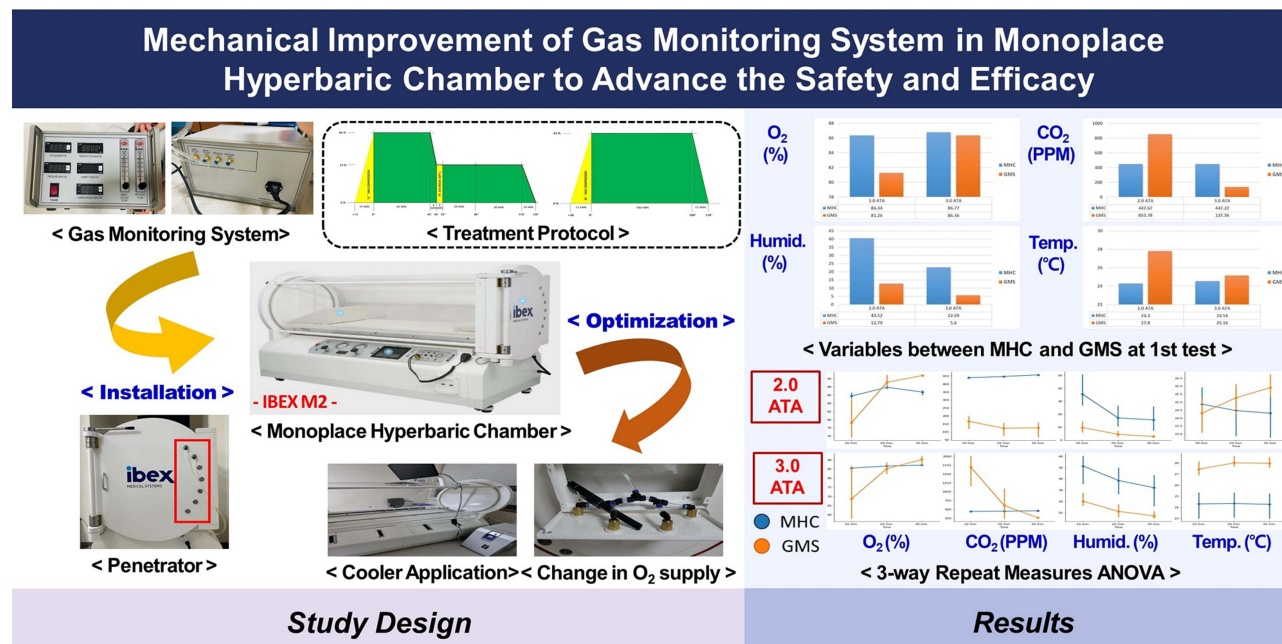
Conclusion: Few studies have verified safety and effectiveness by evaluating the pressure, oxygen concentration, etc. of a monoplace hyperbaric chamber. Further research is expected to verify the effectiveness of providing comfort to patients receiving hyperbaric oxygen treatment and increase the treatment effect.

Keywords: monoplace hyperbaric chamber, gas monitoring system, mechanical improvement, safety and efficacy

Introduction

In humans, oxygen plays an important role in providing energy and maintaining life and function.¹ Oxygen in the blood is divided into combined and dissolved oxygen. Since oxygen has low solubility at atmospheric pressure, most of the oxygen during respiration is combined with hemoglobin in the blood to become bound oxygen, and the remaining oxygen is released into the air through gas exchange or dissolved in plasma (dissolved oxygen).² Combined oxygen, which is bigger than dissolved oxygen, passes smoothly through capillaries to each organ and cell that requires it. When blood vessels are abnormally narrowed due to unhealthy diet, smoking, drinking, obesity, etc., or when blood obesity increases due to diabetes or hyperlipidemia, the bound oxygen combined with hemoglobin cannot pass through capillaries. A variety of diseases and defects are associated with a lack of oxygen supply (hypoxia), which can lead to severe disability and even death.³

Graphical Abstract



The first report on the therapeutic use of hyperbaric oxygen therapy (HBOT) was published in 1879 by Fontaine, a French surgeon who believed that pressurized oxygen chambers could help with anesthesia.⁴ HBOT is used to treat patients by supplying 100% oxygen at more than twice the atmospheric pressure. This therapy is commonly applied to treat cesarean disease, gas poisoning, wounds, and burns. It is known to be effective in decompression sickness (DCS), carbon monoxide (CO) poisonings, diabetic foot, etc. Research on indications is being actively conducted.^{5,6} Also, the Undersea and Hyperbaric Medicine Society (UHMS) in the United States provides guidelines and recommendations for the use of HBOT in various medical conditions. For clinical efficacy, it was specified that the pressure must be greater than or equal to 1.4 atmosphere absolute (ATA), and in clinical practice, pressures applied usually range from 2.0 to 3.0 ATA.⁷

Hyperbaric chambers are used in high-pressure environments where a pressure of at least two or more times atmospheric pressure is set and maintained. It is treated for 45 to 300 minutes in a pressurized environment with 1 atm or 2 atm pressure depending on the purpose or severity. According to recent studies on the indications of HBOT, it is expected to have positive effects in the treatment of hypoxia, oxidative stress, inflammation-induced brain damage, congenital cerebral palsy, stroke, autism, Alzheimer's disease, Parkinson's syndrome, and chronic Lyme disease, and cancers.⁸ Although no clear evidence for the effects and mechanism of HBOT on the treatment of intractable diseases has been discovered, it has shown favorable results in many clinical cases. According to the number of users, it was classified into monoplace and multiplace chamber.⁹ Monoplace chamber delivers oxygen to the patient's tissues through breathing, and multiplace chamber delivers oxygen to the patient's tissues through a breathing aid such as a mask.

In the case of a monoplace chamber, gas monitoring inside the hyperbaric chamber is important because oxygen is consumed, and carbon dioxide is increased because treatment is performed in a closed volume. Values may vary slightly depending on where the oxygen and carbon dioxide concentrations are measured in the hyperbaric chamber. In general, 100% oxygen is supplied during HBOT. Still, the oxygen concentration in the hyperbaric chamber may not be constant because a difference in oxygen partial pressure may occur according to a pressure change. The goal is to increase the effectiveness of HBOT by measuring the concentration change of these internal gases and supplying low carbon and high oxygen to the patient. In addition, the temperature and humidity are fluctuated together according to the change in pressure in the closed treatment device, and when the pressure rises, the temperature and humidity also rise together. It is

important to measure the changing pressure and maintain the temperature and humidity according to the pressure change in order to provide the optimal treatment environment to the patient and to become an accurate HBOT.

Breathing plays a critical role in hyperbaric oxygen therapy (HBOT) as the efficiency of oxygen delivery to tissues is directly influenced by respiratory mechanics. Lung physiology under hyperbaric conditions differs significantly from that at normal atmospheric pressure.¹⁰ The increased pressure enhances the amount of oxygen dissolved in plasma, thereby improving oxygen availability to hypoxic tissues. Conditions in hyperbaric chambers such as pressure, temperature, and humidity must be meticulously controlled to optimize patient safety and therapeutic outcomes.¹¹ Monitoring breathing parameters and ensuring the stability of these environmental conditions is crucial for the effectiveness of HBOT.^{12,13} Furthermore, understanding the impact of lung physiology in hyperbaric environments helps in designing better therapeutic protocols and improving patient care.

According to the current status of hyperbaric chamber deployment investigated by the Korean Academy of Undersea and Hyperbaric Medicine, the number of medical institutions capable of HBOT in Korea increased from 21 to 50 in 2021 compared to the first survey in 2017. The details of number of medical institutions capable of HBOT increased from 8 to 13 for tertiary general hospitals, from 3 to 17 for general hospitals, and from 10 to 20 for hospitals and other medical institutions. The number of multiplace hyperbaric chambers increased from 12 to 30, and the number of monoplace hyperbaric chambers increased from 13 to 43.¹⁴

However, to our knowledge, no study has verified safety and effectiveness by evaluating the pressure, oxygen concentration, etc. of a monoplace hyperbaric chamber. This study aims to advance the safety and efficacy of monoplace hyperbaric chamber through mechanical improvement in a gas monitoring system.

Materials and Methods

Monoplace Hyperbaric Chamber and Ethical Approval

IBEX M2 (IBEX Medical Systems Co. Ltd., Republic of Korea) was a newly developed and recently commercialized monoplace hyperbaric chamber as a medical device for testing (Figure 1). It was applied the automatic control system including the anti-barotrauma technology. This study was approved by the Institutional Review Board of Wonju Christian Hospital (IRB approval no.: CR320074) and the Clinical Research Information Service of Korea Disease Control and Prevention Agency (Trial registration no.: KCT0005974).

Development of the Gas Monitoring System and Its Connection to the Chamber

A sensor kit was developed by selecting a data recorder and a transmitter (Figure 2). The O2H-9903SD (LT Lutron Electronic Enterprise Co. Ltd., Taiwan) is the data recorder that was capable of O₂, humidity, and temperature detection into the SD memory card. It has specified measurement ranges and accuracies for each oxygen, humidity, and temperature as follows; 0~30% O₂±(1% reading+0.2%), 5~95% RH±3% (below 70% RH), and 0~50±0.8°C or 32~122±1.5°F. Also, the CD-100 (ELT SENSOR Corp., Republic of Korea) is the one-board type transmitter that was capable of measuring CO₂ concentration. It has a specified measurement range of 0~2000 ppm and measurement



Figure 1 Equipment information. This device is a newly developed and recently commercialized monoplace hyperbaric chamber as a medical device applied to the automatic control system including the anti-barotrauma technology.

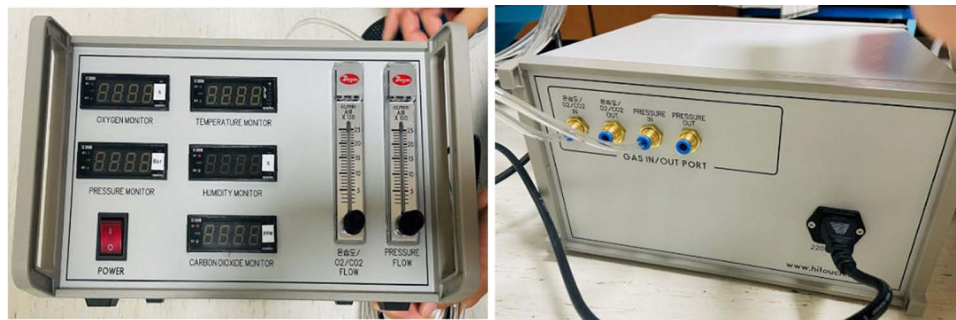


Figure 2 Development of a gas monitoring system. The gas monitoring system is developed as a sensor kit by selecting a data recorder and a transmitter. The data recorder is capable of O₂, humidity, and temperature detection into the SD memory card. Also, the transmitter is a one-board type transmitter capable of measuring CO₂ concentration.

accuracy of $5\% \pm 30$ ppm. The developed gas monitoring system was connected with a hose after installing a penetrator on the door side of the monoplace hyperbaric chamber (Figure 3).

Optimization of Monoplace Hyperbaric Chamber

The sensor kit of the gas monitoring system was upgraded to increase the accuracy of the measured oxygen concentration applied to the chamber. A flow function according to the carbon monoxide concentration value was added, and an air brake system application method was configured. The oxygen supply method was changed in the direction of the patient's face so that the patient could receive a comfortable HBOT (Figure 4A). A cooler was applied to control the temperature inside the chamber, and a smartphone remote function was used (Figure 4B).

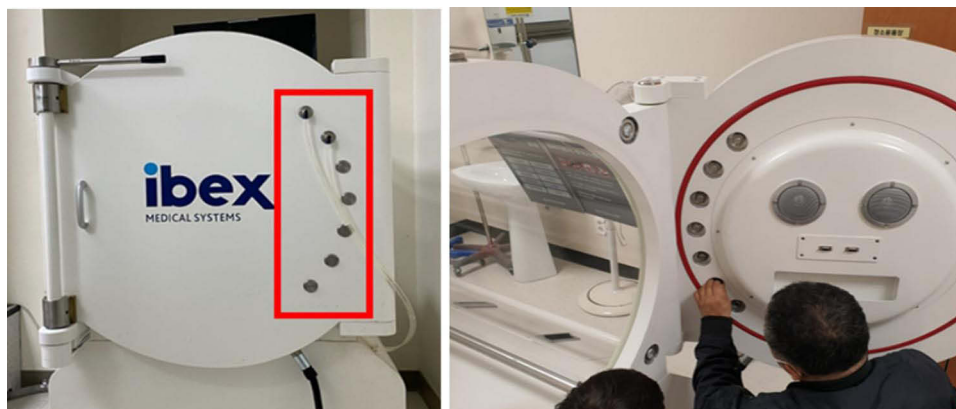


Figure 3 Installation of a penetrator for sensing on the door of the monoplace hyperbaric chamber. The developed gas monitoring system is connected with a hose after installing a penetrator on the door side of the monoplace hyperbaric chamber.

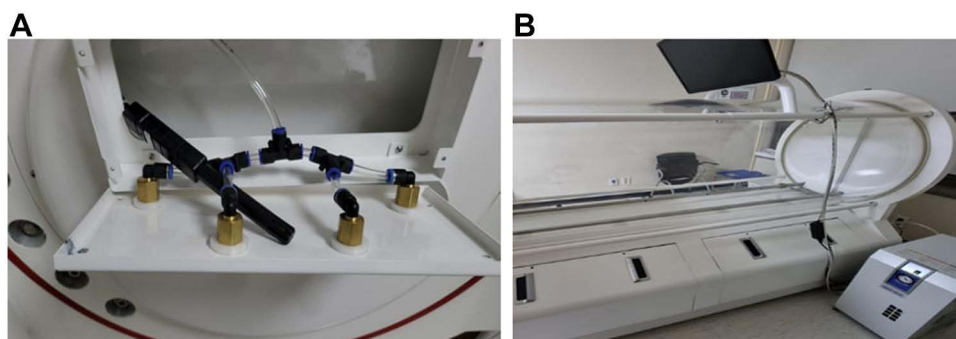


Figure 4 (A) Changes in oxygen supply method **(B)** Cooler application. Optimization of monoplace hyperbaric chamber. A flow function according to the carbon monoxide concentration value was added, and an air break system application method was configured. The oxygen supply method was changed in the direction of the patient's face so that the patient. A cooler was applied to control the temperature inside the chamber, and a smartphone remote function was used.

Pressure Value Settings for the Safety and Effectiveness Evaluation of Monoplace Hyperbaric Chamber

The acute carbon monoxide poisoning treatment protocol, which is the most commonly used indication for hyperbaric oxygen treatment at the HBOT Center at Wonju Severance Christian Hospital, was applied in this study. Based on the hyperbaric oxygen treatment guidelines established by the US Navy, specialists from the Korean Society of Hyperbaric Medicine partially modified this protocol to change the pressure more slowly for 15 minutes so that patients can smoothly adapt to decompression (Figure 5).¹⁵ The 1st treatment was performed using 3.0 ATA and 2.0 ATA for 135 minutes. Before starting the HBOT session, it was prepared for max compression for 15 minutes and then performed initial compression for 45 minutes to 3.0 ATA. Then, the pressure was reduced from 3.0 ATA to 2.0 ATA for 5 minutes, and the air brake was maintained for 5 minutes. Next, it was followed by 2.0 ATA for 55 minutes and ended with depressurization for 10 minutes (Figure 5A). Other than the 1st treatment was performed using 2.0 ATA for a total of 110 minutes. Before starting the HBOT session, it was prepared for max compression for 20 minutes and then performed initial compression for 100 minutes to 2.0 ATA. After that, it was decompressed for 15 minutes (Figure 5B).

Study Design

When the monoplace hyperbaric chamber was operated at 2.0 ATA and 3.0 ATA, the values of oxygen, carbon dioxide, humidity, and temperature were measured in the chamber and gas monitoring system, respectively. In 2.0 ATA, the subject entered the chamber, but in 3.0 ATA, the subject did not enter the chamber to prevent a situation in which the subject could be in danger due to high pressure. The data has measured a total of three times during the chamber operating at 10 minutes, 20 minutes, and 30 minutes after pressurization, and this sequence was repeated three times. Finally, after optimizing the monoplace hyperbaric chamber based on the descriptive results of variables according to air pressure conditions, the second test was performed again using the same process.

Statistical Analysis

Statistical analysis was performed using IBM SPSS Statistics Version 27.0 (IBM Corp., Armonk, NY, USA). In this study, the values measured in the monoplace hyperbaric chamber and a gas monitoring system were compared. Comparisons among the measured data between the monoplace hyperbaric chamber and a gas monitoring system were expressed as means with the standard deviations for continuous variables. Differences were assessed using the independent *t*-test or Mann–Whitney *U*-test for continuous variables and the χ^2 test or Fisher exact test for categorical variables. It was interpreted as statistically significant when the p-value was less than 0.05. To evaluate changes in the values measured at 10, 20, and 30 minutes after pressurization, a 3-way repeated measures analysis of variance (ANOVA) was

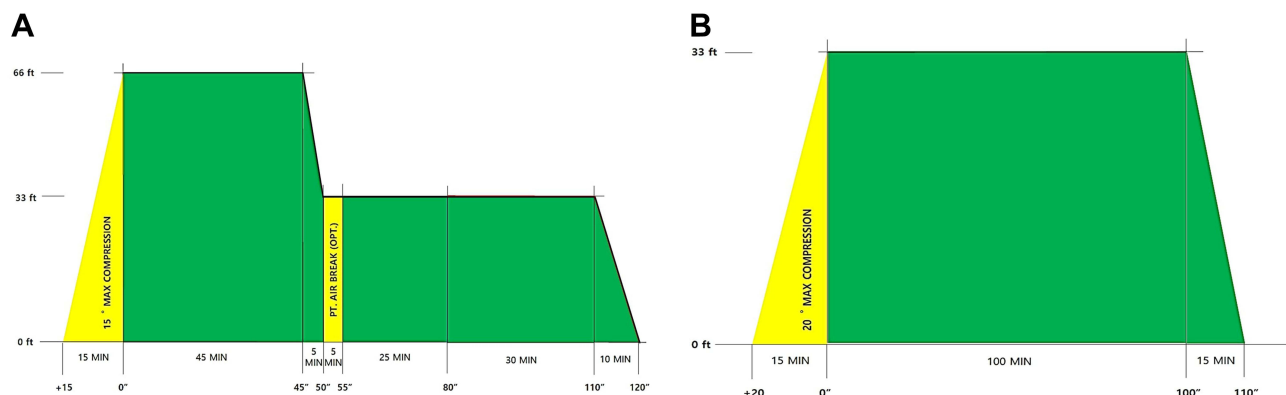


Figure 5 (A) Protocol for 1st treatment **(B)** Protocol other than 1st treatment. This figure shows the acute carbon monoxide poisoning treatment protocol, which is the most commonly used indication for hyperbaric oxygen treatment in the hospital, was applied in this study. The green section is supplied with 100% oxygen, and the yellow section is supplied with air. This protocol allows for changes in pressurization and decompression times and air breaks at the discretion of the physician.

performed. Assumptions of sphericity were examined, and when violated, the Greenhouse-Geisser correction was applied.

Results

In the air pressure condition at 2.0 ATA, the average humidity was higher in the chamber than in the gas monitoring system ($40.52 \pm 14.13\%$ vs $12.79 \pm 7.31\%$, $p < 0.001$). Also, the average temperature was lower in the chamber than in the gas monitoring system ($24.30 \pm 0.99^\circ\text{C}$ vs $27.80 \pm 0.54^\circ\text{C}$, $p < 0.001$). In the air pressure condition at 3.0 ATA, the average carbon dioxide was higher in the chamber than in the gas monitoring system (447.22 ± 7.85 PPM vs 137.56 ± 43.70 PPM, $p < 0.001$). Also, the average humidity was higher in the chamber than in the gas monitoring system ($22.69 \pm 13.14\%$ vs $5.60 \pm 3.93\%$, $p = 0.004$) (Table 1).

The 3-way repeated measures ANOVA showed the significance of the device (D), measuring time (T), and interactions between the device and measuring time (DxT) on the effects of variables across measuring time (Table 2). In 2.0 ATA air pressure conditions, significant main effects were observed for the time in O₂, the device in CO₂, and the time in humidity ($p < 0.001$). Also, the time in CO₂, the device in humidity, the device in temperature, and the time in temperature revealed significant main effects respectively ($p = 0.003$, $p = 0.029$, $p = 0.009$, $p = 0.010$). Significant interactions were observed between device and time in O₂, CO₂, and temperature, respectively ($p = 0.001$, $p = 0.002$, $p = 0.003$). In 3.0 ATA air pressure conditions, significant main effects were observed for the device in CO₂, the time in humidity, and the time in temperature ($p < 0.001$). Also, the time in O₂, the time in CO₂, and the device in humidity revealed significant main effects respectively ($p = 0.008$, $p = 0.016$, $p = 0.047$). Significant interactions were observed between the device and time in CO₂, and temperature ($p < 0.001$). Furthermore, the interaction between the device and time in O₂, and humidity revealed significant effects respectively ($p = 0.020$, $p = 0.010$).

Descriptive statistics for variables according to air pressure conditions before and after optimization of the monoplace hyperbaric chamber (mean \pm SD) are presented (Table 3). In 2.0 ATA air pressure condition, the paired samples *t*-test comparing O₂ before and after optimization of the monoplace hyperbaric chamber revealed a significant difference ($t(df) = -14.133$, $p = 0.040$). The *p*-value of less than 0.001 indicated that the difference between temperature before and after optimization was statistically significant. In 3.0 ATA air pressure condition, the paired samples *t*-test comparing O₂, humidity, and temperature before and after optimization of the monoplace hyperbaric chamber revealed a significant difference, respectively ($t(df) = -7.467$, $p = 0.038$ / $t(df) = -10.800$, $p = 0.002$ / $t(df) = 1.767$, $p = 0.013$).

Table 1 Descriptive Analysis (Mean, SD) of Variables According to Air Pressure at the First Test

Source		MHC	GMS	p-value
2.0 ATA	O ₂ (%)	86.34 \pm 0.75	81.26 \pm 12.62	0.261
	CO ₂ (PPM)	447.67 \pm 8.00	853.78 \pm 719.13	0.129
	Humidity (%)	40.52 \pm 14.13	12.79 \pm 7.31	< 0.001
	Temperature (°C)	24.30 \pm 0.99	27.80 \pm 0.54	< 0.001
3.0 ATA	O ₂ (%)	86.77 \pm 1.06	86.36 \pm 5.98	0.844
	CO ₂ (PPM)	447.22 \pm 7.85	137.56 \pm 43.70	< 0.001
	Humidity (%)	22.69 \pm 13.14	5.60 \pm 3.93	0.004
	Temperature (°C)	24.54 \pm 1.20	25.16 \pm 1.08	0.272

Notes: *p*-value by Fisher's exact test.

Abbreviations: MHC, Monoplace hyperbaric chamber; GMS, Gas monitoring system; ATA, Atmosphere absolute; PPM, Parts per million; °C, Celsius.

Table 2 Results of 3-Way Repeated Measures ANOVA for the Effects of Variables Across Measuring Time

Source			Sum of Squares	df	Mean Square	F	p-value	Mauchly's Sphericity
2.0 ATA	O ₂ (%)	Device (D)	49.882	1	49.882	5.803	0.074	< 0.001
		Time (T)	762.648	1.005	758.880	75.319	< 0.001†	
		DxT	613.990	1.005	610.957	60.638	0.001†	
		Error	40.502	4.020	5.063	-	-	
	CO ₂ (PPM)	Device (D)	247,389.352	1	2,540,636.463	85.751	< 0.001	0.888
		Time (T)	1,633,523.111	1.858	879,108.679	14.892	0.003	
		DxT	1709921.778	1.858	920,223.942	15.588	0.002	
		Error	438,767.778	7.433	54,845.972	-	-	
	Humidity (%)	Device (D)	1248.003	1	1248.003	11.212	0.029	0.492
		Time (T)	996.154	1.453	685.744	72.883	< 0.001	
		DxT	72.601	1.453	49.978	5.312	0.055	
		Error	54.671	5.811	9.409	-	-	
	Temperature (°C)	Device (D)	18.375	1	18.375	23.049	0.009	0.529
		Time (T)	0.310	1.486	0.155	20.667	0.010	
		DxT	0.303	1.486	0.204	20.222	0.003	
		Error	0.060	5.945	0.010	-	-	
3.0 ATA	O ₂ (%)	Device (D)	0.254	1	0.254	0.101	0.767	0.001
		Time (T)	149.818	1.006	148.855	23.218	0.008†	
		DxT	89.298	1.006	88.724	13.839	0.020†	
		Error	25.811	4.026	6.411	-	-	
	CO ₂ (PPM)	Device (D)	143,840.167	1	143,840.167	154.044	< 0.001	0.897
		Time (T)	993.778	1.869	531.781	7.710	0.016	
		DxT	3065.333	1.869	1640.292	23.783	< 0.001	
		Error	515.556	7.475	68.970	-	-	
	Humidity (%)	Device (D)	438.045	1	438.045	8.067	0.047	0.162
		Time (T)	641.364	1.174	546.107	61.683	< 0.001	
		DxT	170.751	1.174	145.391	16.422	0.010	
		Error	41.591	4.698	8.853	-	-	
	Temperature (°C)	Device (D)	0.560	1	0.560	0.412	0.556	0.096
		Time (T)	0.803	1.117	0.719	36.150	< 0.001	
		DxT	3.601	1.117	3.223	162.050	< 0.001	
		Error	0.089	4.469	0.020	-	-	

Notes: † Greenhouse-Geisser correction.

Abbreviations: ATA, Atmosphere absolute; PPM, Parts per million; °C, Celsius; df, Degree of freedom; Device, Monoplace hyperbaric chamber, and Gas monitoring system; Time, measuring time (10 min., 20 min., 30 min.).

Table 3 Comparison of Variables Before and After Optimization of the Monoplace Hyperbaric Chamber

Source			Paired Samples t-test		t (p-value)
			N	Mean±SD	
2.0 ATA	O ₂ (%)	Pre-Optimization	6	76.83±13.47	-14.133 (0.040)
		Post-Optimization	6	90.97±1.26	
	CO ₂ (PPM)	Pre-Optimization	6	1149.17±716.43	185.833 (0.586)
		Post-Optimization	6	963.33±756.80	
	Humidity (%)	Pre-Optimization	6	15.60±7.24	-5.600 (0.215)
		Post-Optimization	6	21.20±11.19	
Temperature (°C)	Pre-Optimization	6	27.72±0.59	3.017 (< 0.001)	
	Post-Optimization	6	24.70±0.28		
3.0 ATA	O ₂ (%)	Pre-Optimization	6	84.17±6.32	-7.467 (0.038)
		Post-Optimization	6	91.63±1.32	
	CO ₂ (PPM)	Pre-Optimization	6	143.83±43.25	-26.333 (0.494)
		Post-Optimization	6	170.17±52.11	
	Humidity (%)	Pre-Optimization	6	7.03±4.12	-10.800 (0.002)
		Post-Optimization	6	17.83±3.07	
Temperature (°C)	Pre-Optimization	6	24.78±1.04	1.767 (0.013)	
	Post-Optimization	6	23.02±0.43		

Notes: p-value by Fisher's exact test.

Abbreviations: ATA, Atmosphere absolute; PPM, Parts per million; °C, Celsius.

Discussion

This study aimed to advance the safety and efficacy of a monoplace hyperbaric chamber through mechanical improvement in a gas monitoring system. The developed gas monitoring system was connected with a hose after installing a penetrator on the door side of the chamber. As the oxygen supply method was changing toward the direction of the patient's face, it made the patient receive a comfortable HBOT while increasing the accuracy of the measured oxygen concentration applied to the chamber. Sechrist et al reported that it described the use of the HBO O₂ Smart Guide, an Excel-based program that could calculate hyperbaric oxygen concentrations, given a set of initial conditions. In contrast, it was described how the program was validated by measuring the oxygen in-chamber, using 4 different monoplace chamber models, and employing 2 types of oxygen purge profiles.¹⁶ Sumbul and Yuzer presented that the integration of MEMS-based diagnostic tools for respiratory monitoring within monoplace hyperbaric chambers offers significant advantages. The detection of respiratory rate (RR), tidal volume capacity (TVC), and forced vital capacity (FVC) using a MEMS-based acceleration sensor can provide real-time, accurate respiratory measurements. These tools record the acceleration data of diaphragm movements in three axes during respiration, ensuring precise monitoring of the patient's respiratory parameters during HBOT sessions.¹³ Also, they reported another study involving five subjects (three healthy and two with COPD) demonstrated the reliability of this MEMS-based system. It was able to measure and record respiratory parameters accurately, with results closely matching those from a medical spirometer. Such accuracy is crucial in hyperbaric environments where precise oxygen delivery and patient monitoring are essential for safety and treatment efficacy.¹⁰ Furthermore, the development of a practical method for estimating air volume using acceleration data from diaphragm movements reinforces the potential of these advanced diagnostic tools. By providing real-time data,

clinicians can dynamically adjust treatment parameters, enhancing the safety and effectiveness of HBOT. This integration aligns perfectly with the goal of mechanical improvement in gas monitoring systems in monoplace hyperbaric chambers, ensuring better patient outcomes.¹²

Table 1 compares the values measured in the existing chamber with those measured in the newly developed gas monitoring system. In this study, the average humidity was higher in the chamber than in the gas monitoring system in the air pressure condition at 2.0 ATA and the average temperature was lower in the chamber than in the gas monitoring system in the air pressure condition at 3.0 ATA. The difference in measured values between the hyperbaric oxygen chamber and gas monitoring system at 2.0 ATA and 3.0 ATA could be caused by a variety of factors. First, the difference in measured values could be caused by the pressure differences. Similarly, Ay et al found that increased pressure in hyperbaric chambers could lead to increased levels of humidity due to reduced ventilation and moisture from medical gases.¹⁷ Second, the difference in measured values could be caused by the ventilation differences. The hyperbaric chamber might have different ventilation settings than the gas monitoring system. At 2.0 ATA, where subjects entered the chamber, the CO₂ levels were notably higher. This elevation in CO₂ levels can be attributed to human respiration, a natural process that results in the production of carbon dioxide as a metabolic byproduct. The relationship between pressure and CO₂ levels is intricate but can be elucidated by considering Henry's Law, which states that the solubility of a gas (such as CO₂) in a liquid (such as blood) is directly proportional to the pressure of that gas above the liquid.¹⁸ Thus, at 2.0 ATA, the higher pressure and increased CO₂ production from respiration likely contributed to the elevated CO₂ readings. Conversely, at 3.0 ATA where no subjects were present due to safety concerns associated with high pressure, the CO₂ levels were lower. Without human respiration as a significant source of CO₂, the primary mechanisms contributing to CO₂ levels would be environmental factors and any potential off-gassing from materials within the chamber. The measured differences in CO₂ levels between the two pressure conditions can be attributed to the presence or absence of individuals and their corresponding metabolic activities, particularly respiration. Understanding the relationship between pressure and CO₂ levels, along with the influence of human presence, is considered crucial for interpreting gas monitoring system data accurately in hyperbaric environments.¹⁹ Lastly, the difference in measured values could be caused by the temperature differences. The hyperbaric chamber might be at a different temperature than the gas monitoring system, which could affect the readings of humidity and temperature. At 2.0 ATA, the lower average temperature readings in the hyperbaric chamber might be due to the lower ambient temperature in the chamber.²⁰

Table 2 presents the significance of the device (D), measuring time (T), and interactions between the device and measuring time (DxT) on the effects of variables across measuring time by the 3-way repeated measures ANOVA. In summary, under 2.0 ATA air pressure conditions, the 3-way repeated measures ANOVA revealed significant main effects of the time in O₂, the device in CO₂, the time in CO₂, the device in humidity, the time in humidity, the device in temperature, and the time in temperature, as well as an interaction between device and time in O₂, CO₂, and temperature. Under 3.0 ATA air pressure conditions, significant main effects were observed for the device in CO₂, the time in humidity, and the time in temperature. Also, the time in O₂, the time in CO₂, and the device in humidity revealed significant main effects. Significant interactions were observed between the device and time in all variables revealing significant effects. Although we could not find any previous studies that measured and monitored O₂, CO₂, humidity, and temperature in the monoplace hyperbaric chamber like this study, Gertner et al reported a time-dependent study using the chamber using a 3-way repeated measures ANOVA methodology. There were several potential confounding factors identified that may have affected the observed results to determine the efficacy of various drugs to decrease the incidence of Eustachian Tube Dysfunction in Divers.²¹

Table 3 compared the values before and after optimization of a monoplace hyperbaric chamber by applying a cooler to prevent temperature rise inside the chamber. In 2.0 ATA air pressure condition, the paired samples *t*-test comparing O₂ and temperature before and after optimization of the monoplace hyperbaric chamber revealed a significant difference. In 3.0 ATA air pressure condition, the paired samples *t*-test comparing O₂, humidity, and temperature before and after optimization of the monoplace hyperbaric chamber revealed a significant difference. In this study, O₂ was increased after optimization of the monoplace hyperbaric chamber. Similarly, the prevailing perspective suggests that the effectiveness of HBOT is positively correlated with increasing oxygen concentrations. In HBOT, patients are exposed to elevated atmospheric pressures while breathing pure oxygen, aiming to augment the amount of dissolved oxygen in the

bloodstream. Although some proponents argue that higher oxygen concentration under hyperbaric oxygen conditions improves oxygen supply, accelerates the healing process, reduces inflammation, and improves the overall treatment effect, what actually increases the effectiveness of hyperbaric oxygen treatment is the medical staff's This can be attributed to careful consideration of factors, adherence to established safety guidelines, and awareness of the potential risks associated with excessive oxygen levels, including oxygen toxicity. Also, the temperature was decreased after optimization of the monoplace hyperbaric chamber. HBOT is often associated with providing a comfortable therapeutic environment, and the temperature during HBOT sessions plays a crucial role in enhancing patient comfort. As the ambient pressure increases within the hyperbaric chamber, the temperature is carefully controlled to ensure the well-being and comfort of the patient. The controlled temperature not only contributes to the overall comfort but also plays a role in mitigating potential discomfort that may arise from the increased pressure. Maintaining a cooler temperature within the chamber is often considered beneficial, as it can alleviate any potential feelings of warmth or claustrophobia associated with the enclosed environment. Additionally, a cooler environment may help minimize the risk of overheating, especially during longer treatment sessions. Overall, careful control of the temperature of the monoplace hyperbaric chamber is a key aspect of ensuring the safety of patients undergoing HBOT and providing them with a positive and comfortable experience while receiving treatment.²²

Conclusion

Despite the limitations in this study, to our knowledge, no study has verified the safety and effectiveness of the monoplace hyperbaric chamber by evaluating its pressure and oxygen concentration. Therefore, this study monitored the conditions inside the hyperbaric oxygen chamber. This study was meaningful in that it attempted to develop a gas monitoring system that operates and improves its safety and efficacy. Lastly, based on the results of this study, additional research is needed to verify whether hyperbaric oxygen treatment actually increases treatment effectiveness and provides comfort to patients receiving hyperbaric oxygen treatment.

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Disclosure

The authors report no conflicts of interest in this work.

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