

Cisplatin-incorporated nanoparticles of poly(acrylic acid-co-methyl methacrylate) copolymer

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Background: Although cisplatin is extensively used in the clinical field, its intrinsic toxicity limits its clinical use. We investigated nanoparticle formations of poly(acrylic acid-co-methyl methacrylate) (PAA-MMA) incorporating cisplatin and their antitumor activity in vitro and in vivo.

Methods: Cisplatin-incorporated nanoparticles were prepared through the ion-complex formation between acrylic acid and cisplatin. The anticancer activity of cisplatin-incorporated nanoparticles was assessed with CT26 colorectal carcinoma cells.

Results: Cisplatin-incorporated nanoparticles have small particle sizes of less than 200 nm with spherical shapes. Drug content was increased according to the increase of the feeding amount of cisplatin and acrylic acid content in the copolymer. The higher acrylic acid content in the copolymer induced increase of particle size and decrease of zeta potential. Cisplatin-incorporated nanoparticles showed a similar growth-inhibitory effect against CT26 tumor cells in vitro. However, cisplatin-incorporated nanoparticles showed improved antitumor activity against an animal tumor xenograft model.

Conclusion: We suggest that PAA-MMA nanoparticles incorporating cisplatin are promising carriers for an antitumor drug-delivery system.

Keywords: cisplatin, nanoparticle, poly(acrylic acid-co-methyl methacrylate), ion complexes

Introduction

Cisplatin has been extensively used for clinical treatment of various tumors.¹ However, its treatment regimen in the clinical field needs to be improved or altered, because cisplatin has severe toxicity against the human body, eg, acquired resistance, acute nephrotoxicity, and chronic neurotoxicity.^{2,3} Novel drug-delivery systems based on nanoparticles, liposomes, polymeric micelles, and polymer conjugates have been proposed as promising vehicles for delivery of cisplatin to tumors and for reduction of neuro-/nephrotoxicity.⁴⁻⁹ Polymeric micelles incorporating cisplatin were reported to have improved anticancer activity, reduced intrinsic toxicity, and prolonged blood circulation.^{4,5} Cisplatin-incorporated polymeric micelles were based on ion complexation between cisplatin and anionic polymers.⁴ Their polymeric micelles revealed significant reduction in intrinsic toxicity of cisplatin, with improvement in anticancer effect against an animal tumor xenograft model. Nanoparticles based on cisplatin/carboxymethyl cellulose core and poly(lactide-co-glycolide)-methoxy-poly(ethylene glycol) copolymers were also reported to improve anticancer activity and reduce toxicity.⁶ Leite et al reported that pH-sensitive liposomes were effective in reducing mass of

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solid tumors in mice and intrinsic toxicity of drugs.⁷ Sonoda et al reported on dextran magnetite-conjugated cisplatin for hyperthermic treatment of solid tumors.⁸ In this report, they obtained magnetic accumulation of dextran magnetite-conjugated cisplatin with hyperthermic effects.

The aim of this study was to investigate the potential of cisplatin-incorporated nanoparticles using poly(acrylic acid-co-methyl methacrylate) (PAA-MMA) copolymer against CT26 colon carcinoma cells. Since PAA-MMA has plenty of carboxylic acid moieties, nanoparticles incorporating cisplatin can be formed through ion complexation between cisplatin and PAA. We investigated the physicochemical properties of the PAA-MMA nanoparticles incorporating cisplatin, and their antitumor activity was evaluated against CT26 color rectal carcinoma cells *in vitro* and *in vivo*.

Materials and methods

Materials

Sodium acrylate (AA), methyl methacrylate (MMA), poly(ethylene glycol) diacrylate (PEGDA; molecular weight [M_w] 575 g/mol), dialysis tubing (M_w cutoff 12,000 g/mol), dimethylformamide (DMF), and dimethyl sulfoxide (DMSO) were purchased from Sigma-Aldrich (St Louis, MO, USA). DMF and DMSO was used as extra-pure grade. Cisplatin was purchased from Wako Pure Chemical Industries (Osaka, Japan).

Synthesis of PAA-MMA copolymer

PAA-MMA sodium salt was synthesized in deionized water by free radical emulsion polymerization, as previously reported.¹⁰ Briefly, AA and MMA were polymerized in the presence of 100 mg ammonium persulfate (APS) as a crosslinker and 100 microliter PEGDA as an initiator. The monomers and chemicals in a round-bottomed flask were stirred at 900 rpm at 65°C for 6 hours under nitrogen gas atmosphere. In this reaction, the feed molar ratio of AA versus MMA ranged from 0.7 to 0.9. The resulting solution was centrifuged at 100,000 g for 30 minutes and dried in a vacuum. White solid was placed in a refrigerator until use.

Analysis of PAA-MMA copolymer

The M_w of copolymer was evaluated by gel-permeation chromatography (GPC). The GPC system was composed of a 515 HPLC pump, 410 differential refractometer, 486 tunable absorbance detector, and ultrahydrogel 250 and 2000 columns (7.8 mm inner diameter \times 30 cm, flow rate 1.0 mL/minute) (all Waters, Milford, MA, USA). The experimental conditions were as follows: mobile phase

0.1 NaNO₃, flow rate 1.0 mL/minute, laser wavelength 690 nm, temperature 60°C. Polyethylene glycol (PEG) was used to determine M_w as a standard.

Preparation of cisplatin-incorporated nanoparticles

PAA-MMA 100 mg was dissolved in 10 mL deionized water. To this solution, 5–20 mg of cisplatin was added as a solid form. This mixture was stirred gently for 3 days under dark conditions. Since cisplatin itself is not soluble in deionized water, it was dispersed in the aqueous solution of PAA-MMA copolymer as a solid form. At least 2 days were required to dissolve cisplatin in PAA-MMA solution and to form nanoparticles. After that, this solution was dialyzed against deionized water for 6 hours to remove the salt and unloaded drug. The dialyzed solution was used to analyze or lyophilize.

The cisplatin-entrapped nanoparticles were kept in 0.3 M NaCl at 37°C for 3 days to liberate whole cisplatin from nanoparticles. Then, this solution was diluted ten times with DMF. An ultraviolet (UV) spectrophotometer (UV-1201; Shimadzu, Kyoto, Japan) was employed to evaluate the content and loading efficiency of the cisplatin in the nanoparticles at 310 nm.⁹

The drug contents and loading efficiency were evaluated using the following equations:

$$\text{Drug content} = (\text{Amount of liberated cisplatin from the nanoparticles} / \text{Weight of nanoparticles}) \times 100$$

$$\text{Loading efficiency} = (\text{Liberated amount of cisplatin from the nanoparticles} / \text{Feeding amount of cisplatin}) \times 100$$

Analysis of cisplatin-incorporated nanoparticles

The morphology of cisplatin-incorporated nanoparticles was observed with transmission electron microscopy (TEM). A drop of polymeric micelle suspension was placed onto a carbon film coated on a copper grid for TEM observation. The observation was performed at 80 kV in a JEM-2000 FX II (JEOL, Tokyo, Japan).

The particle size of the cisplatin-incorporated nanoparticles was assessed with dynamic laser scattering (DLS-7000; Otsuka Electronics, Hirakata, Japan). The zeta potential of the nanoparticles was measured using photon correlation spectroscopy (Zetasizer 3000; Malvern Instruments, Malvern, UK) with an He-Ne laser beam at a wavelength of 633 nm at 25°C (scattering angle of 90°).

Drug-release study

Nanoparticles (10 mg) were reconstituted in 3 mL deionized water, and this solution was put into a dialysis tube (M_w cutoff 12,000 g/mol). Then, the dialysis tube was put into a 50 mL Falcon tube with 37 mL phosphate-buffered saline (PBS; pH 7.4, 0.01 M). Cisplatin was released at 100 rpm and 37°C. Two milliliters of the sample solution was taken at predetermined time intervals for measurement of drug release, and the solution was supplemented with 2 mL fresh medium. The release amount of cisplatin was measured with the UV spectrophotometer at 310 nm. For comparison, 2 mg free cisplatin was added to 5 mL PBS and magnetically stirred for 6 hours. After that, this solution was introduced into a dialysis tube and a release test performed. For blank testing, PAA-MMA was dissolved in PBS and release testing performed similar to the above method. All experiments were performed in triplicate, and the results were expressed as means \pm standard deviation.

Cell-culture study

For the anticancer activity of cisplatin and its nanoparticles, CT26 colorectal carcinoma cells were used throughout this study. CT26 cells were maintained in a 5% CO₂ incubator (37°C) with Roswell Park Memorial Institute (RPMI) 1640 medium supplemented with 10% fetal bovine serum (FBS) and 1% antibiotics.

Anticancer activity of cisplatin-incorporated nanoparticles in vitro

The anticancer activity of cisplatin-incorporated nanoparticles was assessed in vitro using the MTT (3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide) cell-proliferation assay. A total 3×10^3 CT26 cells were seeded onto 96-well plates and incubated overnight in a 5% CO₂ incubator at 37°C. Cisplatin dissolved in DMSO was diluted with RPMI 1640 medium supplemented with 10% FBS (final concentration of DMSO was less than 0.5%, v/v) and treated in 96-well plates. Cisplatin-incorporated nanoparticles reconstituted in deionized water were diluted ten times with RPMI 1640 medium supplemented with FBS. DMSO (final concentration: 0.5%, v/v) was treated as a control. For comparison, PAA-MMA dissolved in deionized water was diluted ten times with RPMI 1640 medium supplemented with 10% FBS and treated with tumor cells. The tumor cells were incubated for an additional 48 hours at 37°C in a 5% CO₂ incubator. 25 μ l of MTT (3 mg/mL in PBS) was added to

96 well plates and then the cells were further incubated for 4 hours. Subsequently, 100 μ l sodium dodecyl sulfate (SDS)-hydrogen chloride (HCl) solution (SDS 10% w/v, 0.01M HCl) was added to each well and incubated for an additional 12 hours at 37°C in a 5% CO₂ incubator. The absorbance was then measured at a wavelength of 570 nm using an Infinite M200 Pro reader (Tecan Austria, Salzburg, Austria). The viable cells were expressed as a percentage of control. The results were calculated from the eight wells as the means \pm standard deviation of three different experiments.

Apoptosis/necrosis analysis of tumor cells

Apoptosis and necrosis of tumor cells were analyzed with a flow cytometer. Fluorescein isothiocyanate (FITC) annexin V (sc-4252 FITC; Santa Cruz Biotechnology, Santa Cruz, CA, USA) was used to analyze apoptosis of cells, and propidium iodide (PI) reagents (P4170; Sigma-Aldrich) were used to identify necrosis of tumor cells. Cells were exposed to cisplatin or cisplatin-incorporated nanoparticles for 24 hours and then harvested. The harvested cells were resuspended in binding buffer (10 mM 4-[2-hydroxyethyl]-1-piperazineethanesulfonic acid, pH 7.4, 150 mM NaCl, 5 mM KCl, 1 mM MgCl₂, and 1.8 mM CaCl₂) containing FITC annexin V (1 g/mL) and then further incubated for 20 minutes. Ten minutes before the end of incubation, PI (10 g/mL) was added to this cell suspension in order to stain necrotic cells. Cells were analyzed with a FACScan flow cytometer (BD Biosciences, San Jose, CA, USA) equipped with an excitation laser line at 488 nm. The PI was collected through a 575 ± 15 nm band-pass filter.

Anticancer activity of cisplatin-incorporated nanoparticles in vivo

CT26 cells (2×10^4 cells/mouse) were implanted subcutaneously into the back of BALB/c nude mice (5 weeks old, approximately 20 g). When the size of solid tumors in the backs of mice reached approximately 3×3 mm, mice were divided into four groups: the control group was injected with PBS, the drug-treatment group was injected with cisplatin (cisplatin in PBS), the vehicle-treatment group was injected with PAA-MMA polymer (PAA-MMA-1 dissolved in PBS), and the nanoparticle-treatment group was injected with cisplatin-incorporated nanoparticles. Each group consisted of five ear-tagged tumor-bearing mice and was followed up throughout the study. The administration of cisplatin, vehicle, or nanoparticles began on day 14. Cisplatin or nanoparticles were administered at doses of 5 mg/kg,

and vehicle (PAA-MMA-1) was administered at doses of 50 mg/kg. Mortality was monitored daily, and tumor growth was measured at 3-day intervals by caliper measurement. Tumor volume was calculated as follows: tumor volume (mm^3) = (length \times width²)/2. Animal experiments using nude mice were conducted in an aseptic environment, and all experiments were performed according to the guidelines approved by the Committee for the Care and Use of Laboratory Animals at Chonnam National University.

Statistical analysis

Student's *t*-test (SigmaPlot; Systat, San Jose, CA, USA) was used to assess the statistical differences between experimental groups. $P < 0.05$ was taken as being statistically significant.

Results

Synthesis and characterization of PAA-MMA copolymer

PAA-MMA copolymer having various ratios of AA:MMA was synthesized by free radical emulsion polymerization, as described by Yan and Gemeinhart.¹⁰ As shown in Figure 1, the AA:MMA ratio changed from 7:3 to 9:1, and single peaks obtained about 8 minutes of retention time. Average M_w of PAA-MMA is presented in Table 1, and this was about 114,400–136,900.

Preparation and characterization of cisplatin-incorporated nanoparticles

Nanoparticles of PAA-MMA copolymer incorporating cisplatin were prepared via ion-complex formation between

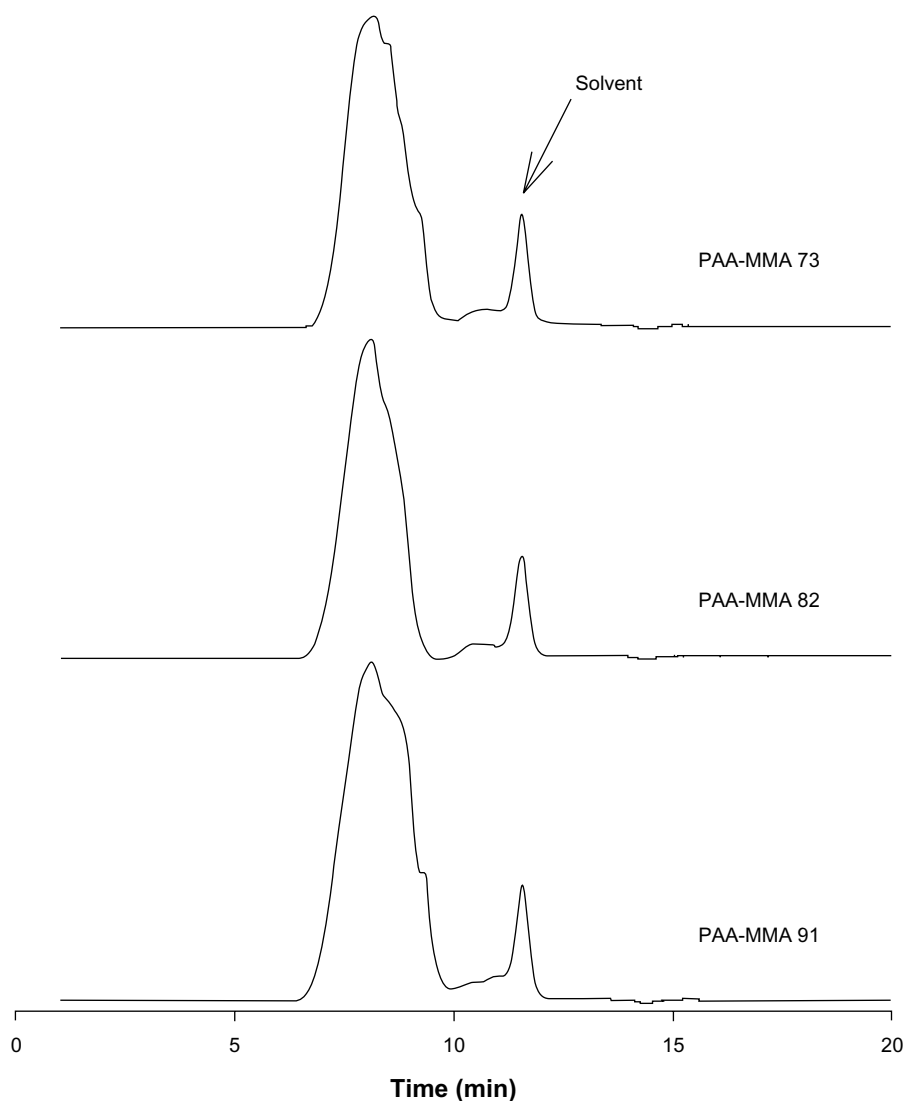


Figure 1 Gel-permeation chromatography profiles of poly(acrylic acid-co-methyl methacrylate) (PAA-MMA) copolymer. Polyethylene glycol as a standard was used for evaluation of molecular weight.

Table 1 Characterization of PAA-MMA copolymers

	Mole ratio of AA:MMA	Molecular weight		PDI
		M_n	M_w	
PAA-MMA-1	7:3	124,200	136,900	1.102 ± 0.130
PAA-MMA-2	8:2	96,530	119,700	1.240 ± 0.101
PAA-MMA-3	9:1	87,000	114,400	1.315 ± 0.167

Abbreviations: M_n , number-average molecular weight; M_w , weight-average molecular weight; PDI, polydispersity index; AA, sodium acrylate.

carboxylic acid of PAA-MMA copolymer and cisplatin, as shown in Figure 2. Cisplatin as a solid form was added to an aqueous solution of PAA-MMA copolymer and stirred for 3 days until a transparent aqueous solution was obtained. Even though cisplatin is not soluble in deionized water, it was completely dissolved in an aqueous solution of PAA-MMA copolymer when cisplatin was complexed with PAA-MMA copolymers and formed nanoparticles, as depicted in Figure 2. This solution was used to identify nanoparticle formation between PAA-MMA and cisplatin using dynamic light scattering and TEM. As shown in Figure 3, spherical particles were observed from this aqueous solution, and their average particle size was around 150 nm (PAA-MMA-2-2 in Table 2). To assess nanoparticle formation, the feeding ratio of cisplatin/PAA-MMA copolymer and series of copolymers was

changed, as shown in Table 2. When the feeding amount of cisplatin was increased, particle size increased from 114 nm to 158 nm, and zeta potential also increased. Drug contents were increased according to the increase of feeding amount of cisplatin and AA content in the copolymer. Higher AA content in the copolymer induced increase in particle size and decrease in zeta potential. Since PAA-MMA copolymer with higher AA content has a higher anionic domain, the zeta potential of nanoparticles should be decreased.

Figure 4 shows cisplatin release from nanoparticles of the PAA-MMA copolymer. As shown in Figure 4A, AA content in the copolymer composition did not significantly change the cisplatin-release rate from PAA-MMA nanoparticles, even though PAA-MMA-3 had a slightly higher release rate than the other formulations. At all formulations, the initial burst effect was observed until 12 hours, and then cisplatin was continuously released from nanoparticles. Otherwise, the higher cisplatin content resulted in a slower release rate than the lower cisplatin content in the nanoparticles, as shown in Figure 4B. For comparison, release of free cisplatin was also studied in vitro. As shown in Figure 4B, most of the free cisplatin was liberated within 24 hours, indicating that nanoparticles have the potential to release cisplatin in a sustained manner.

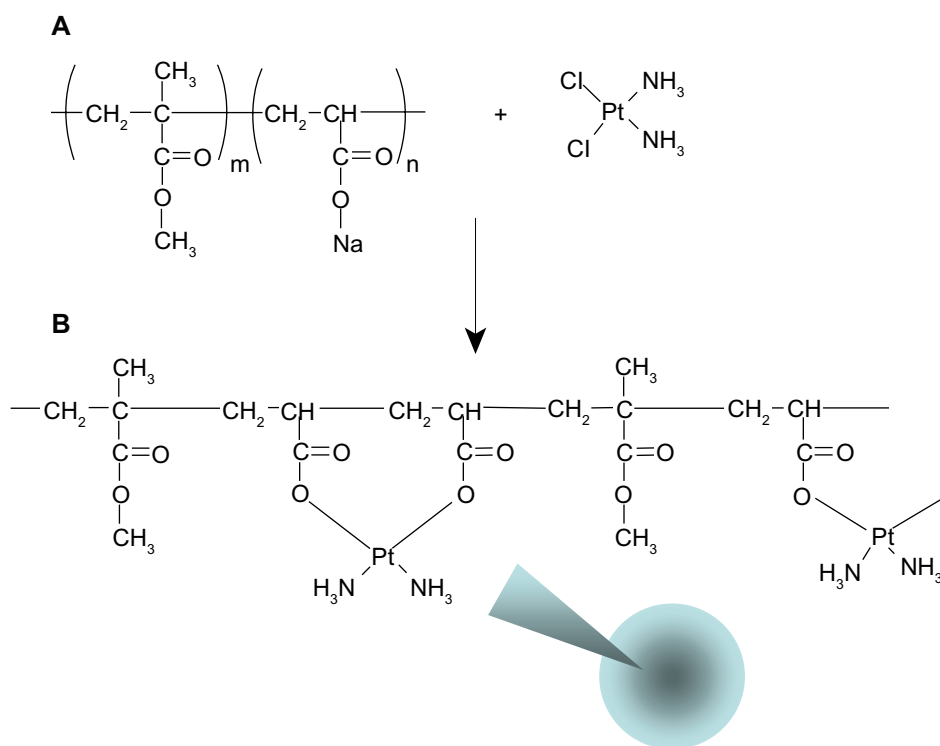


Figure 2 Structure of poly(acrylic acid-co-methyl methacrylate) copolymer (A) and schematic illustration of ion-complex formation between carboxylic acid and cisplatin (B).

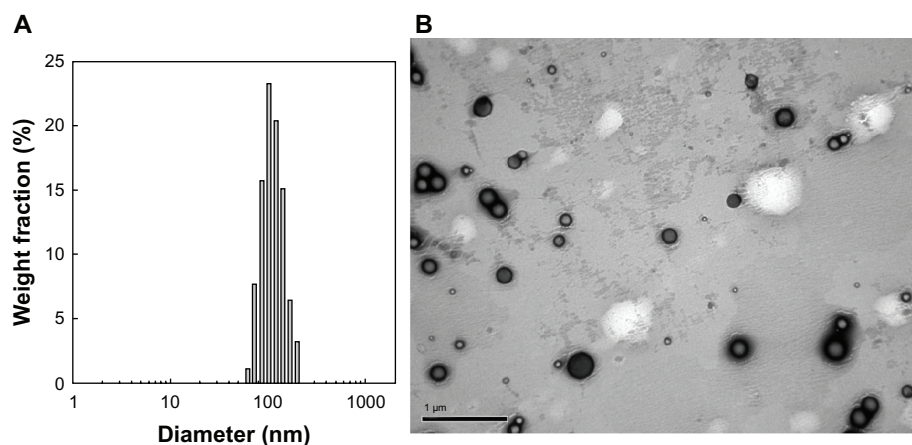


Figure 3 Typical particle size (A) and transmission electron microscopy of cisplatin-incorporated nanoparticles (B) (PAA-MMA-2-2 in Table 2).

Anticancer activity of cisplatin-incorporated nanoparticles

To investigate the antitumor activity of cisplatin-incorporated nanoparticles, free cisplatin, cisplatin-incorporated nanoparticles, and vehicles (PAA-MMA copolymer) were used to treat CT26 tumor cells, and growth inhibition was assessed *in vitro*. As shown in Figure 5, cisplatin showed a dose-dependent growth-inhibitory effect against CT26 cells. Interestingly, nanoparticles incorporating cisplatin showed a practically similar inhibitory effect on the growth of tumor cells, as shown in Figure 5A. Furthermore, the differences between series of PAA-MMA nanoparticles did not significantly change the growth of tumor cells. PAA-MMA copolymer itself did not significantly inhibit growth of tumor cells, as shown in Figure 5B. These results indicated that PAA-MMA nanoparticles incorporating cisplatin have at least similar anticancer activity *in vitro*.

Figures 6 and 7 show apoptosis and necrosis analysis of CT26 cells. Figure 6 shows the effect of cisplatin and cisplatin-incorporated nanoparticles on the apoptosis of CT26 cells. As shown in Figure 6, apoptotic cells were increased dose-dependently with both treatments of cisplatin and cisplatin-incorporated nanoparticles. PAA-MMA

copolymer itself did not significantly affect the apoptosis of tumor cells. Figure 7 shows the necrosis of tumor cells. These results also showed that necrotic cells were increased dose-dependently by both treatments of cisplatin and cisplatin-incorporated nanoparticles, while PAA-PMMA copolymer itself did not affect the necrosis of tumor cells. These results indicated that cisplatin-incorporated nanoparticles induce apoptosis/necrosis of tumor cells as well as free cisplatin.

The antitumor activity of cisplatin-incorporated nanoparticles was assessed *in vivo* using CT26-bearing mice. CT26 cells were implanted in the backs of the mice. About 2 weeks later, cisplatin, cisplatin-incorporated nanoparticles, and vehicles were administered intravenously. As shown in Figure 8, control and vehicle treatment showed a rapid increase of tumor mass, while cisplatin inhibited tumor growth compared to the control treatment. Interestingly, cisplatin-incorporated nanoparticles showed the highest inhibitory effect on the growth of tumor mass. PAA-MMA itself had practically no inhibitory effect on tumor growth. These results indicated that PAA-MMA has promising anticancer activity against an animal tumor xenograft model.

Table 2 Characterization of cisplatin-incorporated nanoparticles

	Polymer: cisplatin ratio (mg/mg)	Drug contents (% w/w)		Particle size (nm)	Zeta potential (mV)
		Theoretical	Experimental		
PAA-MMA-1	100:10	9.1	4.7	149.8 ± 25.4	-5.80
PAA-MMA-2-1	100:5	4.8	2.3	114.3 ± 30.1	-7.84
PAA-MMA-2-2	100:10	9.1	4.9	150.7 ± 22.2	-6.52
PAA-MMA-2-3	100:20	16.7	10.6	158.8 ± 24.3	-4.05
PAA-MMA-3	100:10	9.1	5.3	163.1 ± 28.9	-16.31

Abbreviation: PAA-MMA, poly(acrylic acid-co-methyl methacrylate).

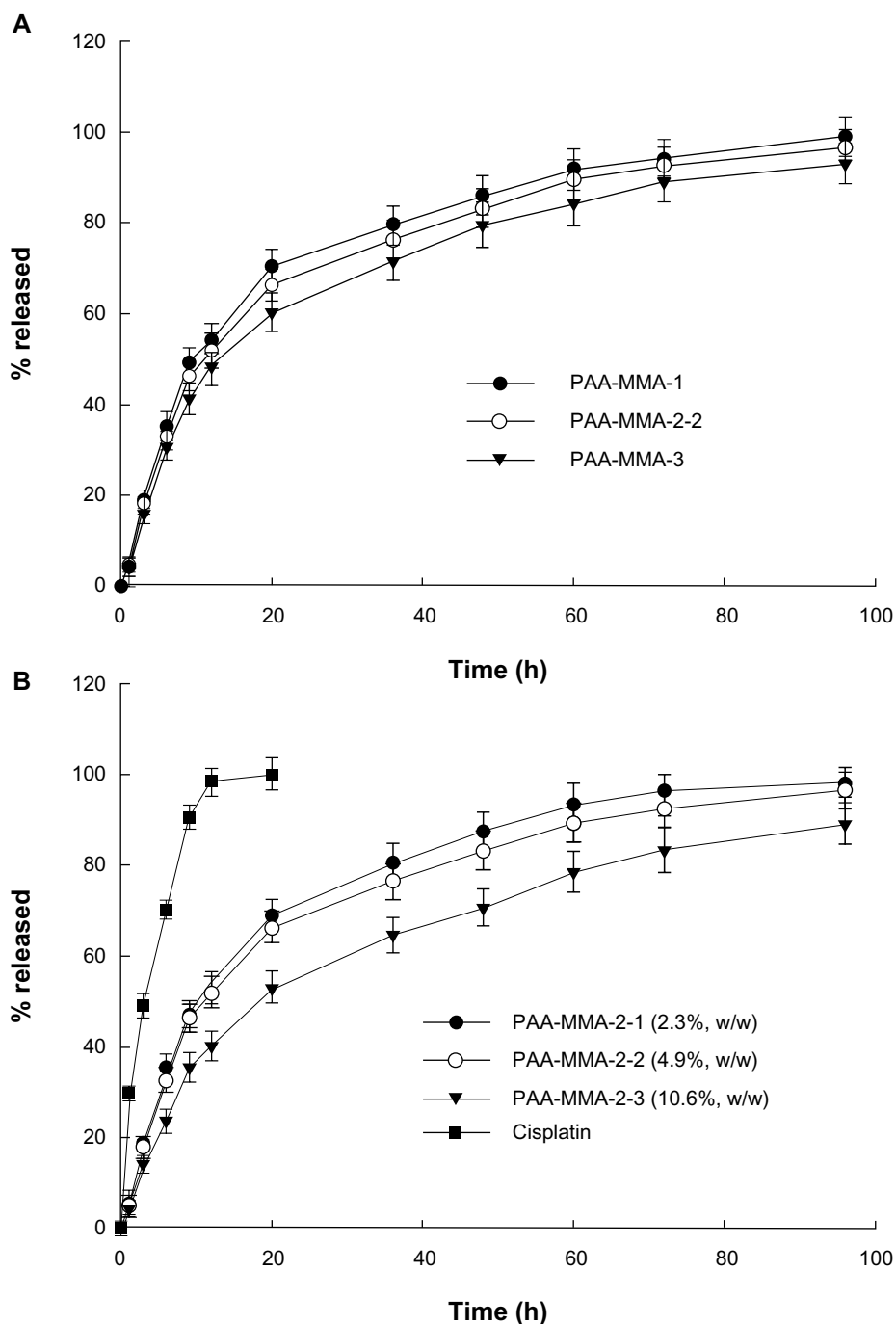


Figure 4 The effect of copolymer ratio (A) and drug contents (B) on the drug release from nanoparticles.
Abbreviation: PAA-MMA, poly(acrylic acid-co-methyl methacrylate).

Discussion

The formation of nanocomplexes between carboxylic acid and cisplatin has been extensively investigated for the last decade.^{4,5,11} Pioneering studies have been performed in the polymeric micelle formation of ion complexes between cisplatin and PEG-poly(aspartic acid) block copolymer.^{4,5,12} Basically, chloride groups of cisplatin were replaced with sodium carboxylate of polymer, and then nanocomplexes

were formed, as shown in Figure 2. They reported that stable micellar solution can be prepared by a simple mixture of cisplatin solid and aqueous polymer solution.⁴ They obtained stable core-shell polymeric micelles incorporating cisplatin because they used block copolymer composed of PEG and poly(aspartic acid), ie, poly(aspartic acid) was complexed with cisplatin and then formed the inner core of the polymeric micelles, while PEG formed the outer shell. In our case, plain

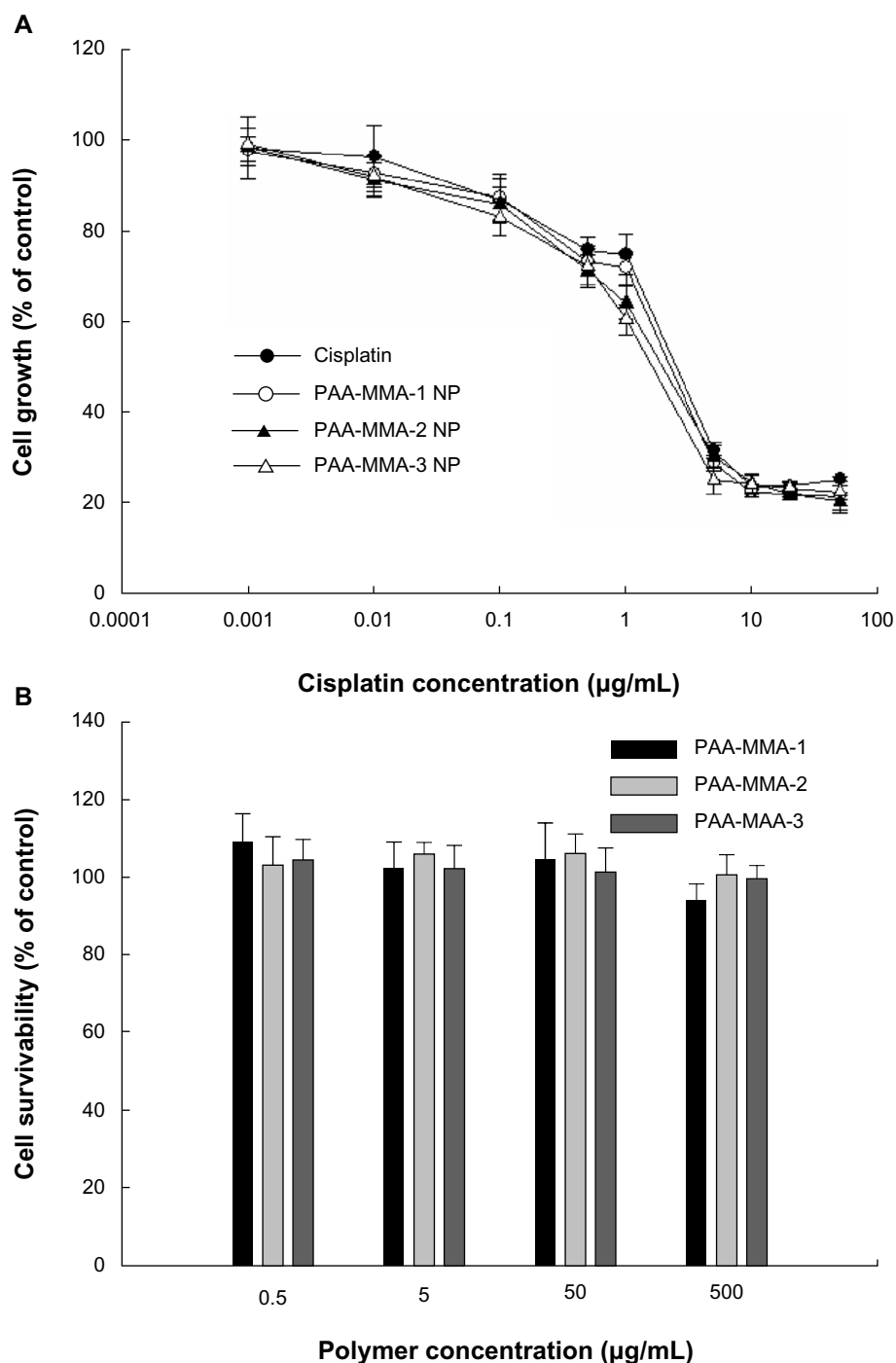


Figure 5 Cytotoxicity of cisplatin or cisplatin-incorporated nanoparticles (NPs) (A) and poly(acrylic acid-co-methyl methacrylate) (PAA-MMA) copolymer (B). Cisplatin, cisplatin-incorporated NPs, and PAA-MMA copolymer were exposed to CT26 cells for 2 days.

nanoparticles incorporating cisplatin were formed because PAA-MMA is a random copolymer.

It has been proved that sustained release formulations of cisplatin such as nanoparticles or polymeric micelles lowered intrinsic toxicity against human body in a Phase I clinical trial.¹² Furthermore, Jeong et al reported that cisplatin and hyaluronic acid (HA) can form nanocomplexes, and cisplatin release can

be controlled by enzymatic degradation of HA.¹¹ They argued that the release rate of cisplatin from nanoparticles became faster in the presence of hyaluronidase and these properties target the modality of cisplatin-HA nanocomplexes.

We synthesized PAA-MMA copolymer with various ratios of AA/MMA to incorporate cisplatin through an ion-complex formation. Aqueous solution of a PAA-MMA

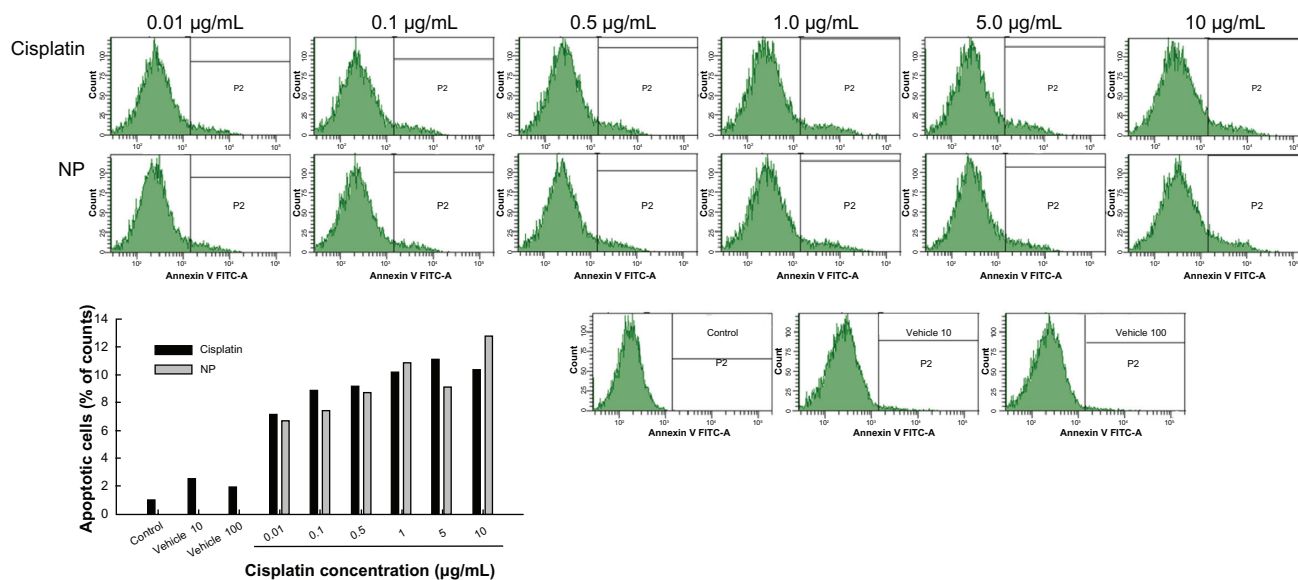


Figure 6 Apoptosis and necrosis analysis of CT26 cells after treatment with cisplatin or cisplatin-incorporated (poly[acrylic acid-co-methyl methacrylate]-2-3) nanoparticles (NPs) for 1 day. Cells were stained with fluorescein isothiocyanate (FITC) annexin V for apoptosis analysis.

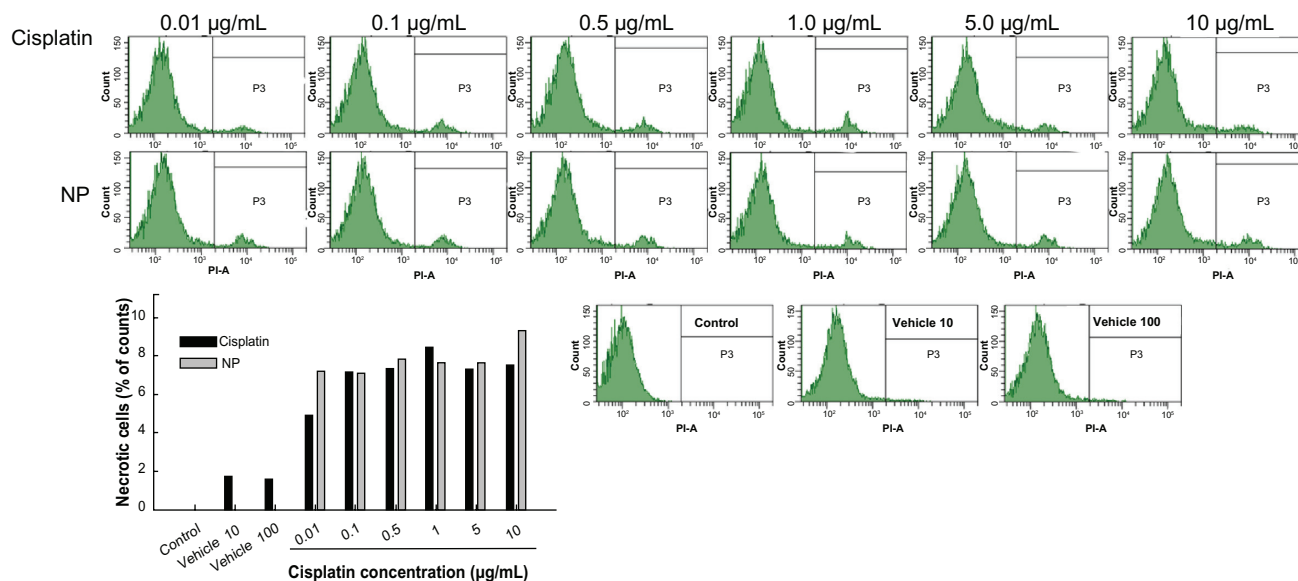


Figure 7 Necrosis analysis of CT26 cells after treatment with cisplatin or cisplatin-incorporated (poly[acrylic acid-co-methyl methacrylate]-2-3) nanoparticles (NPs) for 1 day. Cells were stained with propidium iodide (PI) for necrosis analysis.

copolymer was mixed with cisplatin solid, and nanoparticles were formed by ion complexation between sodium carboxylate of AA and a chloride group of cisplatin. Yan and Gemeinhart reported the formation of submicron particles between PAA-MMA copolymer and cisplatin.¹⁰ Our work was stimulated by their *in vitro* study for the precise characteristics of submicron particles incorporating cisplatin. They described that cisplatin-incorporated microparticles were fabricated using PAA-MMA copolymer, and diameters of cisplatin-incorporated microparticles were in the range of 500 nm–2,300 nm. Furthermore, they described

that PAA-MMA microparticles incorporating cisplatin have significantly lower cytotoxicity against NIH/3T3 cells and lower acute toxicity against small animals (female rats) than cisplatin. However, they did not describe antitumor activity of cisplatin-incorporated microparticles using tumor cells. Therefore, we focused on the antitumor activity of cisplatin-incorporated nanoparticles of PAA-MMA copolymer. We obtained significantly smaller particles incorporating cisplatin than Yan and Gemeinhart's study,¹⁰ ie, cisplatin-incorporated nanoparticles of PAA-MMA copolymer in this study had smaller average particle size – less than 170 nm.

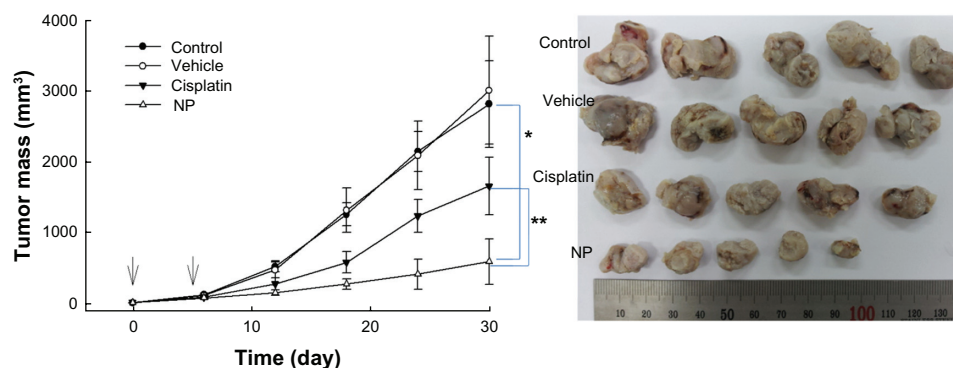


Figure 8 Changes of tumor mass after administration of cisplatin or cisplatin-incorporated (poly[acrylic acid-co-methyl methacrylate]-2-3) nanoparticles (NPs). CT26 cells were implanted into the backs of mice, and drug was injected after 2 weeks of tumor-cell implantation. The time for first drug injection was determined as day 0. Drug injection was done twice at an interval of 4 days. Phosphate-buffered saline was injected into control mice. Cisplatin or NPs (dose: 5 mg/kg) were administered via the tail vein. The vehicle was administered with poly(acrylic acid-co-methyl methacrylate)-2 (50 mg/kg) into the mice. Five mice were used for each group.

Note: *** $P < 0.001$.

Nanoparticles smaller than 200 nm are believed to be an appropriate vehicle for prevention of reticuloendothelial system uptake and for passive targeting to specific organs.^{13–15} Therefore, the cisplatin-incorporated nanoparticles in this study may be more suitable for antitumor drug targeting than the microparticles of Yan and Gemeinhart's study.¹⁰ The higher AA content in the copolymer composition may have higher potential to form a complex with cisplatin. We were able to obtain stable nanoparticles at a polymer:cisplatin weight ratio of less than 100:20 mg, as shown in Table 2. Significant aggregation or precipitants were observed when cisplatin feeding weight was higher than 30 mg. Sustained release of cisplatin from nanoparticles was observed in in vitro drug-release studies while free cisplatin was rapidly liberated from dialysis membrane as shown in Figure 4, indicating that PAA-MMA nanoparticles are promising vehicles as sustained release formulations.

PAA-MMA nanoparticles showed similar anticancer activity against CT26 tumor cells in vitro, as shown in Figures 5–7, while PAA-MMA copolymer did not solely affect the growth of tumor cells. Furthermore, cisplatin-incorporated nanoparticles induced apoptosis/necrosis of tumor cells as well as free cisplatin. PAA-MMA copolymer did not affect the apoptosis/necrosis process of tumor cells at equivalent doses of nanoparticle treatment. In contrast to the in vitro study, cisplatin-incorporated nanoparticles showed improved antitumor activity in vivo using animal tumor-bearing mice. These results might be due to the fact that sustained-release properties of nanoparticles render longer half-lives of cisplatin in blood, and then cisplatin concentration must be preferentially increased in the tumor tissue.^{5,12} Furthermore, enhanced permeation and retention of nanoparticles against solid tumors may

contribute to these results.^{13,14} Liposomal carriers for cisplatin were reported to have potential in reduction of acute toxicity, increase in half-lives, and improvement of antitumor effects of cisplatin.^{7,16} In particular, cisplatin-incorporated polymeric micelles with intravenous administration were demonstrated to have higher and longer serum levels than free cisplatin.¹⁷ Thus, treatment with polymeric micelle or colloidal carriers against solid-tumor bearing mice can reduce nephrotoxicity with promising anticancer activity.^{4,5,12,17} Furthermore, many reports have described the relationship between improved anticancer activity and longer blood-circulation time of colloidal carriers with small particle sizes.^{5–7,12–17} Since PAA-MMA nanoparticles in the present study also had small particle sizes and sustainably released cisplatin to the aqueous phase, these nanoparticles might also have longer half-lives in biological systems, inducing improved anticancer activity.

In conclusion, cisplatin-incorporated nanoparticles were fabricated using PAA-MMA copolymer through an ion-complex formation, and the antitumor activity of these nanoparticles was assessed against CT26 colon carcinoma cells in vitro and in vivo. PAA-MMA nanoparticles incorporating cisplatin were spherical, less than 200 nm in size, and showed sustained-release properties. The improved antitumor activity of cisplatin-incorporated nanoparticles was observed with an animal tumor xenograft model. We suggest that cisplatin-incorporated nanoparticles of PAA-MMA copolymer are a promising candidate for antitumor drug delivery.

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Disclosure

The authors report no conflicts of interest in this work.

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