

Some inferences from in vivo experiments with metal and metal oxide nanoparticles: the pulmonary phagocytosis response, subchronic systemic toxicity and genotoxicity, regulatory proposals, searching for bioprotectors (a self-overview)

Boris A Katsnelson¹
Larisa I Privalova¹
Marina P Sutunkova¹
Vladimir B Gurchich¹
Nadezhda V Loginova¹
Ilzira A Minigalieva¹
Ekaterina P Kireyeva¹
Vladimir Y Shur²
Ekaterina V Shishkina²
Ya B Beikin³
Oleg H Makeyev⁴
Irene E Valamina⁴

¹The Medical Research Center for Prophylaxis and Health Protection in Industrial Workers, Ekaterinburg, Russia; ²The Institute of Natural Sciences, The Ural Federal University, Ekaterinburg, Russia; ³The City Clinical Diagnostics Centre, Ekaterinburg, Russia; ⁴The Ural State Medical University, Ekaterinburg, Russia

Correspondence: Boris A Katsnelson
The Medical Research Center for Prophylaxis and Health Protection in Industrial Workers, 30 Popov Street, Ekaterinburg 620014, Russia
Tel +7 343 253 0421
Fax +7 343 371 7740
Email bkatsnelson@etel.ru

Abstract: The purpose of this paper is to overview and summarize previously published results of our experiments on white rats exposed to either a single intratracheal instillation or repeated intraperitoneal injections of silver, gold, iron oxide, copper oxide, nickel oxide, and manganese oxide nanoparticles (NPs) in stable water suspensions without any chemical additives. Based on these results and some corroborating data of other researchers we maintain that these NPs are much more noxious on both cellular and systemic levels as compared with their 1 µm or even submicron counterparts. However, within the nanometer range the dependence of systemic toxicity on particle size is intricate and non-unique due to complex and often contra-directional relationships between the intrinsic biological aggressiveness of the specific NPs, on the one hand, and complex mechanisms that control their biokinetics, on the other. Our data testify to the high activity of the pulmonary phagocytosis of NPs deposited in airways. This fact suggests that safe levels of exposure to airborne NPs are possible in principle. However, there are no reliable foundations for establishing different permissible exposure levels for particles of different size within the nanometric range. For workroom air, such permissible exposure levels of metallic NP can be proposed at this stage, even if tentatively, based on a sufficiently conservative approach of decreasing approximately tenfold the exposure limits officially established for respective micro-scale industrial aerosols. It was shown that against the background of adequately composed combinations of some bioactive agents (comprising pectin, multivitamin-multimineral preparations, some amino acids, and omega-3 polyunsaturated fatty acid) the systemic toxicity and even genotoxicity of metallic NPs could be markedly attenuated. Therefore we believe that, along with decreasing NP-exposures, enhancing organisms' resistance to their adverse action with the help of such bioprotectors can prove an efficient auxiliary tool of health risk management in occupations connected with them.

Keywords: nanoparticles, metals, metal oxides, toxicity

Introduction

Nanoparticles (NPs) of metals and of their oxides are of special interest in the light of health risks' assessment and management because, along with engineered NPs, there exists usually a substantial fraction of nanoscale ("ultrafine") particles of the same substances within the particle size distribution of condensation aerosols generated by arc-welding and metallurgical technologies. However in such aerosols chemically

similar micrometer particles (MPs), including submicron ones having dimensions >100 nm, are usually present as well.

Thus, for metallic NPs more urgently than for many others, some theoretical problems of nanotoxicology in everyday practice needs solving:

1. Are NPs recognized by organisms' main defenses worse than or just as well as respective MPs are?
2. Are they much more toxic as compared with MPs on both cellular and systemic levels?
3. Is there a definite dependence of defense and adverse responses to NPs on their dimensions within the conventional nanometer range and/or on their chemical nature?
4. The last but not the least, is it possible to protect health of people dealing with either engineered or "spontaneous" NPs a) by establishing low but still practicable permissible exposure levels (PELs) and b) by enhancing organisms' resistance to NPs toxicity (genotoxicity included)?

During 2009–2014 our experimental work was focused on trying to find answers to these important questions.^{1–10}

Although in the vast nanotoxicological literature of the last decade studies concerned with the assessment of metal or metal-oxide NP toxicity are not at the top of the list, such publications are nevertheless quite numerous. To illustrate this, we may refer to several works devoted to NPs of the metals that were the subject-matter of our own studies considered below: iron oxide,^{11–20} silver,^{21–45} gold,^{24,45–58} copper and copper oxide,^{43,59–68} nickel oxide,^{65,69–73} and manganese oxides.^{74–76} A comprehensive critical review of relevant publications is out of this paper's scope. It should be stressed, however, that the prevailing majority of published researches assessed the cytotoxicity and genotoxicity of NPs on stable cell lines and only rarely on laboratory animals. No doubt, *in vitro* experiments

feature a number of well-known advantages, in particular, relating to analysis of primary mechanisms of toxicity. At the same time, any extrapolation of the results of these experiments to the organism level is associated with a number of uncertainties and assumptions. Moreover, some important aspects (in particular, toxicokinetics, dose-systemic response relationships, the functioning and efficiency of protective mechanisms) can generally be addressed only through experiments on mammals' whole organism.

Materials and methods

Here we are giving but a brief overview of methodology used in our experiments which were described in detail in original articles.^{1–10}

We experimented with NPs of iron oxide Fe_3O_4 (magnetite) having mean diameters 10 nm, 50 nm or 1 μm , gold (4 nm or 50 nm), silver (4 nm, 49 nm or 1.1 μm), copper oxide (20 nm or 340 nm), nickel oxide NiO (30 nm), and manganese oxide Mn_3O_4 (32 nm). Keeping in mind the above mentioned theoretical premises of our research, we used purposefully prepared and accurately characterized NPs of pure metals or their oxides suspended in deionized water rather than commercial nanomaterials. Only iron oxide magnetite NPs were synthesized chemically^{1–3} while in all other cases we used a technique of laser ablation of a 99.99% pure metal target in water followed by laser fragmentation for preventing particle aggregation. The main advantages of this technique are as follows: a) it provides nano-suspensions with a sufficiently narrow particle size distribution (examples are shown in Figures 1 and 2) and b) these suspensions are highly stable as a rule, maintaining their characteristics without any noticeable particle aggregation over periods sufficient for carrying out experiments described below. In no experiment did we add any chemical stabilizer to the suspension,

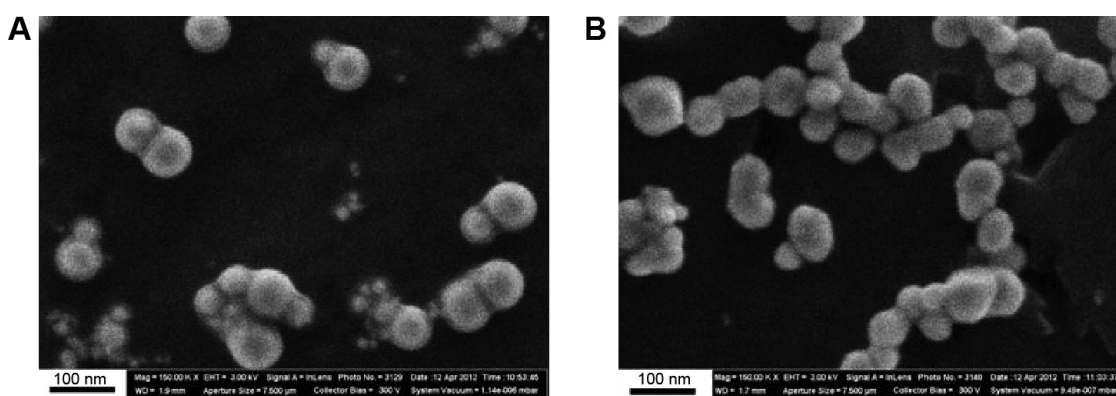


Figure 1 Images of nanoparticles in suspension obtained by scanning electron microscopy at $\times 150,000$.

Notes: (A) Nanogold; (B) nanosilver.

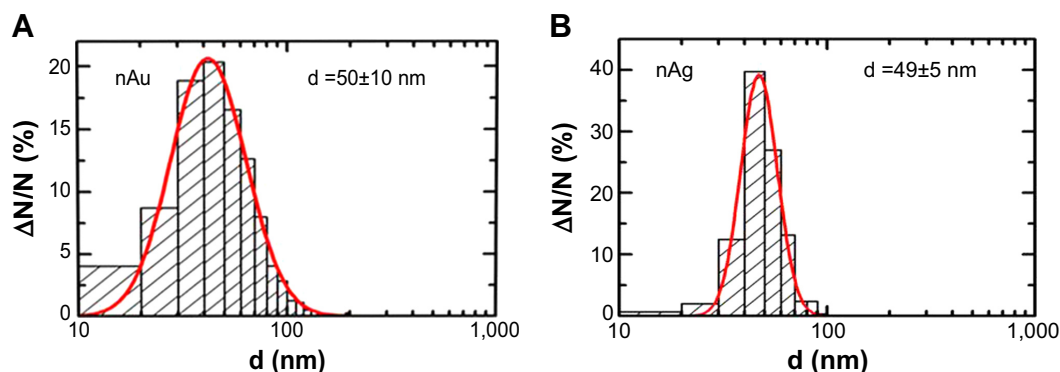


Figure 2 Nanoparticle (NP) size distribution function.

Notes: The results of statistical processing of 800 images of gold NPs (A) and 650 images of silver NPs (B) obtained by scanning electron microscopy.

Abbreviations: d, diameters of nanoparticles; N/N, number of nanoparticles.

and only in experiments with magnetite and NiO NPs did we have to sonicate the suspensions just before instillations or injections.

All experiments were carried out on outbred white female rats from our own breeding colony with an initial body weight of 150 to 220 g, with a minimum of 12 animals in different exposed and control groups. Rats were housed in conventional conditions, breathed unfiltered air, and were fed standard balanced food. The experiments were planned and implemented in accordance with the “International Guiding Principles for Biomedical Research Involving Animals” developed by the Council for International Organizations of Medical Sciences (1985) and approved by the Committee on Ethics of the Ekaterinburg Medical Research Center for Prophylaxis and Health Protection in Industrial Workers.

A single intratracheal (IT) instillation of 1 mL of NP or MP water suspension (or of the same deionized water without any particles to controls) served as an experimental model for the response of the low airways to particle deposition. It is well-known that important qualitative and quantitative patterns of the response of the pulmonary free cell population (in particular, its dependence on the cytotoxicity of deposited particles) observed in inhalation exposures to dust particles are principally the same in the case of their IT administration.⁷⁷ At the same time, IT model provided cellular material for studying the phagocytizing activity of pulmonary macrophages and polymorphonuclear leukocytes, as well as intracellular localization of NPs engulfed by them and ultrastructural damage caused to the cell by those NPs. The results thus obtained are comparable to the data obtained by other researchers in experiments on cell cultures but give a valuable addition to the latter because in vivo interaction between cells and particles occurs in a microenvironment which is not reproducible by artificial cell culture media.

A cell population of bronchoalveolar lavage fluid (BALF) obtained 24 hours after IT instillation to rats of NP or MP suspensions was studied with optical microscopy, transmission electron microscopy (TEM), and semi-contact atomic force microscopy (sc-AFM).

Subchronic toxicity was modeled by means of repeated intraperitoneal (IP) injections of the same particles during 5–7 weeks. Using IP injections for modeling systemic intoxication, which in real conditions can be induced by long-term inhalation exposure of workers needs some justification. It is well-known that “NPs deposit with high efficiency in the entire respiratory tract, from the head airways to the alveoli, due to diffusion”.⁷⁸ For instance, the widely used Human Respiratory Tract Model of the International Commission of Radiological Protection^{79,80} predicts 100% total deposition of 0.001 μm (ie, 1 nm) and ~90% for 0.01 μm (ie, 10 nm) particles for a normal adult mouth breathing male human subject. However there are many anatomical, functional, and aerodynamic differences between humans and rodents which make one assume interspecies distinctions in regional particle deposition and thus in the kinetics of their elimination from the airways to the gastrointestinal tract and/or absorption. No wonder that the authors of a comprehensive review of nanotoxicological assessment techniques⁸¹ maintain that:

Rodents, the commonly used species for toxicology testing, are obligatory nose breathers and, therefore, not representative models for human respiratory inhalation exposure.

In other words, NP inhalation by laboratory rodents is not as ideal a model of real human exposure as it is often deemed to be.

The IP model permits one to circumvent these interspecies differences of inhaled NP deposition and is adequate enough when one wants to look into body distribution and elimination of NPs, and for organisms’ reactions to NPs after

they have penetrated into the blood from a primary deposit. Like any model (always a necessary simplification of a complicated system deliberately omitting some sub-systems and some material or informational flows and feedbacks), it has both drawbacks and virtues. Among the latter, one should take into consideration that dosing by injection is much more accurate and reliable as compared with the more “natural” experimental methods. This reason is crucial for experiments of comparative design like ours. IP modeling of subchronic intoxications is well-known and recognized in general experimental toxicology. Moreover, it was used just in experimental nano-toxicological studies also by other researchers.^{82,83}

In our experiments, systemic toxic effects of particles under comparison were characterized by a lot of biochemical and functional tests, typically comprising:

- weighing of the body and inner organs;
- estimation of the central nervous system ability for the temporal summation of sub-threshold impulses (a variant of withdrawal reflex and its facilitation by repeated electrical stimulations in intact, conscious rat);⁸⁴
- recording of the number of head-dips into the holes of a hole-board, which is frequently used for studying behavioral effects of toxicants and drugs;^{85,86}
- 24 hours collection of urine for measuring its density, urine output, coproporphyrin, delta-aminolevulinic acid (δ -ALA), and creatinine contents;
- sampling of capillary blood from a notch on the tail for examining the hemogram, hemoglobin content, and for cytochemical determination of SDH activity in lymphocytes (by the reduction of nitroterazolium violet to formazan);
- collecting blood by exsanguination from rats killed by decapitation for measuring concentrations of reduced glutathione, total serum protein, albumin, globulin, bilirubin, ceruloplasmin, malondialdehyde, ALP, ALT, AST, CAT, GGT, SH-groups, and creatinine.

The metal content of the liver, spleen, kidneys, and brain was determined with the help of atomic emission spectrometry and, for iron oxide particles, also by the electron paramagnetic resonance method.

Thin sections of different organs were prepared for histological examination by hematoxylin and eosin stain and, where necessary, periodic acid Schiff, Nissl and Perl's stains. We used a planimetric ocular grid for morphometry of spleen and liver and an image recognition programmed system for that of kidneys and brain.

In the subchronic experiments with NPs and MPs of Ag, Au, and CuO the genomic DNA fragmentation was

assessed in cells of several tissues using the random amplified polymorphic DNA test.

Both systemic and genotoxicity of nAg and of nCuO were assessed also against the background of oral administration of bio-protective complexes (BPC) comprising pectin, multivitamin–multimineral preparations, some amino acids, and omega-3 polyunsaturated fatty acid.^{7,8} Pre-medication with a similar BPC (without pectin) was administered to rats instilled afterward intratracheally with NiO+Mn₃O₄ NPs.¹⁰

Main results and discussion

We propose to summarize and discuss our results in light of the questions suggested in the Introduction.

Are NPs recognized and dealt with by the organism worse than, or just as well as or even better than respective microparticles are?

Practical implications of this seemingly theoretical question hardly needs explanation. Indeed, if physiological defenses against NPs are of a low effectiveness one should assume that non-harmful exposures to NPs would be hardly possible, in principle. In other words, no safe exposure levels for NPs in the environment should be established.

This problem is not contrived by us. In the beginning of the “nanotoxicological era”, it was maintained by some authors (for instance Donaldson et al and Oberdörster et al)^{87,88} that the physiological protective mechanisms enabling animals and humans to exist in an atmosphere unavoidably polluted with suspended particles of a wide range of sizes and chemical compositions have, for whatever reasons, little effectiveness in relation to NPs. It was thought, specifically, that NPs deposited in the pulmonary region are not efficiently phagocytized by alveolar macrophages (AM) due either to their inability to recognize the smallest particles or to the failure of the latter to generate a chemotactic signal at the site of their deposition. For instance, 10 years ago a highly reputable group of experts⁸⁸ maintained that “very small particles [...] may not be detected by the normal phagocytic defenses”. Even as recently as in 2012 the author of a comprehensive review⁸⁹ did not present this problem as solved:

Macrophages engulf microbes and apoptotic debris, but the question is: are NPs recognized by phagocytes or do such particles fly under the radar and escape immune recognition?

We started off our own research into this field^{1,2} with some criticism of the research data seemingly corroborating those views and with skepticism based on evolutionary considerations. We stressed that terrestrial vertebrates have had to inhale ultrafine nanoscale particles (such as volcanic ash, airborne particles of dispersed seawater, forest fire smoke, sulfates generated in the atmosphere as a result of sulfur dioxide oxidation) for as long as they have inhaled fine microparticles. We reminded that both main mechanisms of pulmonary clearance (engulfing of particles by phagocytes and mucociliary transport) had been present already in amphibians,⁹⁰ that is even before the morphological structuring of the lungs was complete. We maintained that it would be difficult to understand why these ancient defenses, highly efficient in relation to MPs, should have been selected and fixed by the evolution had they been ineffective in relation to NPs against which defense is even more necessary since they are presumably more noxious (see below).

From the very first experiment, however, our data suggested that organisms are not defenseless when confronted with metallic NPs' exposures. In particular, we have found that just the pulmonary phagocytosis response to deposition of NPs is quite potent. In several experiments it was found that, given equal mass doses and identical metal composition, NPs induced intensive recruitment of phagocytes increasing cell count in the BALF to a greater degree than MPs did. One example illustrating this is given in Table 1. Moreover, this recruitment and especially an increase in neutrophil leukocytes (NL) to AM ratio in the BALF cell count is the more pronounced, the smaller is the particle within the nanoscale range as illustrated by Figure 3.

Such NLs' recruitment toward the lower airways in response to the deposition of particles is quite often described as "inflammation" and, thus, as a pathological phenomenon rather than a pulmonary defense mechanism. This concept is common to nanotoxicological studies as well.^{73,91-95} We maintain, however, that it can be misleading. Beyond any doubt, enhanced recruitment of NLs onto the free surface of lower

airways is typical of acute and, to a lesser degree, chronic inflammatory processes induced by microbial or chemical agents. However, a certain number of NLs is always present in the BALF of healthy animals, at least when they are constantly inhaling unfiltered air. They are hardly always living with chronic inflammation of the respiratory system. Meantime, it is long ago that we⁹⁶⁻¹⁰⁰ found fairly strong reasons for considering the NL recruitment response to be an important mechanism of partial compensation for the damage caused by cytotoxic micrometric particles to AMs, the main effector of the pulmonary clearance. A mathematical multi-compartmental model of pulmonary region clearance which comprised just this compensatory mechanism simulated very well the pulmonary retention of dusts of varying degrees of cytotoxicity (quartzite rock, titanium dioxide, standard quartz DQ12) under long-term inhalation exposure as well as a decrease in this retention under the effect of such potent protectors of the macrophage against particle cytotoxicity as glutamate.

In the same way, data obtained in our experiments with metallic NPs demonstrate that in this case too the enhanced NL recruitment, be it a manifestation of the inflammation or a normal physiological pulmonary particle clearance mechanism, takes an important part in this cleaning process. It should be stressed that both types of recruited phagocytes (that is not only AMs but NLs as well) engulf NPs much more avidly compared with MPs of the same metal, and the smaller the diameter of particles, the more active their phagocytosis by these cells. We illustrate these relationships with an example of magnetite particles (Figure 4).

sc-AFM reveals multiple "pits" on the surfaces of both AMs and NLs, and the greater the particles, the greater the dimensions of these pits (examples are shown in Figures 5 and 6). At the same time, the smaller the particle, the higher the surface concentration of these pits (Figure 7). We consider them to be not mere "footprints" of particles passively penetrating into cells through the cell membrane and leaving a hole but a fixed moment of its invagination at the first stage of active endocytosis. Indeed, although the mean diameter

Table 1 Number of cells in the bronchoalveolar lavage fluid (BALF) 24 hours after the intratracheal instillation of silver nano- or microparticles to rats at a dose of 0.2 mg per rat ($\bar{x} \pm se$)

Substance administered	Number of cells $\times 10^6$			NL/AM
	Total	Neutrophil leukocytes (NL)	Alveolar macrophages (AM)	
Nanosilver (49 nm)	4.25 \pm 0.77 [°]	2.99 \pm 0.71 ^{*°}	1.16 \pm 0.14	2.47 \pm 0.33 ^{*°}
Microsilver (1.1 μ m)	1.99 \pm 0.25	0.73 \pm 0.15 [*]	1.24 \pm 0.19	0.66 \pm 0.13 [*]
Water (controls)	1.41 \pm 0.33	0.13 \pm 0.04	0.89 \pm 0.18	0.14 \pm 0.023

Notes: Statistically significant difference ^{*}from control group; [°]from microsilver group ($P < 0.05$ by Student's *t*-test).

Abbreviation: se, standard error.

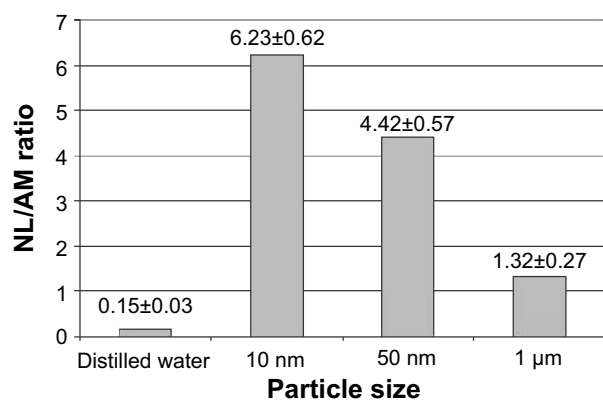


Figure 3 The ratio of the number of neutrophil leukocytes (NL) to the number of alveolar macrophages (AM) in the bronchoalveolar lavage fluid of rats 24 hours after the instillation of magnetite particles of different sizes at a dose of 2 mg in 1 mL of distilled water ($x\pm S_x$).

of these pits correlates with that of IT instilled NPs, the former is not equal to but usually greater than the latter. For instance, under exposure to 49 nm silver particles and 50 nm gold particles the average diameter of pits was 75.2 ± 0.3 nm and 77.6 ± 1.5 nm, respectively.⁷

The TEM images (Figure 8) also testify to the physiological endocytosis of NPs as opposed to their direct penetration through cell membrane by diffusion, which, above any doubts, is also possible.⁴⁻⁶

When NPs are of given dimensions, the pulmonary phagocytosis response strongly depends on their chemical nature. For instance, as can be seen from Table 2, the total cell, NL and AM counts, and the macrophage count of the BALF increased under the impact of silver NPs more than it did under the impact of virtually equidimensional gold NPs. A similar dependence of NL recruitment on the chemical nature of NPs was shown in a comparative experiment with nano-suspensions of NiO and Mn_3O_4 .¹⁰

Judging by the sc-AFM images, nanosilver was engulfed by phagocytic cells twice more avidly than nanogold since the average number of surface “pits” per square micrometer was 18.0 and 9.0, respectively.⁷ Under TEM it was evident that intracellular distribution of the internalized silver and gold NPs was also different, the most significant being a more pronounced affinity of nAg for mitochondria and a lesser ability to penetrate into cell nuclei as compared with nAu (Figures 9 and 10).⁷

In another experiment we compared responses to IT instillation of CuO NPs (20 nm) and submicron particles (340 nm) and again found that NPs evoked more significant recruitment of NLs than even these fine MPs did ($9.41\pm 2.01\times 10^6$ and $3.64\pm 0.90\times 10^6$ respectively, the control value being $0.05\pm 0.01\times 10^6$). The respective NL/AM ratio

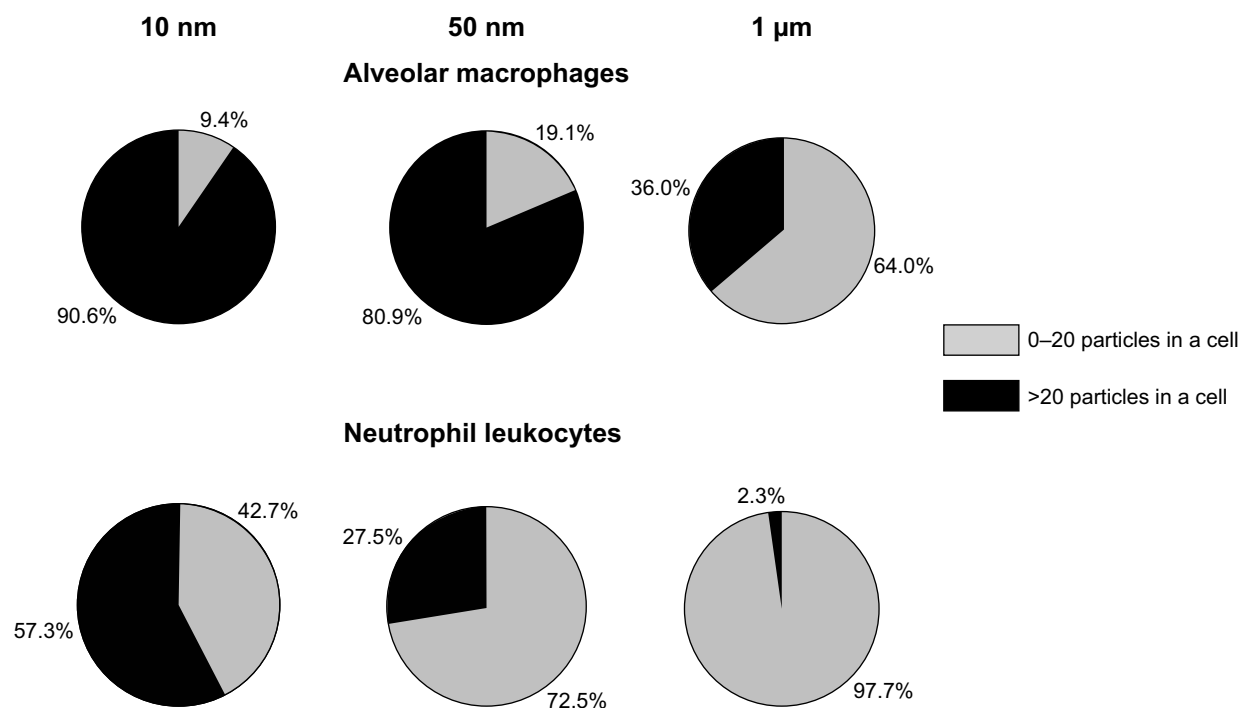


Figure 4 Percentage of phagocytic cells with different levels of particle burden in the bronchoalveolar lavage fluid of rats 24 hours after the intratracheal instillation of Fe_3O_4 particles having different diameters.

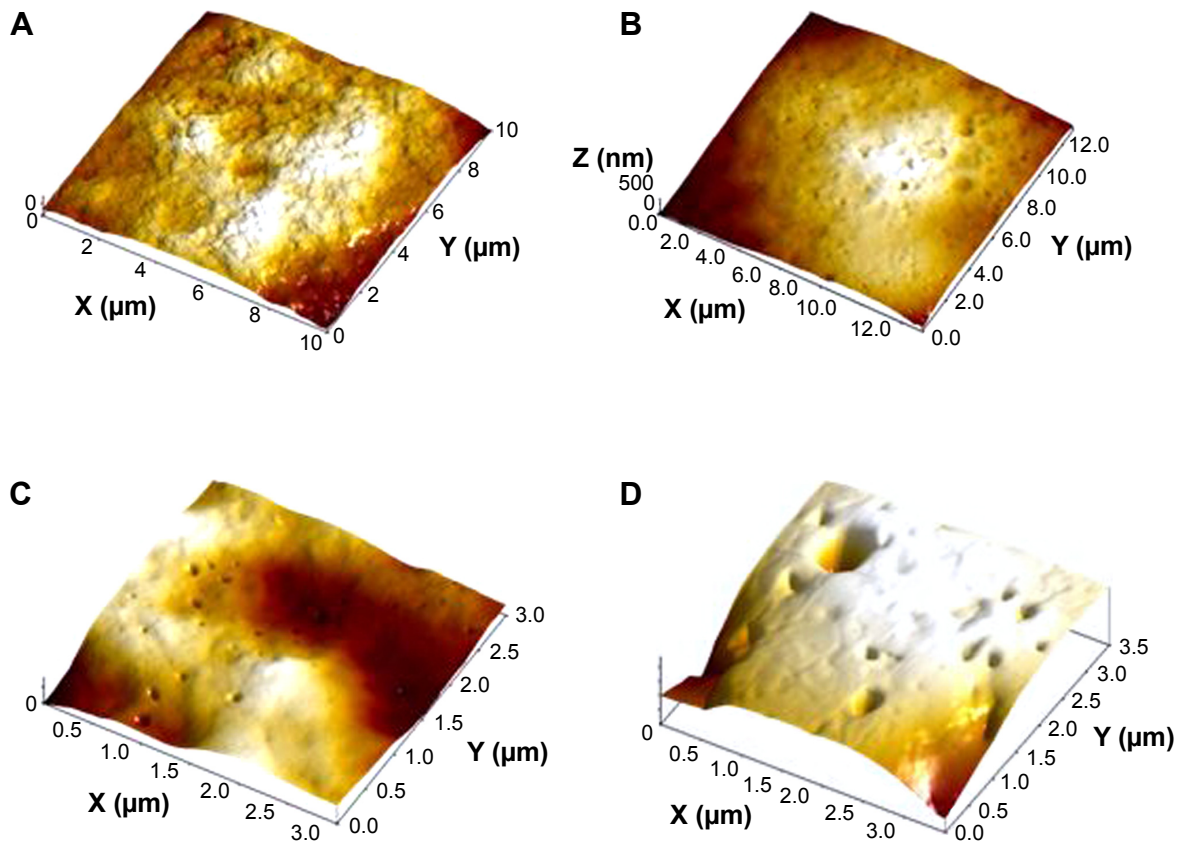


Figure 5 Alveolar macrophage surface topography measured by semi-contact atomic force microscopy.

Notes: (A) Control; (B) after instillation of 10 nm magnetite; (C) after instillation of 50 nm magnetite; (D) after instillation of 1 μm magnetite particles.

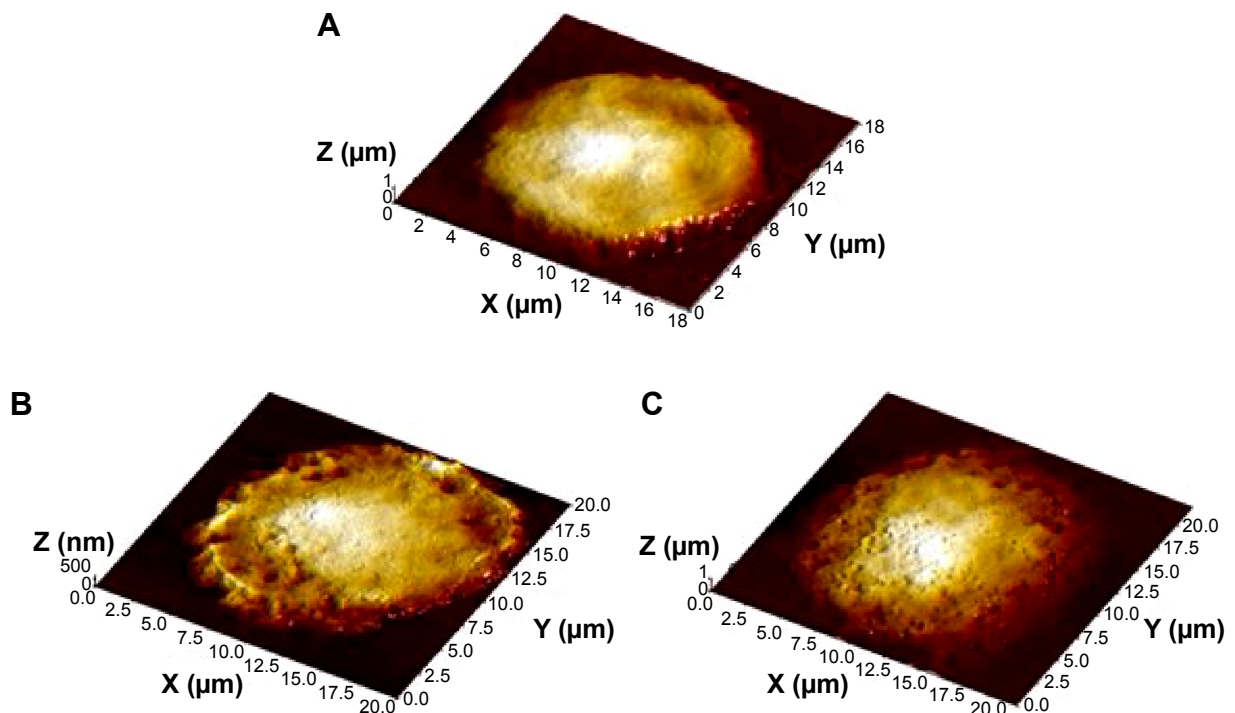


Figure 6 Alveolar macrophage surface topography measured by semi-contact atomic force microscopy.

Notes: (A) Controls; (B) after instillation of 340 nm copper oxide; (C) after instillation of 20 nm copper oxide particles.

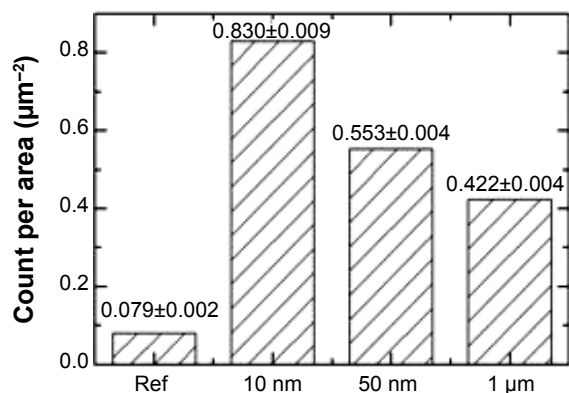


Figure 7 Average ($x\pm S_x$) surface concentration of pits of all transverse dimensions detected on the surfaces of cells of each group of rats administered magnetite particles of different diameters and of control rats (Ref).

values were 4.76 ± 1.39 , 1.39 ± 0.16 , and 0.06 ± 0.01 . sc-AFM again revealed pits on the cell surface (Figure 6) with average diameters 25 ± 1 nm under exposure to NPs and 290 ± 14 nm under exposure to MPs, the average surface density of these pits being 32.6 and 1.9 per μm^2 , respectively. A hypothetical explanation of positive rank correlations between NL/AM ratio and phagocytosis activity as judged by sc-AFM will be discussed in the next section.

To sum up the data discussed in this sub-section, it seems evident that:

- the pulmonary phagocytosis response to metallic NPs' deposition (namely, phagocytes recruitment toward the free surface of lower airways and phagocytic activity of these cells) is more pronounced than to the deposition of respective MPs,
- this response depends on both NP dimensions and their chemical characteristics.

Most importantly, we found that pulmonary tissue was indeed liberated from deposited NPs quicker than from their

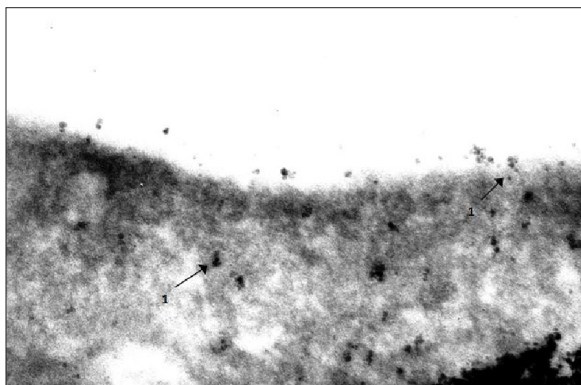


Figure 8 Engulfment of 10 nm magnetite particles by an alveolar macrophage. **Notes:** Transmission electron microscopy, magnification $\times 140,000$. Phagosomes are shown with arrows 1.

micrometric counterparts, the smaller the NPs, the quicker the process.^{1,3} We believe that this is due not only to quicker dissolution but also to more active phagocytosis of the smallest NPs. However, both physicochemical and physiological mechanisms of NPs pulmonary clearance, while attenuating the harmful impact of NPs on lungs, serve as prerequisites for the toxic action of NPs on other organs either through the gastrointestinal tract (where they are translocated with the normal mucociliary transport) or directly through blood (as a result of their dissolution/absorption or of penetration/diffusion). Thus we should discuss the further two crucial questions posed in the Introduction.

Are metallic NPs really much more toxic as compared with microparticles of the same chemical nature on both cellular and systemic level? If they are, is there a definite dependence of organisms' adverse responses to NP exposure on their dimensions within the conventional nanometer range and/or on their chemical nature?

The paradigm of a considerably higher biological aggressiveness of NPs as compared with particles of the same substance in the micrometric range first emerged as a theoretically sound perception.^{87,88,101} Very soon some experimental facts corroborating this perception were obtained by different researchers, but this corroboration was during a period neither unanimous nor absolutely reliable due to some drawbacks and lack of comparability between different experimental designs. Thus, even as recently as 5 years ago very reputable authors would justifiably state in the oldest specialized nanotoxicological journal that "this common perception of greater NP toxicity is based on a limited number of studies",⁹⁵ and still less was known about the comparative toxicity of chemically identical particles having different sizes within the conventional nanometric range.

However, quickly the situation has changed substantially. Today's body of facts allows one to say without qualification that, given identical exposure pathways and similar chemistry, the toxic effects of metal and metal oxide NPs are much higher than those of their micrometer counterparts, even of minimal (including submicron) sizes of the latter, while for a given size the toxicity of NPs depends on their chemical nature and related properties, including solubility. Along with the latter property, quite often causing "the Trojan horse effect" (intracellular release of toxic metal ions by engulfed

Table 2 Number of cells in the bronchoalveolar lavage fluid (BALF) 24 hours after the intratracheal instillation of suspension of gold (50 nm) or silver (49 nm) particles to rats at a dose of 0.2 mg per rat (\times ±se)

Substance administered	Number of cells $\times 10^6$			NL/AM
	Total	Neutrophil leukocytes (NL)	Alveolar macrophages (AM)	
Nanosilver	4.25±0.77*	2.99±0.71*	1.16±0.14	2.47±0.33*
Nanogold	2.30±0.93*	0.63±0.15**	0.94±0.09	0.63±0.13**
Water (controls)	1.41±0.33	0.13±0.04	0.89±0.18	0.14±0.023

Notes: Statistically significant difference *from control group; **from nanosilver group ($P < 0.05$ by Student's *t*-test).

NPs), another now widely recognized major mechanism of NP cytotoxicity and, especially, genotoxicity is the generation of ROS.¹⁰²

The toxic impact of particles on lung phagocytes is of a special significance not only as an important unfavorable factor influencing pulmonary clearance but also as an in vivo assessment of comparative particle cytotoxicity in a broader sense of the term. As it was mentioned above, recruitment of new echelons of NLs (beginning from enhanced differentiation of stem cells) prevailing over that of macrophages is a mechanism of compensation, even if partial, for the breakdown of AMs damaged by cytotoxic particles. It was found that this recruitment is controlled by the mass of macrophage breakdown products (MBP) and especially by their lipid fraction.^{96–100} Therefore, the more cytotoxic for AMs are particles deposited in the lungs (or the higher the dose of MBPs obtained by aseptic freezing-thawing or ultrasonic destruction of non-activated peritoneal macrophages and then instilled intratracheally), the higher is a count ratio of NLs to AMs in the BALF. Therefore this ratio (NL/AM) can be used as an indirect but highly informative

comparative in vivo index for the cytotoxic action of any low-soluble particle, and it was demonstrated that ranking of dusts by this index was well correlated with the ranking of their cytotoxicity based on the Trypan blue exclusion test for cell viability in vitro. All these earlier revealed facts permitted us to use the BALF NL/AM ratio as an in vivo estimate of comparative cytotoxicity for NPs as well.

Using this index, we showed that the investigated NPs were far more cytotoxic as compared with even the smallest MPs of the same metal (as examples, see Table 1 for silver NPs versus MPs as well as the values given in the text above for copper oxide NPs versus submicron particles). Moreover, within the nanoscale range, the smaller a particle, the more cytotoxic it was (see Figure 3 for iron oxide particles with different dimensions). Given a virtually equal nanoscale size, the cytotoxicity of particles can be quite different depending on their chemical nature. Thus we found, judging by NL/AM ratio, that nano-Ag was far more cytotoxic as compared with nano-Au (see the NL/AM column in Table 2) and that nano-NiO was more cytotoxic as compared with nano-Mn₃O₄.¹⁰

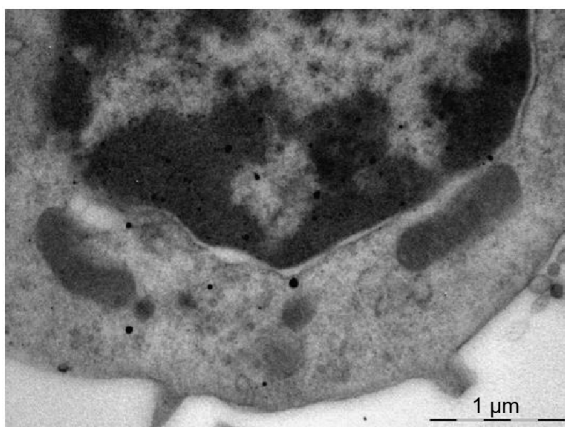


Figure 9 Gold nanoparticles uniformly distributed throughout the cytoplasm and nucleus of an alveolar macrophage.

Notes: The two-contour organization of the nucleus membrane is intact throughout. There is a mitochondrion visible which is not interacting with nanoparticles but, nevertheless, is intact only partly. Transmission electron microscopy, magnification $\times 22,000$.

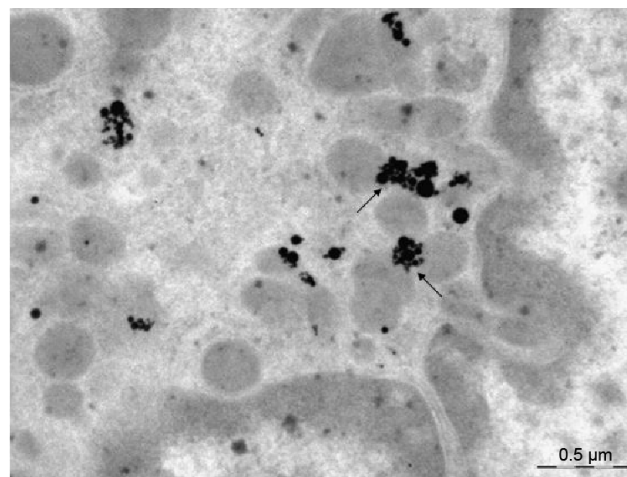


Figure 10 An alveolar macrophage.

Notes: Penetration of silver nanoparticles from aggregates in the cytoplasm into mitochondria (arrows). No silver nanoparticles are discovered in the nucleus. Transmission electron microscopy, magnification $\times 28,000$.

There are differences also in ultrastructural damage to the macrophage caused by different nanometals. For instance, nanosilver was found more prone to damage mitochondrial membranes and cristae than nanogold.⁷

Different solubility of different nanometals is one of probable explanations of their different cytotoxicity (in particular, silver NPs were proved more soluble as compared with equidimensional gold ones) but hardly the only one. Based on a review of extensive literature, Fröhlich¹⁰² believes that, commonly for many metals and their oxides, the release of free ions is a second important mechanism of NP cytotoxicity (the first one being the generation of ROS as a result of various chemical reactions).

When comparing the BALF cell absolute counts and quantity of “pits” on the surface of these cells revealed by the sc-AFM with the NL/AM ratio values, we can see that the more cytotoxic the NPs are (because of either their smaller dimensions or their chemical nature, or both) the more active the pulmonary phagocytosis response is and the more avidly these NPs are engulfed by phagocytes. This phenomenon can be easily explained, as MBPs stimulate not only the recruitment but also the phagocytic activity of viable macrophages judging by the results of an in vitro test with 1 μm polystyrene beads.¹⁰³ It stands to reason, however, that such compensation for AMs’ breakdown cannot be complete and, in the final count, the especially high cytotoxicity of NPs can be regarded as their very dangerous characteristic increasing along with decrease in their diameter.

The relationship between particle dimensions and their toxicity is not so unequivocal on the organs and organism levels. For instance, judging by our results,^{1,3} the subchronic

toxicity of magnetite NPs was higher as compared with that of 1 μm particles, but within the nano-scale range the relationship under consideration proved inverse for some effects. Such seemingly paradoxical dependence of adverse effects of NPs on their diameter is quite characteristic of target organs rich in reticuloendothelial system and thus capable of actively accumulating NPs from blood – such as liver and spleen.

This fact should be attributed to differences in NP toxicokinetics. The latter is controlled, first of all, by their more or less easy penetration through biological barriers into the bloodstream from sites of their primary deposition to be then captured by reticuloendothelial system cells of this or that organ. It may be assumed that this mechanism of particle translocation has to be more effective for smaller NPs. However, the smaller the particle, the quicker it dissolves from these secondary depots due to its immense specific surface. Besides, the smallest NPs are presumably more cytotoxic for any cells, resident macrophages included, and so cause their destruction (with eventual liberation of NPs back to bloodstream) more effectively. The balance between these oppositely acting mechanisms of NP retention in different tissues depends on many variables. The result can be a greater load of an organ with larger NPs than with smaller ones (but, at the same time, much greater for both as compared with microparticles) as illustrated for magnetite NPs and MPs in Figure 11. It is no wonder that some morphometric indices for the toxic damage to the liver were more prominent in rats exposed to 50 nm particles than to 10 nm particles of magnetite, while NPs of both sizes proved more hepatotoxic as compared with microparticles (Table 3).

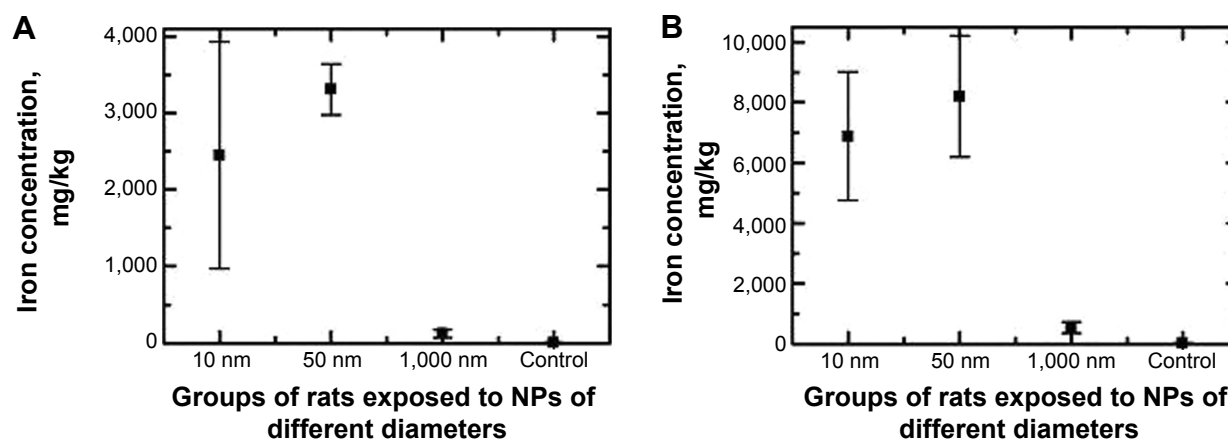


Figure 11 Mean value (\pm standard deviation) of iron oxide (magnetite) concentration in the rat tissues of (A) liver and (B) spleen after repeated intraperitoneal injections of magnetite particles of different size.

Note: Electron paramagnetic resonance method.

Abbreviation: NPs, nanoparticles.

Table 3 Some morphometric indices for the state of rat's liver after subchronic exposure to iron oxide (magnetite) particles ($\bar{x}\pm se$)

Count per 100 liver cells	Groups of rats			
	Control	Exposed to magnetite particles		
		10 nm	50 nm	1 μ m
Akaryotic hepatocytes	13.1 \pm 0.9	37.0 \pm 1.6**	44.4 \pm 1.3**	19.6 \pm 1.7*
Binucleated hepatocytes	5.1 \pm 0.5	2.3 \pm 0.3**	2.0 \pm 0.3**	6.1 \pm 0.6
Kupffer cells	27.2 \pm 1.6	31.8 \pm 2.0*	19.2 \pm 3.3*	33.2 \pm 0.4*

Notes: *A statistically significant difference from the control group; *from the 50 nm group; *from the 1 μ m group ($P<0.05$ by Student's *t*-test and by Mann-Whitney test).

Above any doubt, the systemic toxicity of equidimensional NPs depends on the toxic properties of the metal as such and can differ very much between different metals. For instance, nanosilver is more toxic than nanogold when tested in parallel, and specifically, nanosilver is more genotoxic in vivo compared with nanogold judging by the results of the random amplified polymorphic DNA test performed on cells of different tissues of rats under subchronic intoxication (Table 4).⁷ The fact that these differences in the genotoxicity of nanometals compared are less pronounced in the kidneys, can, most probably, be explained by the fact that silver NPs, which, judging by their considerably higher solubility in a model biomedium in vitro as compared with gold NPs,⁷ release substantially higher concentrations of ions into the blood as well in vivo, and, therefore, possess higher nephrotoxicity. We believe that marked toxic damage of kidney epithelial cells can mask the metal's genotoxic effect. This hypothesis found confirmation in an experiment with copper oxide NPs,⁸ which are even more soluble in vivo and thus even more nephrotoxic compared with silver NPs. In that experiment, the kidneys proved to be the only organ in which the DNA fragmentation coefficient was not elevated in response to subchronic intoxication by copper oxide NPs in comparison with controls. Meanwhile, in a parallel group, in which the animals were exposed to the same CuO NPs but against the background of bioprotectors (see "Is it possible to protect

the health of people exposed occupationally to metallic NPs in the workroom air by enhancing organism's resistance to their adverse action?" section), the morphometric data revealed a considerably lower death rate of tubular epithelial cells, while it was just against this background that a significant genotoxic effect of nano-CuO became quite manifested.

Is it possible to protect the health of people exposed occupationally to metallic NPs in the workroom air by establishing very low but still practicable PELs?

Although the problem of nanomaterials' health risk assessment and management is discussed widely and has been comprehensively reviewed by many authors,^{104,105} a general methodology of safe exposure standard setting in this field is still lacking, and there are but few examples of establishing or recommending reference exposure levels for specific engineered NPs in workroom air. Thus in most cases, in the field of nano-safety both risk assessments and risk management scenarios still have to be implemented with the qualification that exposure limits for individual nano-materials are not available.¹⁰⁶ Given this serious uncertainty, the precautionary approach is that of choice in the worker's health protection in nanotechnology.¹⁰⁷

Irrespective of whether toxicologists of this or that country believe in a no-threshold character of the dose-effect relationship for any or, at least, for genotoxic and carcinogenic harmful substances or stick to the paradigm of the existence of thresholds for any harmful effects, there are little doubts as to the practical need for, and the possibility of establishing presumably safe exposure levels. In a growing number of countries such levels are set as law enforceable standards, in particular, for the workplace air quality. Whatever such standards are called (maximal allowable concentrations, PELs, reference exposure levels, workplace exposure limit, occupational exposure limits, etc), establishing them is justified by assuming, explicitly or by default, that the evolutionarily acquired protective mechanisms and

Table 4 Coefficients of the genomic DNA fragmentation in rats exposed to subchronic administration of silver or gold nanoparticles (based on the results of RAPD-test), $\bar{x}\pm se$

Group of rats exposed to	Tissues				
	Liver	Bone marrow	Spleen	Kidney	Nucleated cells of peripheral blood
Water (controls)	0.399 \pm 0.001	0.385 \pm 0.003	0.379 \pm 0.002	0.385 \pm 0.003	0.383 \pm 0.001
Nanogold	0.392 \pm 0.010°	0.412 \pm 0.014*	0.397 \pm 0.008°	0.422 \pm 0.009*	0.403 \pm 0.018
Nanosilver	0.461 \pm 0.002*	0.455 \pm 0.032*	0.462 \pm 0.001*	0.423 \pm 0.008*	0.413 \pm 0.012*

Notes: Statistically significant difference *from the control group; °between the group receiving nanogold and the group receiving nanosilver ($P<0.05$ by Student's *t*-test).

Abbreviation: RAPD, random amplified polymorphic DNA.

the compensation abilities of the organism enable humans to exist under some low level of harmful exposure without any appreciable impairment of health and any perceptible rise over the background risks of stochastic harmful effects.

The experimental results as discussed above permitted us to maintain that the concept of quasi-defenselessness of the organism against NPs had to be critically re-evaluated. On the other hand, however, they provided evidence supporting the prevailing presumption that particulates, even ones that are relatively innocuous in the micrometer range (like iron oxides), may be markedly toxic in the nano-state while within the conventional nanometer range the dependence of toxicity on particle size proved intricate and non-unique. Therefore we believe that: a) establishing permissible or reference values for controlling and assessing health risks due to NPs exposures is possible in principle; b) such levels should be significantly lower as compared with actual exposure limits established (or recommended) for chemically identical MPs; and c) there are no reliable foundations as yet for differentiating such standards for particles of different nano-size.

As to a well-known example of such approach, we may refer to the recommended exposure limit (REL) 0.3 mg/m³ proposed by the US National Institute for Occupational Safety and Health (NIOSH) for the ultrafine (including engineered nanoscale) TiO₂ particles that is eight times lower than REL 2.4 mg/m³ for the fine TiO₂.¹⁰⁸ It is true that the NIOSH maintains that “similarly for other particulate materials, if the toxicity varies by particle size (at the same mass dose), distinct exposure limits for distinct particle size fractions might be a reasonable approach for protecting exposed workers”. Nevertheless, one and the same REL is proposed for all the TiO₂ nano-scale fractions while just the TiO₂ commercial nano-powders are being manufactured and marketed for a variety of uses in a rather wide range of primary particle diameters (for instance, ~10 nm, 10–30, or ≤50 nm). Thus we may say that the NIOSH REL proposal is in good accordance with, though established independently of, the above postulates (b and c) of ours.

A similar but even more conservative approach to establishing exposure standards for engineered nanomaterials was proposed in 2010 by the Australian Government agency “Safe Work Australia”: for nanocrystals, quantum dots, ceramic oxides, and metals the suggested benchmark exposure level should be 0.066 WEL (workplace exposure limit for bulk material), that is by diminishing the latter by a factor of 15.

In accordance with this regulatory practice, we also believe that for workroom air concentrations of metallic NPs, PELs at this historical stage can and should be based, even if

tentatively, on a sufficiently conservative rule of decreasing approximately tenfold the exposure limits previously established for respective micro-scale industrial aerosols. Acting along this line, we propose tentative PELs for NPs of iron oxide (0.4 mg/m³), copper oxide (0.05 mg/m³), silver (0.1 mg/m³), and gold (0.2 mg/m³). As an example, we will discuss here in more detail the case of Fe₃O₄ NPs as presented by us.⁴

To begin with, workroom exposure standards are established in different countries for only one of the iron oxides – namely Fe₂O₃. It may be assumed, however, that there is no significant difference in toxicity between different iron oxides. For example, the Joint Expert Committee on Food Additives gives one and the same acceptable daily intake value for Fe₃O₄ (the so-called black iron [II, III] oxide) and for anhydrous or hydrated Fe₂O₃ (respectively, red and yellow iron [III] oxides). So, as a starting point for calculating a tentative PEL for nano Fe₃O₄ one might assume by inference the actual standards for Fe₂O₃ dust and/or fumes which are:

- USA Occupational Safety and Health Administration PEL – time-weighted average (TWA) 10 mg/m³ (as Fe);
- USA NIOSH REL – TWA 5 mg/m³ (as Fe);
- Canadian Occupational Exposure Limit – TWA 5 mg/m³ for respirable dust (in Alberta) or 10 mg/m³ for total dust (in British Columbia);
- Russian Maximum Allowable Concentration (MAC) – work shift average 6 mg/m³.

Note that the Russian MAC established for whole Fe₂O₃ substance is equivalent to ~2 mg/m³ if re-calculated for Fe, and thus it is actually the most conservative of all these standards. Taking into consideration the precautionary principle, we proposed to start just from this MAC and, moreover, to diminish it not eight but 15 times (in accordance with the above-mentioned Australian suggestion as the most conservative, too). Thus we obtained the value 6×0.066=0.4 mg/m³ (as TWA). This value is close enough to the above-mentioned NIOSH REL for ultrafine TiO₂ (0.3 mg/m³).

There is no doubt that a more sophisticated study of the same nanomaterials’ adverse effects (their possible

Table 5 Succinate dehydrogenase activity (number of formazan granules per 50 blood lymphocytes) in rats exposed to subchronic administration of nanosilver particles with and without protection by a bio-protective complex (BPC) (x±se)

Groups of rats given			
Nanosilver (NS)	NS + BPC	BPC	Controls
679.9±12.4* [*]	827.8±22.1	834.1±11.2	805.33±12.6

Notes: Statistically significant difference *from the controls; ^{*}from the NS + BPC group (P<0.05 by Student’s t-test).

Table 6 Morphometric indices for tubular epithelium damage in kidneys of rats after repeat intraperitoneal injections of copper oxide nano-suspensions and/or oral administration of the bio-protective complex (BPC) ($\bar{x}\pm\text{se}$)

Groups of rats given	Brush border loss (% lengthwise)	Epithelial desquamation (% lengthwise)
Water (control)	5.39±0.42	0.33±0.13
Nano copper oxide	8.36±0.76*	1.16±0.38*
Nano copper oxide + BPC	5.98±0.46*	0.98±0.35
BPC	6.03±0.57	0.73±0.21

Notes: Statistically significant difference *from the control group; †from the group given nano copper oxide (without the BPC); $P<0.05$ ($P<0.05$ by Student's *t*-test).

carcinogenicity included) and of dose-response relationships, as well as further discussion of a general methodology for substantiating reference exposure levels for nanomaterials will be necessary to make such safety standards more reliable. Nevertheless, we believe that, for the time being, it is wiser to use the standards obtained with the above-discussed approach (tentative but still not spun out of thin air and obligatorily based on results of actual comparative toxicological experiments) than to be passively waiting for better ones.

Is it possible to protect the health of people exposed occupationally to metallic NPs in the workroom air by enhancing organisms' resistance to their adverse action?

However low the levels of exposure to NPs were established as permissible, the especially high toxicity of this class of occupational hazards makes it worthwhile to search for possibilities to render exposed people less sensitive to their effects with the help of a combination of specific and unspecific agents which are innocuous themselves in preventively effective doses. The general concept of such "biological prophylaxis" of different occupational and environmental

intoxications, its theoretical premises and general principles along with numerous examples of their realization have been published by us over several decades repeatedly, including, in review articles.^{109,110} Now we began to use this experience for trying to protect the organism against the impacts of nanometals.

In particular, our experiments have demonstrated that both systemic toxicity and in vivo genotoxicity of silver⁷ and of copper oxide⁸ NPs were markedly attenuated against the background of oral administration of multi-component BPC. These BPCs comprised pectin, multivitamin–multimineral preparations, some amino acids, and omega-3 polyunsaturated fatty acid (for the detailed BPCs' composition and doses see the above references).^{7,8} In both cases the protective action was demonstrated in respect to many adverse effects of sub-chronic toxicity. Some illustrations are given in Tables 5–7, and in Figures 12 and 13.

We found as well¹⁰ that in rats that were being given glutamate, glycine, acetyl cysteine, iodide, and a Se-containing multivitamin preparation orally during 4 weeks before a single IT instillation of NiO+Mn₃O₄ NPs (0.25 mg each), the latter evoked a significantly lower NL recruitment with lower NL/AM ratio than in rats so exposed without any pretreatment.

Conclusion

The investigated metallic NPs are much more noxious on both cellular and systemic levels as compared with their fine micrometric or even submicron counterparts. However, within the conventional nanometer range the dependence of systemic toxicity on particle size is intricate and non-unique due to complex and often contra-directional relationships between the intrinsic biological aggressiveness of specific NPs (in particular, their cytotoxicity), on the one hand, and complex mechanisms that control their biokinetics, on the other.

Our data testify to a high activity of at least one of the key physiological defense mechanisms (pulmonary

Table 7 Coefficients of the genomic DNA fragmentation in rats exposed to subchronic administration of nanosilver particles with and without protection by a bio-protective complex (BPC) (based on the results of RAPD-test), $\bar{x}\pm\text{se}$

Group of rats exposed to	Tissues				
	Liver	Bone marrow	Spleen	Kidney	Nucleated cells of peripheral blood
Water (controls)	0.399±0.001	0.385±0.003	0.379±0.002	0.385±0.003	0.383±0.001
Nanosilver (NS)	0.461±0.002*	0.455±0.032*	0.462±0.001*	0.423±0.008*	0.413±0.012*
NS + BPC	0.408±0.011†	0.373±0.003*†	0.419±0.003*†	0.407±0.006*†	0.390±0.007

Notes: Statistically significant difference *from the control group; †between the group given NS together with BPC and the group given NS only ($P<0.05$ by Student's *t*-test).

Abbreviation: RAPD, random amplified polymorphic DNA.

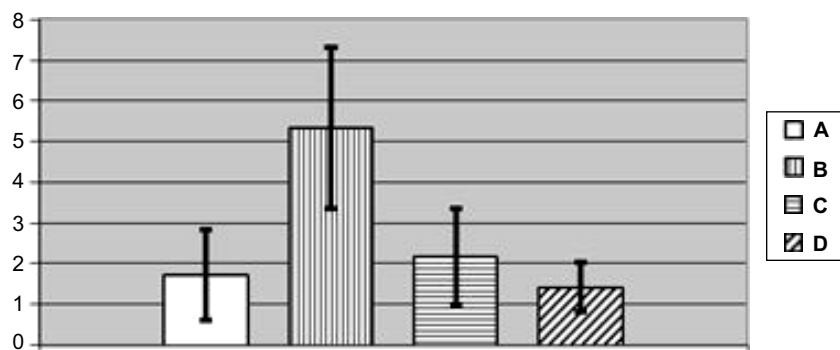


Figure 12 Number of brown pigment micro aggregates per square of Avtandilov's grid in the spleen's red pulp of rats exposed (A) to water (control); (B) to water suspension of copper oxide nanoparticles; (C) to the same against the background of BPC administrations; and (D) to the BPC only.

Notes: Average values with 95% confidence intervals. Differences are statistically significant between (B) or (C) versus (A), and between (C) versus (B) ($P < 0.05$ by Student's *t*-test).

Abbreviation: BPC, bio-protective complexes.

phagocytosis of deposited particles) against NPs deposited in airways. This fact suggests that safe low levels of exposure to airborne NPs are possible in principle. For the time being, however, there seems to be no reliable foundations for establishing different PELs for particles of different sizes within the established nanometric range. For workroom air concentrations of metallic NPs, PELs can be proposed at this stage, even if tentatively, based on a sufficiently conservative approach of decreasing by approximately an order of magnitude the exposure limits established for respective micro-scale industrial aerosols. Acting along this line, we propose tentative PELs for NPs of iron oxide (0.4 mg/m^3), copper oxide (0.05 mg/m^3), silver (0.1 mg/m^3), and gold (0.2 mg/m^3). However, to substantiate more reliable standards of presumably safe occupational exposures to metallic NPs, a wide international program of long-term inhalation experiments with such low-level concentrations and with a

broad spectrum of toxicity, genotoxicity, and carcinogenicity indices has to be designed and implemented.

It has been shown that against the background of adequately composed combinations of some bioactive agents used in innocuous doses the toxicity and even genotoxicity of metallic NPs could be markedly attenuated. Therefore we believe that, along with decreasing exposures to NPs, enhancing organisms' resistance to their adverse action with the help of such bioprotectors can prove an efficient auxiliary tool of health risk management in related occupations and strongly recommend to further develop this vector of nanotoxicological research.

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Disclosure

The authors declare no conflict of interest.

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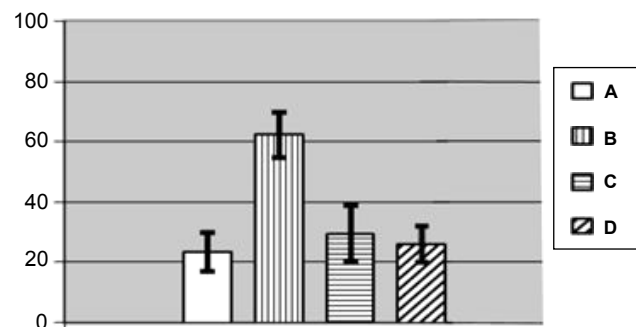


Figure 13 Number of cells without a nucleolus per 100 Golgi cells in nucleus caudatus of rats exposed (A) to water (control); (B) to water suspension of copper oxide nanoparticles; (C) to the same against the background of BPC administrations; (D) to the BPC only (average values with 95% confidence interval).

Notes: Differences are statistically significant between (B) and (A); (C) and (B) ($P < 0.05$ by Student's *t*-test).

Abbreviation: BPC, bio-protective complexes.

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